

**Magma storage and plumbing of adakite-type post-ophiolite intrusions**

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**Magma storage and plumbing of adakite-type post-ophiolite intrusions in the Sabzevar ophiolitic zone, NE Iran**

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Received: 24 July 2014 – Accepted: 24 July 2014 – Published: 8 August 2014

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Published by Copernicus Publications on behalf of the European Geosciences Union.

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## Abstract

Subduction-related adakite-type intrusive rocks emplaced into the late Cretaceous-Paleocene Sabzevar ophiolite zone, NE Iran, range from Mg-andesite to rhyodacite in composition. Here we investigate the magma supply system to these subvolcanic intrusive rocks by applying thermobarometric mineral and mineral-melt equilibrium models, including amphibole thermobarometry, plagioclase-melt thermobarometry and clinopyroxene-melt barometry. Based on the results of these thermobarometric models, plagioclase crystallized dominantly at pressures of  $\sim 350$  (468–130) MPa, while amphiboles record both low pressures ( $\sim 300$  MPa) and very high pressures ( $> 700$  MPa) of crystallization. The latter is supported by the calculated pressures for clinopyroxene crystallization (550 to 730 MPa). The association of amphibole with clinopyroxene and no plagioclase in the most primitive samples (Mg-andesites) is consistent with amphibole fractionation from very hydrous magmas at deep crustal levels of the plumbing system, which may have been a key process to intensify adakite-type affinities in this rock suite. Barometry, combined with frequent disequilibrium features, such as oscillatory-zoned and sieve-textured plagioclase crystals with An-rich overgrowths in more evolved samples, imply final magma differentiation occurred in an open upper crustal magma system that developed progressively stronger compositional modifications during high-level magma storage.

## 1 Introduction

The Sabzevar ophiolitic zone (SOZ) in northeast Iran is part of the eastern Tethyan ophiolite belt and represents a remnant of the Cretaceous Tethyan ocean lithosphere which was obducted in late Cretaceous to early Paleocene time (Shojaat et al., 2003). The ophiolite belt contains ultramafic rocks (harzburgite, dunite and lherzolite), small masses of gabbro, and a thick sequence of submarine basaltic lavas (Shojaat et al., 2003; Khalatbari Jafari et al., 2013). The ophiolite is intruded by widespread subvol-

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canic stocks and dykes of intermediate to felsic compositions and the main purpose of this study is to investigate the compositional spectrum and ascent history of these post-ophiolite subvolcanic rocks.

The detailed evolution and ascent history of the Sabzevar post-ophiolite rocks has not been investigated previously and we present new major and trace element data on whole rocks plus major element data from the main minerals to improve our knowledge on the history of the post-ophiolite intrusive suite. We attempt to compare post-ophiolite felsic rocks in the southern Sabzevar belt with dominantly intermediate ones in the North and aim to test if differences between the post-ophiolite subvolcanic rocks of the two sectors exist. Finally, we hope to establish the  $P$ - $T$  conditions of amphibole and plagioclase crystallization through mineral-melt equilibrium thermobarometry to help reconstruct the former plumbing system to these rocks.

Amphibole occurs as essential rock forming mineral in a wide variety of igneous and metamorphic rocks and is especially abundant in calc-alkaline igneous rocks. Experimental studies have synthesized amphibole over a pressure range of up to 2300 MPa and 400 °C to 1150 °C in calc-alkaline compositions (Blundy and Holland, 1990) and amphibole has therefore considerable potential as an indicator of crystallization conditions, both as geothermometer and geobarometer (Blundy and Holland, 1990; Ridolfi, 2010).

The composition and growth morphology of igneous feldspar usually reflects progressive changes in crystallization conditions that give a reliable record of crystallization dynamics of an associated melt and its thermal and compositional history (e.g. Troll and Schmincke, 2002; Slaby and Götze, 2004). In the studied rocks, amphibole and plagioclase are found most widely in the northern suite of intermediate rocks and we employed the thermobarometers of Ridolfi and Renzulli (2012) for selected amphiboles to determine crystallization and storage conditions for these rocks. Plagioclase-liquid thermobarometry (Putirka, 2008) and clinopyroxene-melt barometry (Putirka, 2008) were employed in order to complement and test the results from amphibole thermobarometry.

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## 2 Geological setting

The Sabzevar ophiolite zone is ~ 200 km long and ~ 10 km wide and represents a tectonically dismembered ophiolite complex, located along the northern boundary of the Central Iranian microcontinent (Shojaat et al., 2003). Plate tectonic reconstructions suggest that the Sabzevar ophiolite was part of the Tethyan Ocean that formed during Cretaceous rifting from a narrow but deep seaway, separating the Central Iranian microcontinent from the Eurasian plate. The ophiolite was emplaced during an episode of northeast dipping subduction (i.e. closure) of the Tethyan seaway (the Sabzevar Ocean) in Upper Cretaceous–early Paleocene time (Shojaat et al., 2003). The emplacement of the Sabzevar ophiolitic belt was followed by lasting post-ophiolite volcanism from at least the Eocene (~ 40 Ma) to the end of the Pliocene (~ 2.3 Ma) (Lensch et al., 1977; Spies et al., 1983; Shojaat et al., 2003). This Eocene volcanism is expressed in central and in northern Iran (Berberian and King, 1981; Berberian et al., 1982; Bina et al., 1986; Stampfli and Borel, 2002; Agard et al., 2005; Shabanian et al., 2012) and Eocene volcano-sedimentary rocks cover extensive areas in the Sabzevar region. Since the Oligocene, volcanism has been intermittent (Jahangiri, 2007) and, for ~ 10 Ma focussed predominantly on the Turkish-Iranian plateau (e.g., Berberian and Berberian, 1981; Keskin et al., 1998; Azizi and Moinevaziri, 2009). The Oligo-Miocene sedimentary rocks are composed of marine flysch-type facies in their lower parts, while younger facies are of continental origin and contain, in some parts, extremely high percentages of volcanic detritus. These sedimentary rocks are tectonically imbricated, often steeply dipping and penetrated by Oligocene alkali-olivine-basalt magmatism (Ghasemi et al., 2011). Following the Eocene volcanic and volcano-sedimentary and the Oligocene volcanic activity, magmatism recommenced in the region and continued to the end of the Pliocene. In the Sabzevar belt, characteristically, post-ophiolite subvolcanic rocks occur in the north and in the south of the ophiolite zone (Fig. 1) and magmatism comprised intermediate rocks in the northern part, while felsic rocks dominate the southern part (Fig. 2). Fragments of host rocks, including serpentinized harzburgite

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gites, tuffaceous siltstone and marl are found in some intrusions as xenoliths (Fig. 3). Large volumes of detritus originated from these intrusive domes, forming Pliocene-Pleistocene conglomerates (Fig. 3d), which indicate that the studied domes eventually outcropped at the Earth's surface around the Miocene-Pliocene boundary (Salehinejad, 2008).

### 3 Method

#### 3.1 Analytical techniques

Twenty-three fresh whole rock samples from the northern sector and 24 samples from the southern sector were selected for major and trace element analysis at AMCE analytical laboratory, Vancouver, Canada. Total abundances of the major oxides were determined by ICP-emission spectrometry following a Lithium metaborate/tetraborate fusion and dilute nitric digestion. Loss on ignition (LOI) is by weight difference after ignition at 1000 °C. Trace elements and REE concentrations were obtained by solution inductively coupled plasma-mass spectrometry (ICP-MS) for whole rocks. For further analytical details and uncertainties please see "www.acmelab.com". Major, trace element data for post-ophiolite subvolcanic rocks are listed in Table 1.

Major element analyses on selected minerals (amphibole, plagioclase and pyroxene) were performed on Jeol JXA 8530F Hyperprobe at CEMPEG. Analytical conditions included an accelerating voltage of 15 kV, a beam current of 10 nA and counting times of 10 s on peaks and 5 s on  $\pm$  background. For calibration of all elements a set of mineral and synthetic standards has been used. All elements were analysed using K (alpha) lines. The analysed data set consists of 197 points on amphibole, 212 on plagioclase and 29 on clinopyroxene, collected from 30 amphibole, 26 plagioclase and five complementary clinopyroxene crystals. Representative compositions of selected minerals are given in Tables 2 and 3.

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## 3.2 Amphibole thermobarometry

Temperature, pressure, oxygen fugacity and  $H_2O_{\text{melt}}$  conditions were estimated from amphibole compositions using the recent thermobarometric formulation of Ridolfi and Renzulli (2012), which is able to estimate the  $P$ ,  $T$  and  $fO_2$  in a wide range of amphibole crystallization conditions. Their new single-crystal model (requiring only the amphibole compositions) allows to estimate the physico-chemical parameters at low uncertainties ( $T \pm 23.5^\circ\text{C}$ ,  $P \pm 11.5\%$ ,  $H_2O_{\text{melt}} \pm 0.78\text{ wt.}\%$ ) for calc-alkaline and alkaline magmas up to  $1130^\circ\text{C}$  and  $2200\text{ MPa}$  and  $\Delta\text{NNO}$  values ( $\pm 0.37$  log units) up to  $500\text{ MPa}$  (NNO = nickel-nickel oxide buffers). Application of this method is limited to igneous amphibole phenocrysts and cannot be applied to fluid-related (hydrothermal) amphibole veins, to microlites or quenched amphibole zones in erupted products that likely reflect variable disequilibrium conditions (Ridolfi and Renzulli, 2012).

## 3.3 Plagioclase-melt thermobarometry

To test and complement the results of the amphibole thermobarometry, the plagioclase-melt thermobarometer of Putirka (2008), calibrated for hydrous systems, was put to use. The standard error of estimation (SEE) for this thermobarometer is  $\pm 36^\circ\text{C}$  and  $\pm 247\text{ MPa}$  (Putirka, 2008), which is less precise than the amphibole barometer employed. The most commonly used nominal melts are whole rock, groundmass and glass compositions (Putirka et al., 2003; Putirka, 2008) and the respective whole rocks were applied as the nominal melt in our study.

## 3.4 Clinopyroxene-melt barometry

This barometry model for clinopyroxene-melt equilibria after Putirka (2008, Eq. 32c) has been applied to cross-check the result from amphibole and plagioclase-melt barometry. This model is based on Al partitioning between clinopyroxene and coexisting melt in hydrous magmatic systems. The SEE for this barometer is  $\pm 150\text{ MPa}$  and

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the respective input melt was the corresponding whole rock composition (cf. Putirka, 2008).

### 3.5 Equilibrium tests

Putirka (2008) present an equilibrium test for plagioclase-melt thermobarometry based on the partitioning coefficients of the anorthite and albite components,  $K_d[\text{An-Ab}]$ . The equilibrium constant is sensitive to temperature and should be  $0.10 \pm 0.05$  at  $T < 1050^\circ\text{C}$  and  $0.27 \pm 0.11$  at  $T > 1050^\circ\text{C}$  (Putirka, 2008). The plagioclase components which are near to equilibrium with a given melt at the given pressure and temperature conditions should fall within the  $K_d[\text{An-Ab}]$  envelope appropriate for this melt and mineral composition.

Clinopyroxene-melt equilibrium was tested using two models. Firstly, we looked at the partition coefficient of Fe and Mg between clinopyroxene and melt. The acceptable range of equilibrium values for  $K_d(\text{Fe-Mg})$  is  $0.27 \pm 0.08$ . All datapoints plotting within this range were further tested using the equilibrium test presented in Putirka (1999), based on the equilibrium partitioning of Na-Ca-Al. There, predicted values of different components that crystallize from a nominal melt are compared with observed clinopyroxene components and mineral-melt equilibrium pairs require a close match between predicted and observed components to satisfy equilibrium conditions.

## 4 Results

### 4.1 Petrography

Post-ophiolite magmatism in the northern part of Sabzevar belt comprises intermediate rocks of andesite, trachyandesite, trachydacite to dacite composition. Andesite/trachyandesite rocks intrude into the ophiolite complex and younger volcanoclastic rocks and appear as grey-green dome-forming intrusions in the field, but occur as dykes also. In thin section, the andesite/trachyandesite rocks show porphyritic,

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acicular and fluidal textures and contain plagioclase and amphibole as the main minerals together with variable amounts of plagioclase, biotite and Fe-Ti oxide microlites in the groundmass (Fig. 4a and b). Clinopyroxene phenocrysts of mostly diopside to augite composition are only found in a high Mg-andesite samples (Fig. 4c). Notably, this rock is a mafic basaltic andesite and shows amphibole and clinopyroxene phenocrysts, but lacks plagioclase as phenocryst phase. Amphibole phenocrysts are generally star shaped or long prisms up to 1 cm and the groundmass of the dyke is composed of fine-grained hornblende, plagioclase and augite.

Euhedral to subhedral biotite is present and in more evolved trachydacites and is moderately enriched in Mg (average Mg#  $\approx$  0.65). Monomineralic glomerocrysts of amphibole do occur occasionally in the trachyandesite samples. Accessory minerals include apatite, Fe-Ti oxides and rare alkali feldspar. Apatite is ubiquitous, occurring as euhedral needle-like phenocrystals and as inclusions in larger phenocrysts. Fe-Ti oxides (mostly magnetite) occur as euhedral microphenocrysts and as inclusions in amphibole and biotite. Some trachyandesite samples also contains small but coarse-grained and rounded crystal clots up to 2 mm in diameter (Fig. 4d) that are made up of euhedral microcrystals of plagioclase, amphibole and accessory apatite.

Dacite samples from light-grey domes and dykes show plagioclase as the volumetrically dominant phenocryst phase, which usually displays polysynthetic twinning and compositional zoning (Fig. 4e). Amphibole and biotite phenocrysts are apparent and the medium to fine-grained groundmass is usually composed of plagioclase and amphibole microcrystals with biotite, Fe-Ti oxides and apatite occurring in smaller proportions.

Felsic rock types of broadly rhyolitic compositions crop out as dominantly white-coloured domes in the southern and southeastern part of the Sabzevar belt. Quartz, sanidine, and plagioclase (typically albite in composition, An<sub>5</sub>-An<sub>15</sub>) form phenocrysts in a fine-grained groundmass (Fig. 4f). Although smaller quartz phenocrysts have frequently straight edges, many large crystals display embayments and round edges. Some embayments are short and close to the margin, whereas others are penetrating





typical arc-related calc-alkaline composition (cf. Defant and Drummond, 1990). However, a few samples plot near the intersection of the adakite and the arc-related fields, while a plot of Sr against CaO+Na<sub>2</sub>O (after Castillo, 2011), shows most of our samples in the high silica adakite field (Fig. 8b).

Notably, the samples of the southern part have higher silica contents than the northern suite and range from 68.9 to 73.6 wt.% in SiO<sub>2</sub>, falling into the rhyolite field of the TAS diagram (Fig. 5a). They have high alkali contents with K<sub>2</sub>O contents of 2 to 4.2 wt.% and Na<sub>2</sub>O of 4.6 to 7.1 wt.% resulting in Na<sub>2</sub>O/K<sub>2</sub>O ratios of 1.3 to 3 and the rocks classify as high-K calc-alkaline and peraluminous (Fig. 5b and c). The Al<sub>2</sub>O<sub>3</sub> contents range from 15 to 21 wt.% and the calculated Mg# numbers from 25 to 55 (Fig. 5d). Sr concentrations vary from 208 to 893 ppm, with low Y (2.2–13 ppm) and Yb (0.5–1 ppm) contents similar to the range observed in typical adakites (Defant and Drummond, 1990). The adakite-like geochemical features are further underlined by their fractionated REE patterns with a mean (La/Yb)<sub>N</sub> of 20.2 (Fig. 7c and d), depletions in Nb, Ta and Ti, enrichments in U and K, a mildly positive to absent Eu anomaly, and high Sr/Y and La/Yb but low Y and Yb. The absence of significant correlations between major and incompatible element ratios vs. SiO<sub>2</sub> indicates that the magma may have experienced other processes in addition to closed system fractional crystallization (Fig. 6).

## 4.3 Mineral chemistry

### 4.3.1 Amphibole

Amphibole phenocrysts are found as euhedral crystals with normal zoning in the trachyandesite and dacite samples and mostly as acicular or star-shape crystals in andesite dykes from the northern part of the study area. Selected amphiboles were analysed for major element compositions. Amphibole classification (after Leake et al., 2004) is based on the general chemical formula A<sub>0–1</sub>B<sub>2</sub>C<sub>5</sub><sup>V1</sup>T<sub>8</sub><sup>IV</sup>O<sub>22</sub>(OH; F; Cl)<sub>2</sub>. Because the water and halogen contents of the amphiboles are unknown, the amphibole formula

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is calculated to 23(O). Amphiboles are classified through the number of atoms per formula unit (apfu) of Ca and Na in the B site, (i.e.  $(Ca+Na)_B$ ) and amphiboles from the northern subvolcanic suite plot in the calcic amphibole field. These amphiboles have Si between 6 and 7 values per formula, characteristic of igneous amphiboles (Leake, 1978). The compositions of the amphiboles from the andesite, trachyandesite, trachydacite and dacite samples are shown in the  $Mg/(Mg + Fe^{2+})$  vs. Si classification diagram (Fig. 9). The amphiboles from the andesite samples are Mg-hastingsites to tschermakites, and those in the trachyandesite and trachydacite samples are tschermakite, while amphiboles from dacites are mostly magnesio-hornblendes (Fig. 9c). Mg-hastingsite can be separated from pargasite by values of  $Al^{VI}$  and  $Fe^{+3}$  cations per formula;  $Al^{VI}$  values in hastingsite are less than  $Fe^{+3}$  (Fig. 9b). According to the plot of  $^AK$  vs.  $^{IV}Al$  after Ridolfi and Renzulli (2012), our calcic amphiboles are in equilibrium with calc-alkaline liquids (Fig. 9d). A plot of six-fold Al ( $Al^{VI}$ , C site) vs. four-fold Al ( $Al^{IV}$ , T site) for amphiboles from all rock suites (Fig. 9e) indicates that Al preferentially resides in the tetrahedral site, which, together with the negative correlation between  $(Na + K)^A$  vs. Si p.f.u (Fig. 9f), implies the dominance of edenite-type ( $Na, K_A + Al^{IV} = Si^{IV}$ ) substitution for the presented amphiboles (cf. Murphy et al., 2012).

Although, the amphiboles are unzoned in the andesites, simple and oscillatory zoning is present in amphiboles from the trachyandesites and trachydacites. The simple-zoned amphibole phenocrysts are characterized by broad cores that commonly host small subhedral inclusions of Fe-Ti oxides (not shown). The cores of the simple-zoned phenocrysts are tschermakite and show a progressive rimward increase in Mg that is coupled with decrease in  $^{IV}Al$  and  $(Na + K)^A$ . The oscillatory zoned amphibole phenocrysts in the dacite and trachydacite samples are illustrated by alternating dark and bright euhedral zones in back-scattered images (Fig. 10a). This zoning is characterized by a tschermakitic core that also commonly hosts small inclusions of Fe-Ti oxides and apatite. The cores of these minerals shows a rimward decrease in Mg (from 3.44 in the core to 2.8 cations p.f.u. in the rim) and an increase in  $Al^{IV}$  (from 1.59 to 1.63 cation p.f.u) and  $(Na + K)^A$  (from 0.30 to 0.41). The cores are overgrown by a layer

of dark Mg-hornblende and show significantly lower  $Al^{IV}$  (1.36) and  $(Na + K)^A$  (0.28) compared to the cores. Interval changes in this zone are marked by decreases in Mg (from 3.17 to 2.62 cations p.f.u) and increases in  $Al^{IV}$  (from 0.26 to 0.46 p.f.u) and  $(Na + K)^A$  (0.17–0.47) toward the rim. The Mg-hornblende zones then grade into outer rims with ferrian-tschermakite composition, reflecting normal zoning with a decreasing trend in  $Fe^{+2}$  (from 1 to 0.41 p.f.u),  $Al^{IV}$  (1.77 to 0.59 p.f.u) and  $(Na + K)^A$  (from 0.43 to 0.33), while a very mild increase in Mg is observed (from 3 to 3.14 cation p.f.u) (Fig. 10b and c).

### 4.3.2 Plagioclase

Plagioclase occurs abundantly together with amphibole in the northern subvolcanic samples as individual phenocrysts, microphenocrysts, and microlites. Although plagioclase crystals commonly range between 0.5 and 3 mm, large phenocrysts up to 7 mm in length do occur in some dacite samples from the northern part. The southern rhyolite suite comprises plagioclase as phenocrysts and microlites. The composition of plagioclase phenocrysts in the andesite and trachyandesite rocks ranges from  $An_{53}$  to  $An_{70}$  (labradorite) and  $An_{36}$  to  $An_{56}$  (andesine), respectively. Plagioclase is the volumetrically dominant phenocryst phase in the trachydacite ( $An_{30-49}$ ) and dacite ( $An_{40-24}$ ) samples, but is rare in the rhyolites ( $An_{15-8}$ ). Despite some homogeneous plagioclase phenocrysts that occur in the andesites, most of the samples in the northern part exhibit plagioclase with complex zoning of variable types. Plagioclase microlites, in turn, are usually unzoned and have homogeneous compositions or minor normal zoning. Figure 11a shows the compositional spectrum of all plagioclase crystals analysed in the feldspars composition triangle. Notably, plagioclase phenocrysts with pronounced sieved textures are preferentially found in dacite rocks (Fig. 11b and c). Two types of primary zoning in plagioclase are dominant and include normal and oscillatory (complex) types:

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nesian hornblendes from the dacite samples range from 836 to 873 °C and estimated temperatures for amphiboles in the clots range between 874 and 974 °C.

Amphiboles from andesites yield pressures of on average 556 MPa (820–349) (Fig. 12a). The highest crystallization pressures (~ 820–700 MPa) correspond to the Mg-hastingsites from the basaltic andesite. Average pressure estimates of ~ 370 MPa (714–246) for tschermakites in trachyandesites and ~ 320 MPa (388–268) in trachydacites are derived. Mg-hornblendes in dacites yield pressure of ~ 190 MPa (226–145), while tchermakites from crystal clots gave an average pressures of ~ 395 MPa (780–217). To reconstruct the plumbing system, we used the density of the ophiolite complex (3.0 g cm<sup>-3</sup>) and applied this value to the sub-ophiolite units also (see Discussion).

The oxygen fugacity of a crystallising magma is reflected in the Mg content of resulting amphiboles and values of log (*f*O<sub>2</sub>) for Mg-hastingsites in andesites range from –10 to –8.9 (Fig. 12b). Values of log (*f*O<sub>2</sub>) for tschermakites from trachyandesites and trachydacites range from –11 to –9.5 and –11.0 to –9.8, respectively. Mg-hornblendes of dacites show log (*f*O<sub>2</sub>) from –12.1 to –10.9 (Fig. 12b). Tschermakites from crystal clots have a value of log (*f*O<sub>2</sub>) similar to trachyandesitic samples ranging between –10.4 to –9.3 (Fig. 12b). Oxygen fugacity generally increases from high *P-T* magnesiohastingsite (avg. 700 MPa, 990 °C) to lower *P-T* magnesiohornblende (avg. 190 MPa, 840 °C), consistent with the compositional variations known for calc-alkaline magmas from experiments (Gill, 1981; Martel et al., 1999; Müntener et al., 2001; Behrens and Gaillard, 2006).

A diagram of H<sub>2</sub>O<sub>melt</sub>-*T* of the amphibole suite is given in Fig. 12c, where the stability field of experimental amphiboles is outlined by a dotted curve. The experimental amphiboles comprise crystals with Al# ≤ 0.21 that were synthesized in equilibrium with melts overlapping the main Al<sub>2</sub>O<sub>3</sub> vs. SiO<sub>2</sub> pattern of extrusive rocks and glasses (cf. Ridolfi et al., 2010). All studied amphibole samples fall within the range of the experimentally determined equilibrium amphiboles with low-Al# (< 0.21) (Ridolfi et al., 2010). This Al-value is, in turn, in equilibrium with H<sub>2</sub>O<sub>melt</sub> values of 3.7 to 8.2 wt.% (±0.5) (cf. Ridolfi and Renzalli, 2012), which agrees with the H<sub>2</sub>O values known in calc-alkaline

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magmatic suites (e.g. Gill, 1981). The  $H_2O_{melt}$  contents obtained from Mg-hastingsites range between 5 and 8.3 wt.% ( $\pm 0.5$ ), while tschermakite and Mg-hornblende yield 4.2 to 10.3 and 5 to 7 wt.% ( $\pm 0.5$ ) melt  $H_2O$ , respectively. Tschermakitic clots suggest corresponding  $H_2O_{melt}$  values between 5.5 to 9.3 wt.% ( $\pm 0.5$ ).

#### 4.5 Pressure estimate from Plagioclase-melt thermobarometry

To test the results of the amphibole thermobarometry, the plagioclase thermobarometer of Putirka (2008), calibrated for hydrous systems, was applied. Standard error of estimation (SEE) for the plagioclase-melt thermobarometer is  $\pm 36^\circ C$  and  $\pm 247$  MPa (Putirka, 2008). The Kd [Ab–An] equilibrium test shows that  $An_{50-70}$  is in equilibrium with the whole rock composition of the andesites and these plagioclase-melt pairs record pressures between  $\sim 470$  and 335 MPa (Fig. 13a and b). The lower-anorthite plagioclase ( $An_{50-30}$  and  $An_{30-20}$ ) shows disequilibrium with the andesite nominal melt, but yields equilibrium with the available trachydacite and dacite compositions, respectively. Plagioclases with  $An_{50-30}$  record pressures between  $\sim 435$  and 190 MPa and plagioclase with  $An_{30-20}$  records between  $\sim 390$  and 130 MPa (Fig. 13a and b). Plagioclase in the southern rhyolites ( $An_{20-10}$ ) yield pressures between  $\sim 275$  and 110 MPa. The results of the plagioclase-melt thermobarometer record overall mid- to shallow crustal magma storage (468–110 MPa), thus broadly overlapping with the amphibole thermobarometry results.

#### 4.6 Pressure estimate from clinopyroxene-melt barometry

Following equilibrium testing between clinopyroxene and available mafic whole rocks as nominal melt (see Methods), pressure estimates were determined. The selected clinopyroxenes (sample *N* – 1, Fig. 14a and b) satisfy equilibrium conditions with the analysed andesite whole rock and using clinopyroxene-melt barometry after Putirka (2008, Eq. 32c), average pressures of  $\sim 600$  MPa (730 to 510) are retrieved (Fig. 14c). These results are higher than the results from plagioclase-melt barometry, but overlap

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ations of the amphibole reflect changes in  $T$ ,  $P$ ,  $p\text{H}_2\text{O}$ ,  $f\text{O}_2$  and melt composition (e.g. Rooney et al., 2010) and intense fractionation of amphibole will increase the Sr/Y ratios and decrease the Y concentrations in the resulting melt (Foley et al., 2013), driving compositions to that of typical adakite magmas. Crystal clots composed of amphibole have been suggested to represent clusters of phenocrysts (Garcia and Jacobson, 1979), the products of amphibole breakdown (Stewart, 1975), cooler wall-rock material (Humphreys et al., 2009), disrupted cumulates or crystal mush zones (e.g. Seeman, 2000; Chiaradia et al., 2011), or xenoliths (Chiaradia et al., 2009). Our investigation shows that the mineral compositions and the crystallization conditions ( $T$ ,  $P$ , and  $f\text{O}_2$ ) of the crystal clots overlap with the phenocryst assemblage of the host-rock, implying that the clots in the present system are associated with cooler wall-rock crystallisation.

The core to rim variations in the simple-zoned amphibole phenocrysts shows an overall rimward decrease in  $\text{Al}^{\text{IV}}$ ,  $(\text{Na} + \text{K})^{\text{A}}$  and  $\text{Fe}^{+2}$  accompanied by an increase in  $X_{\text{Mg}}$ . This pattern can be explained by gradual cooling during crystallization (e.g. Humphreys et al., 2006) where composition of the amphibole has likely been modified in response to changing melt composition controlled by plagioclase crystallization (cf. Holland and Blundy, 1994). For example, slight increases in melt- $\text{H}_2\text{O}$  result in dissolution and crystallization of An-rich compositions (Housh and Luhr, 1991). Although normal zoned plagioclases with small-amplitude variation are probably a result of minor local disequilibrium or small-scale pressure and temperature changes (Pearce and Kolisnik, 1990), the occurrence of disequilibrium phenocryst composition and textures such as oscillatory zoned amphibole and plagioclase and sieved textured plagioclase cannot be explained by simple closed system fractionation alone.

Oscillatory and especially complex zoning in plagioclase is often marked by resorption surfaces presented by rounded or wavy truncated interfaces (e.g. Fig. 11b) which is attributed to large-scale changes in temperature, pressure, melt- $\text{H}_2\text{O}$  content and/or melt composition and is frequently associated with magma recharge or mixing events. Pressure changes that exceed several Pascal are already capable of changing plagioclase composition (Nelson and Montana, 1992), which implies that the broad diffuse

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cores of the simple-zoned amphibole and oscillatory-zoned plagioclase phenocrysts then likely reflect crystallization under stable magma chamber conditions over longer periods of time (e.g. Foley et al., 2013). However, the presence of several successive zones with markedly increased An content in oscillatory-zoned plagioclase in the dacites and trachydacites (up to 15 mol %) likely reflects magma mixing and incoming batches of more mafic melts, in line with sieve-textured plagioclase phenocrysts that show An-rich overgrowths rims and embayments in quartz (when present). When hot and more mafic magma intrudes a colder, more differentiated reservoir, crystals near the mixing interface are re-melted and produce sieve textures and embayments (e.g. Donaldson and Henderson, 1988; Tepley et al., 2001; Andrews et al., 2008), but ultimately causes more anorthitic plagioclase to grow onto e.g. the remaining plagioclase fragments (e.g. Troll et al., 2004). This interpretation is supported by sieved textures plagioclase cores in dacites that are overgrown by rims with elevated  $X_{An}$  ( $An_{44}$ ) and Mg and Fe contents compared to their core ( $\sim An_{29}$ ).

The rhyolite/rhyodacite domes in the southern part of the Sabzevar zone have high Sr/Y and La/Yb ratios but low Y (2.2–13 ppm) and Yb (0.5–1 ppm) contents, resembling typical adakites as defined by Defant and Drummond (1990). However, the Sr contents (209–377.5 ppm) for 13 of 25 rhyolite samples are not as high as known from typical adakites (> 400 ppm). According to the available geochemical data, the southern post-ophiolite adakite rocks might have been derived from a different source compared to the northern adakite rocks which, however, awaits to be tested by isotopic methods. Notably, however, oceanic, slab-derived adakites are generally metaluminous, whereas the southern adakite rocks are peraluminous. Compared to the adakite rocks derived from partial melting of subducted oceanic crust, the southern adakite domes are high silica and potassium-rich ( $K_2O = 4.6–7$  wt.%), with low A/NK ratios, and are thus more akin to adakites of lower-crustal derivation (e.g. Wang et al., 2012) (Fig. 5c). There is no significant correlation between major and incompatible element ratios vs.  $SiO_2$  between the southern high-silica samples and the northern intermediate rocks, but instead there is a large compositional gap. A direct co-evolution of northern intermediate







in producing the diverse range of post-ophiolite rocks observed in the northern sector.

- 5 Our textural study documents that mafic magma recharge was driving the evolution of magmatic differentiation and disequilibrium features such as complex zoning and sieved textures in plagioclase, embayed quartz and zoned amphibole with high Mg# in outer margin and suggests that conditions changed for hotter and variably high  $fO_2$  conditions due to magmatic replenishments.
- 10 6. Selected plagioclases from southern rhyolites ( $An_{8-15}$ ) show pressures of crystallization between 273 to 110 MPa, implying a shallow upper crustal magma storage region for the southern sector also. According to geochemical characteristics, it seems that the southern peraluminous, silica-rich and high-K adakite rocks have been produced by partial melting of crustal material, possibly from thickened lower crust in response to prolonged underplating and repeated magmatic activity or from extensive assimilation of mid- to upper crustal materials.

15 *Acknowledgements.* We are most grateful to the staff of Centre for Experimental, Mineralogy, Petrology and Geochemistry (CEMPEG), Uppsala University, Sweden, especially Jaroslaw Majka and Abigail Barker for valuable help during microprobe analyses and Lisa Samrock for help with manuscript preparation. Financial support for this work was provided by the  
20 Ministry of Science and Technology of Iran, with complementary support through Uppsala University and the Swedish Science Foundation.

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**Table 1.** Major and trace element contents of representative post- ophiolite subvolcanic samples of NE Iran.

Samples	N-1	N-2	N-3	N-4	N-5	N-6	N-7	N-8	N-9	N-10	N-11	N-12	N-13	N-14	N-15	N-16
SiO <sub>2</sub>	55.01	57.35	57.42	58.28	58.29	58.74	58.74	55.53	56.86	58.81	58.97	59.49	59.28	60.10	60.41	62.56
TiO <sub>2</sub>	0.68	0.58	0.52	0.48	0.53	0.71	0.56	0.70	0.64	0.66	0.44	0.59	0.43	0.50	0.45	0.48
Al <sub>2</sub> O <sub>3</sub>	17.25	14.25	18.06	17.08	15.03	16.70	18.59	18.24	18.22	16.75	17.27	15.20	15.59	19.07	17.82	17.74
FeO	4.30	3.37	2.44	2.57	1.94	1.97	2.19	3.03	2.69	2.78	3.11	2.70	2.24	1.68	1.52	1.78
Fe <sub>2</sub> O <sub>3</sub>	2.87	3.37	4.53	2.57	1.94	1.97	4.06	4.55	4.03	1.85	2.07	2.70	1.49	3.12	2.28	2.68
MnO	0.14	0.13	0.13	0.15	0.18	0.21	0.12	0.11	0.11	0.24	0.16	0.13	0.17	0.04	0.06	0.09
MgO	4.23	6.03	3.91	3.25	2.83	3.82	2.61	4.98	4.38	2.76	1.86	3.74	4.56	2.06	4.62	3.06
CaO	5.21	5.38	7.06	3.96	3.51	1.50	6.44	4.31	4.40	5.25	5.63	3.08	4.59	5.21	4.12	5.41
Na <sub>2</sub> O	5.27	5.26	3.91	5.42	4.80	5.47	4.06	6.05	6.68	4.15	4.13	6.97	4.38	3.98	4.93	4.39
K <sub>2</sub> O	1.15	2.95	1.28	1.74	2.50	1.75	0.59	1.23	1.10	1.25	0.63	1.02	1.59	1.73	1.87	0.72
P <sub>2</sub> O <sub>5</sub>	0.13	0.11	0.19	0.11	0.06	0.07	0.16	0.11	0.15	0.18	0.10	0.10	0.07	0.18	0.15	0.11
L.O.I.	1.8	0.5	2.10	2.1	4.2	3.5	3.30	2.70	2.10	2.6	2.8	2.1	2.7	3.70	3.10	2.40
Sum	99.87	99.85	101.5	99.89	99.93	101.43	99.89	101.63	101.33	99.89	99.93	99.90	99.93	101.35	101.91	101.41
Ba	122	96	357.7	143	288	470	130.1	135.7	149.7	238	153	102	270	206.0	381.0	130.3
Co	44.3	54.3	21.90	35.6	28.6	30.5	13.10	24.80	21.00	46.7	27.4	18.9	44.2	8.70	15.60	13.50
Cs	0.1	0.1	0.60	0.5	1.4	1.4	0.40	1.00	0.10	1.6	0.2	0.2	0.6	0.50	0.60	0.20
Ga	16.82	12.69	17.40	16.8	14.02	16.41	17.70	16.40	17.50	16.72	16.85	14.67	15.82	16.30	15.80	15.90
Hf	1.8	1.62	2.00	0.94	1.22	1.49	2.20	2.10	2.20	2.89	1.18	2.41	1.11	2.70	2.90	2.40
Nb	5	5	2.00	5	5	8	2.70	1.70	1.90	8	5	5	5	4.10	4.80	2.40
Rb	4	12.4	19.80	12.7	39.1	18.5	9.20	22.60	10.20	17.6	3.1	5.7	7.7	22.40	31.80	10.10
Sr	390	64	780.6	350	403	465	366.1	463.9	208.6	476	407	138	783	470.0	731.7	507.1
Ta	0.2	0.3	0.10	0.3	0.4	0.6	0.20	0.10	0.20	0.5	0.3	0.2	0.5	0.30	0.40	0.10
Th	0.5	0.6	2.50	0.3	1.6	1.2	1.30	1.10	0.90	2.2	0.5	0.8	1.2	1.50	3.00	0.80
U	0.2	0.1	0.80	0.1	0.5	0.6	0.40	0.30	0.40	0.8	0.1	0.3	0.3	0.50	1.20	0.40
V	176	167	208.00	111	76	80	134.00	207.00	170.00	117	87	119	82	70.00	76.00	86.00
Zr	81	66	69.20	87	109	124	72.60	64.60	66.60	132	87	93	110	110.60	103.30	85.70
Y	17	14	11.70	14	10	11	14.60	17.30	17.10	14	14	19	8	14.50	8.70	16.80
La	3.3	4.7	10.90	3.7	6	5.8	7.50	5.80	5.60	7.8	5	4.3	7.3	8.80	11.90	6.10
Ce	9.13	11.31	23.40	9.97	14.03	14.18	17.20	13.10	13.00	18.36	11.46	11.1	16	21.40	22.20	15.80
Pr	1.4	1.7	2.94	1.4	1.9	1.9	2.31	1.92	1.96	2.3	1.6	1.9	2	2.70	2.66	2.25
Nd	6.2	7.5	12.70	6	7.2	7.4	10.70	9.60	9.30	8.3	7	7.1	7	10.90	10.60	10.30
Sm	1.9	2.3	2.60	1.7	1.7	1.7	2.30	2.70	2.60	1.9	1.7	2.2	1.5	2.50	2.10	2.50
Eu	0.8	0.8	0.79	0.7	0.5	0.6	0.72	0.77	0.80	0.6	0.6	0.8	0.4	0.79	0.59	0.72
Gd	2.7	2.8	2.16	1.7	1.7	1.6	2.81	3.12	3.02	2.1	2	2.4	1.1	2.67	1.97	2.80
Tb	0.4	0.4	0.38	0.3	0.3	0.2	0.45	0.55	0.52	0.4	0.3	0.5	0.3	0.43	0.30	0.49
Dy	2.3	2.8	2.06	1.5	1.5	1.7	2.62	3.21	2.80	2.5	1.9	2.6	1.1	2.54	1.63	2.88
Ho	0.6	0.5	0.37	0.4	0.3	0.3	0.50	0.61	0.57	0.5	0.3	0.7	0.2	0.43	0.29	0.54
Er	1.5	1.7	1.14	1	0.9	0.7	1.50	1.76	1.84	1.2	0.9	1.9	0.6	1.35	0.88	1.82
Yb	1.3	1.6	1.10	0.9	0.8	0.7	1.54	1.67	1.73	1.3	0.9	2	0.6	1.35	0.84	1.67
Lu	0.2	0.3	0.18	0.1	0.1	0.1	0.24	0.29	0.28	0.2	0.1	0.3	0.1	0.23	0.14	0.29

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Samples	N-17	N-18	N-19	N-20	N-21	N-22	N-23	S-24	S-25	S-26	S-27	S-28	S-29	S-30	S-31	S-32
SiO <sub>2</sub>	60.07	62.31	64.09	62.78	68.61	63.60	63.71	69.72	70.23	70.28	70.40	70.72	70.78	70.95	71.27	70.78
TiO <sub>2</sub>	0.48	0.42	0.32	0.41	0.15	0.47	0.31	0.02	0.04	0.10	0.04	0.07	0.07	0.08	0.07	0.08
Al <sub>2</sub> O <sub>3</sub>	17.13	15.51	17.04	18.31	18.82	17.29	16.78	20.64	20.24	18.73	19.81	18.67	16.60	18.2	17.14	15.44
FeO	2.43	2.33	1.53	1.63	0.81	1.80	1.88	0.25	0.30	0.51	0.28	0.41	0.36	0.44	0.42	0.40
Fe <sub>2</sub> O <sub>3</sub>	1.62	1.55	1.53	1.63	0.81	2.70	1.25	0.25	0.30	0.51	0.28	0.41	0.36	0.44	0.24	0.49
MnO	0.22	0.12	0.20	0.05	0.08	0.08	0.15	0.06	0.04	0.04	0.04	0.03	0.06	0.04	0.05	0.05
MgO	2.17	2.96	1.17	2.45	0.75	2.92	2.10	0.10	0.18	0.25	0.16	0.24	0.22	0.23	0.43	0.25
CaO	4.38	4.26	3.31	3.62	0.87	5.25	4.59	1.25	0.46	1.97	0.42	0.81	2.68	1.81	0.35	1.75
Na <sub>2</sub> O	4.58	4.19	4.69	5.27	6.51	4.51	4.35	4.57	5.38	5.33	5.37	5.73	4.85	5.26	6.32	5.13
K <sub>2</sub> O	1.54	1.35	2.13	2.05	2.24	0.77	1.11	3.57	3.62	2.90	3.48	3.35	1.92	2.67	3.46	2.62
P <sub>2</sub> O <sub>5</sub>	0.06	0.08	0.09	0.22	0.13	0.13	0.06	0.03	0.05	0.04	0.03	0.03	0.06	0.06	0.06	0.05
L.O.I.	2.6	2.4	1.9	3.10	1.60	1.90	1.8	1.10	0.70	0.80	1.10	1.00	3.50	1.20	0.80	1.4
Total	99.94	99.98	99.95	101.43	101.32	101.4	99.93	101.50	101.53	101.41	101.41	101.44	101.43	101.34	100.62	99.98
Ba	272	181	300	445.6	463.9	323	518.0	545.1	481.0	559.6	539.2	318.5	413.4	424.3	408	408
Co	22.2	49.5	26.6	9.30	1.20	13.60	31.8	2.00	2.00	2.00	1.00	2.00	1.00	1.00	1.00	36.9
Cs	0.4	0.4	1	0.90	0.80	0.20	0.7	< 0.5	0.50	1.00	< 0.5	1.10	1.20	1.30	0.50	1.3
Ga	15.97	16.28	15.36	16.80	17.70	15.80	15.71	2.60	1.80	0.80	1.80	17.40	17.20	17.80	16.50	17.98
Hf	2.69	1.95	0.58	3.20	2.70	2.80	0.9	18.70	16.10	16.40	16.00	1.90	2.10	2.40	2.00	0.96
Nb	9	8	15	5.80	7.70	2.60	5	2.10	1.90	2.00	1.90	5.20	5.40	5.00	5.20	5
Rb	9.3	11.6	42.4	38.70	43.10	10.5	10.9	6.80	3.70	5.00	4.10	65.50	35.70	52.70	65.30	60
Sr	550	283	358	848.5	673.5	400	670	70.10	73.60	58.00	68.30	811.00	512.2	810.6	208.2	774
Ta	0.6	0.4	0.9	0.40	0.60	0.10	0.5	337.80	344.50	894.50	335.10	0.40	0.30	0.30	0.40	0.6
Th	3.6	1.4	4.7	3.60	3.80	0.70	1.1	0.60	0.30	0.30	0.30	0.50	0.70	0.70	0.80	0.9
U	0.9	0.5	1	1.20	1.20	0.40	0.2	1.60	1.70	1.50	1.50	1.10	0.70	1.10	1.30	0.8
V	64	88	31	58.00	19.00	90.00	68	1.80	1.00	1.10	1.00	6.00	7.00	7.00	8.00	6
Zr	160	92	186	124.30	90.10	88.00	71	< 5	< 5	9.00	< 5	46.40	50.0	53.8	49.0	59
Y	12	10	15	9.20	9.20	17.00	5	38.40	40.50	61.40	38.80	2.90	2.70	3.20	2.70	3
La	12.6	4.9	15.9	13.00	15.40	6.90	6.1	6.30	5.70	3.40	5.50	2.70	2.50	3.40	2.40	3.3
Ce	26.61	11.37	35.1	26.40	31.50	16.30	12.27	3.80	5.50	5.10	5.90	4.70	4.60	6.80	4.50	7.05
Pr	3	1.7	4	3.04	3.60	2.34	1.4	7.50	10.10	9.20	10.20	0.63	0.59	0.82	0.53	0.9
Nd	10.9	6.2	13.6	11.10	12.90	10.10	5.9	0.93	1.21	1.14	1.29	2.70	2.20	3.20	2.10	3.3
Sm	1.8	1.7	2.4	2.10	1.90	2.50	1.5	3.50	4.30	4.20	5.10	0.50	0.40	0.60	0.40	0.5
Eu	0.7	0.6	0.6	0.61	0.55	0.69	0.4	0.80	1.00	0.80	1.00	0.14	0.16	0.18	0.16	0.2
Gd	1.9	1.7	1.7	1.59	1.59	2.83	0.8	0.26	0.25	0.24	0.29	0.39	0.38	0.50	0.38	0.5
Tb	0.3	0.3	0.4	0.27	0.26	0.48	0.1	1.02	0.96	0.61	0.97	0.08	0.07	0.09	0.08	< 0.1
Dy	1.4	1.6	1.7	1.46	1.39	2.74	0.7	0.18	0.17	0.10	0.16	0.39	0.39	0.45	0.37	0.5
Ho	0.4	0.4	0.4	0.29	0.26	0.59	0.1	0.96	0.77	0.54	0.91	0.08	0.07	0.09	0.08	0.1
Er	1	1	0.9	0.86	0.78	1.70	0.4	0.18	0.16	0.10	0.16	0.21	0.22	0.27	0.20	0.2
Yb	1.1	0.8	0.7	0.77	0.78	1.73	0.4	0.52	0.43	0.25	0.41	0.17	0.18	0.22	0.19	0.1
Lu	0.2	0.1	0.1	0.14	0.13	0.30	0.1	0.47	0.38	0.26	0.42	0.03	0.03	0.03	0.03	0.1

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Samples	S-33	S-34	S-35	S-36	S-37	S-38	S-39	S-40	S-41	S-42	S-43	S-44	S-45	S-46	S-47
SiO <sub>2</sub>	71.39	71.17	71.59	72.30	72.59	73.22	72.09	72.30	72.32	72.46	72.54	72.64	72.92	73.14	73.62
TiO <sub>2</sub>	0.08	0.10	0.02	0.03	0.04	0.02	17.64	17.53	17.62	16.88	16.94	17.23	17.52	17.55	0.03
Al <sub>2</sub> O <sub>3</sub>	16.63	16.34	15.24	15.53	14.90	16.45	0.32	0.34	0.38	0.34	0.28	0.37	0.35	0.36	15.19
FeO	0.66	0.41	0.33	0.36	0.34	0.39	0.32	0.34	0.38	0.34	0.28	0.37	0.35	0.36	0.25
Fe <sub>2</sub> O <sub>3</sub>	0.66	0.50	0.40	0.44	0.42	0.39	0.12	0.18	0.24	0.13	0.14	0.18	0.16	0.16	0.31
MnO	0.15	0.05	0.09	0.09	0.04	0.08	1.52	0.40	0.26	0.51	0.80	0.60	1.34	1.63	0.05
MgO	0.53	0.26	0.20	0.25	0.21	0.13	5.12	6.50	7.07	5.57	5.27	6.22	5.58	5.17	0.15
CaO	0.97	1.85	0.37	1.26	0.42	1.19	3.48	3.10	2.65	4.21	3.55	3.17	2.36	2.32	0.92
Na <sub>2</sub> O	5.27	5.50	5.80	5.18	5.43	5.10	0.03	0.05	0.05	0.02	0.02	0.05	0.01	0.01	5.35
K <sub>2</sub> O	3.50	2.76	2.91	2.56	3.47	3.75	0.06	0.04	0.06	0.03	0.01	0.06	0.01	0.01	3.33
P <sub>2</sub> O <sub>5</sub>	0.17	0.05	0.08	0.07	0.05	0.05	0.06	0.03	0.03	0.06	0.09	0.03	0.07	0.06	0.04
L.O.I.	1.40	0.4	1.4	0.9	1	0.80	0.90	0.70	0.60	1.00	1.50	0.60	0.80	0.70	0.3
Total	101.42	99.97	99.96	99.98	99.98	101.53	101.63	101.50	101.61	101.52	101.42	101.55	101.41	101.45	99.97
Ba	558.9	416	541	471	479	493.8	507.5	466.5	445.5	510.1	502.8	487.8	310.8	304	508
Be	2.00	38.6	31.7	14.3	34.9	3.00	1.00	2.00	1.00	3.00	2.00	2.00	1.00	1.00	42.2
Cs	1.30	0.4	0.9	0.6	2.3	3.20	2.50	0.70	0.50	1.30	2.00	1.20	1.20	1.70	2.2
Ga	17.40	17.2	17.66	17.67	16.22	16.90	18.20	18.40	16.90	18.10	16.80	17.20	18.30	19.80	17.5
Hf	3.20	0.66	0.96	1.09	1.18	2.10	2.00	1.90	1.40	2.10	2.10	1.90	1.90	1.70	1.01
Nb	9.40	5	12	7	5	9.30	7.20	6.40	6.70	9.30	7.10	6.30	10.60	10.60	9
Rb	70.90	49.3	60.4	52.7	68.2	83.10	72.70	67.80	52.40	92.10	71.30	65.90	46.00	47.50	74.5
Sr	338.8	790	350	221	362	365.7	475.5	537.7	377.5	459.9	246.7	644.3	283.2	314.10	523
Ta	0.80	0.6	1	0.7	0.6	0.70	0.60	0.50	0.40	0.80	0.70	0.50	1.00	1.00	0.9
Th	4.80	1	3.1	2.7	0.8	1.70	1.20	1.00	0.10	1.00	2.10	1.20	0.40	0.30	1.7
U	2.60	0.7	1.5	1.3	1.3	1.90	1.40	1.70	1.10	2.40	2.00	1.50	1.90	1.70	1.7
V	10.0	7	1	1	2	< 5	< 5	5.00	< 5	< 5	< 5	< 5	< 5	< 5	1
Zr	74.0	67	52	45	41	39.50	41.1	43.9	31.60	39.4	41.9	43.70	24.80	24.70	42
Y	13.1	3	7	7	5	3.60	5.60	4.70	2.20	3.40	6.40	4.30	4.80	4.90	5
La	15.6	4.3	9.6	8.9	1.8	5.10	5.10	4.40	1.00	4.60	4.10	3.40	3.00	2.90	4.8
Ce	30.8	9.12	20.09	17.59	4.07	9.90	10.0	7.10	1.40	9.00	7.90	6.90	4.70	5.30	9.53
Pr	3.72	1	2.2	2	0.5	1.19	1.19	0.91	0.17	1.08	0.98	0.84	0.61	0.63	1.2
Nd	13.4	3.7	8.4	7.2	1.9	5.00	4.70	3.60	0.70	4.10	3.30	3.60	2.50	2.50	4.6
Sm	2.50	0.6	1.4	1.3	0.3	1.00	0.90	0.80	0.20	0.80	1.00	0.70	1.00	1.00	0.9
Eu	0.42	0.2	0.4	0.3	0.2	0.24	0.24	0.27	0.13	0.24	0.23	0.26	0.30	0.32	0.2
Gd	2.22	0.5	1.2	1.5	0.7	0.94	0.87	0.78	0.28	0.83	1.04	0.64	1.23	1.20	0.7
Tb	0.37	< 0.1	0.2	0.2	0.1	0.14	0.17	0.14	0.06	0.14	0.18	0.14	0.19	0.21	0.2
Dy	2.15	0.3	1	1.1	0.7	0.72	0.85	0.71	0.36	0.63	0.96	0.68	0.89	1.03	0.8
Ho	0.37	0.1	0.2	0.2	0.2	0.09	0.14	0.12	< 0.05	0.07	0.16	0.12	0.11	0.11	0.2
Er	1.07	0.1	0.4	0.7	0.4	0.17	0.41	0.33	0.13	0.14	0.45	0.30	0.19	0.21	0.3
Yb	1.06	0.1	0.4	0.6	0.5	0.11	0.40	0.33	0.05	0.10	0.46	0.32	0.08	0.11	0.3
Lu	0.17	0.1	0.1	0.1	0.1	0.01	0.06	0.05	0.01	0.01	0.07	0.05	0.01	0.01	0.1

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**Table 3.** Representative compositions of plagioclases and clinopyroxenes from post-ophiolite rocks in Sabzevar zone.

Mineral	Oscillatory-zone Plagioclase			Sieved texture Plagioclase		CPx	CPx
	No. Position	N-17 Core, An <sub>29</sub>	N-17 inner zone, An <sub>55</sub>	N-17 Rim, An <sub>44</sub>	N-18 Core, An <sub>25</sub>	N-18 Rim, An <sub>47</sub>	N-1 Core, Mg ≠ 0.8
SiO <sub>2</sub>	63.58	54.97	60.38	67.11	58.68	48.94	52.7
TiO <sub>2</sub>	0	0.054	0	0.02	0.02	1.44	0.32
Al <sub>2</sub> O <sub>3</sub>	23.72	28.27	25.64	22.22	27.07	4.91	2
FeO	0.08	0.22	0.17	0.12	0.41	7.41	6.06
MnO	0	0.03	0.03	0.00	0	0	0.21
MgO	0.02	0	0	0.04	0.02	13.67	16.52
CaO	5.61	11.32	8.09	4.09	9.54	22.91	21.5
Na <sub>2</sub> O	7.34	5.1	5.64	6.59	5.69	0.42	0.43
K <sub>2</sub> O	0.36	0.14	0.26	0.46	0.23	0	0.01
Cr <sub>2</sub> O <sub>3</sub>	0	0	0	0	0	0	0.25
Total	100.72	100.11	100.22	100.67	101.66	99.71	100.00

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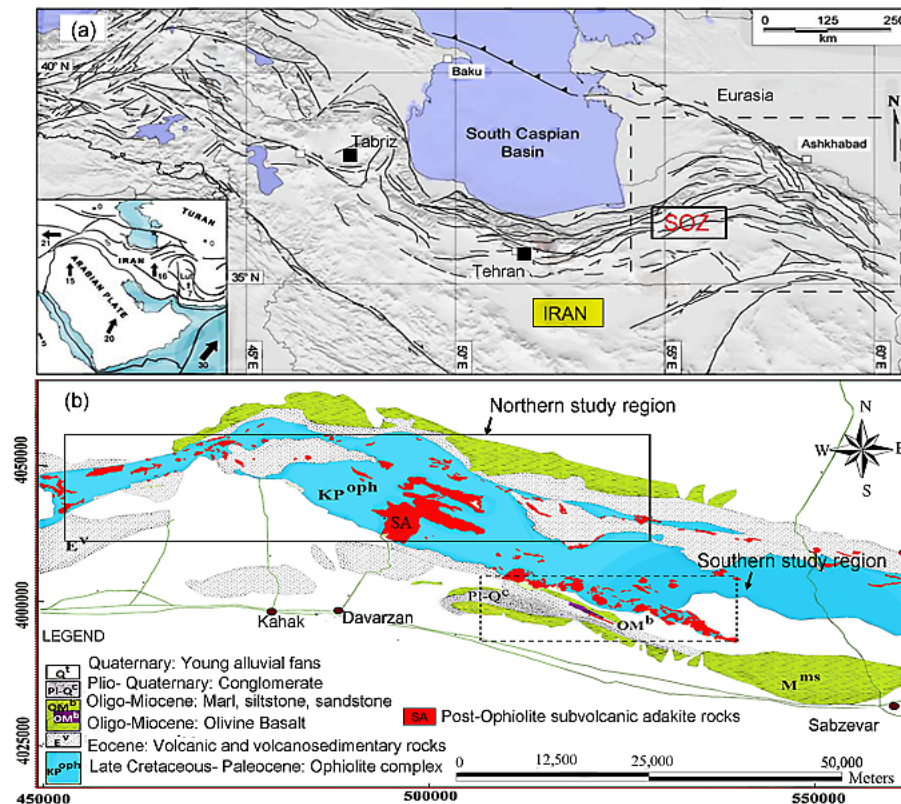
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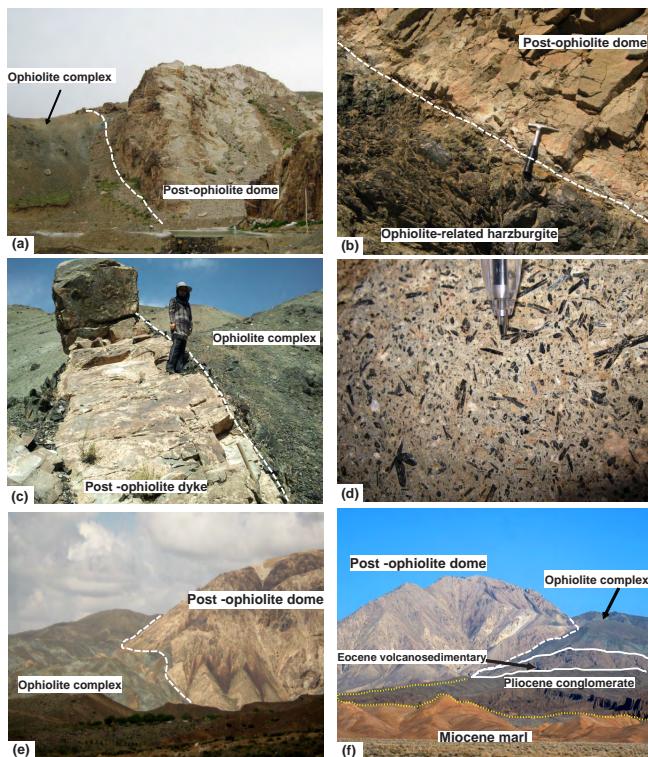


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**Figure 1.** (a) Shaded relief image (after Shabaniyan et al., 2012) showing the location of Sabzevar Ophiolitic Zone (SOZ). Lower left inset shows the geodynamic setting of Arabia-Eurasia collision/subduction framework. Black arrows and associated numbers represent the present-day Arabia-Eurasia plate movement velocities after Reillinger et al. (2006) in  $\text{mm yr}^{-1}$ . (b) Simplified geological map of the Sabzevar area and the post-ophiolite subvolcanic rocks. The northern (dash square) and southern (dotted square) sector of the study area are indicated.



**Figure 2.** Field photographs. **(a)** and **(b)** northern grey dome of trachydacite exposed in ophiolite-related harzburgite. **(b)** Contact between post-ophiolite subvolcanic rock and host harzburgite. **(c)** Photograph of post-ophiolite dyke in the northern sector. **(d)** Presence of star-shape amphibole aggregates in the andesitic dyke. **(e)** and **(f)** photographs of southern post-ophiolite domes that outcrop in the ophiolite complex. Eocene volcanoclastic complex, Miocene sedimentary and Pliocene conglomerates are seen in **(f)**.

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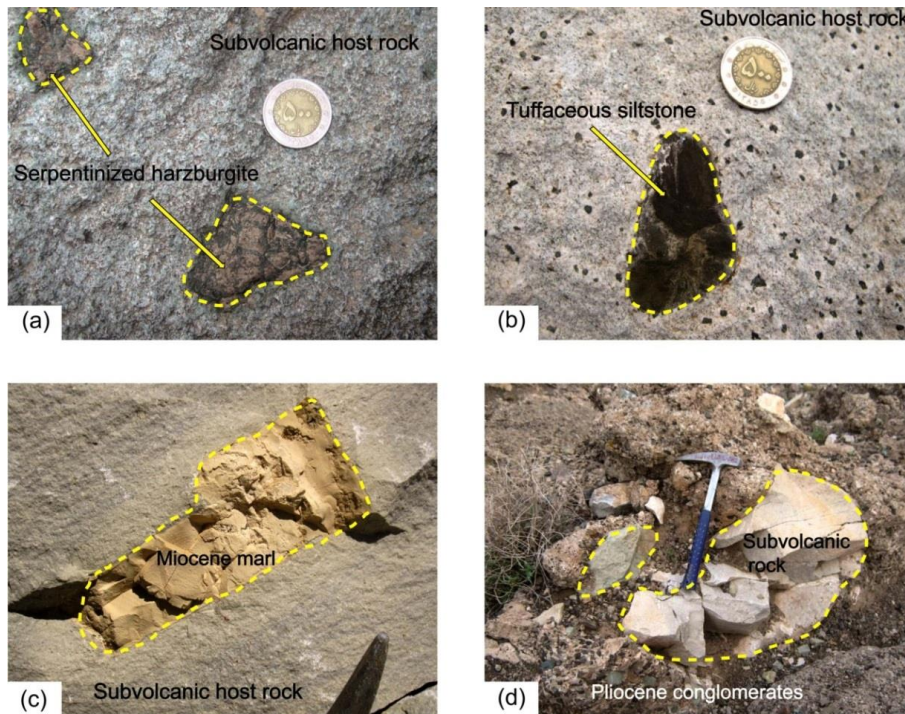
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**Figure 3.** (a–c) Field photographs of host rock xenoliths. (a) Harzburgite belongs to ophiolite complex. (b) Eocene tuffaceous siltstone. (c) Miocene marl. (d) Rhyolitic fragments in the Pliocene conglomerate.

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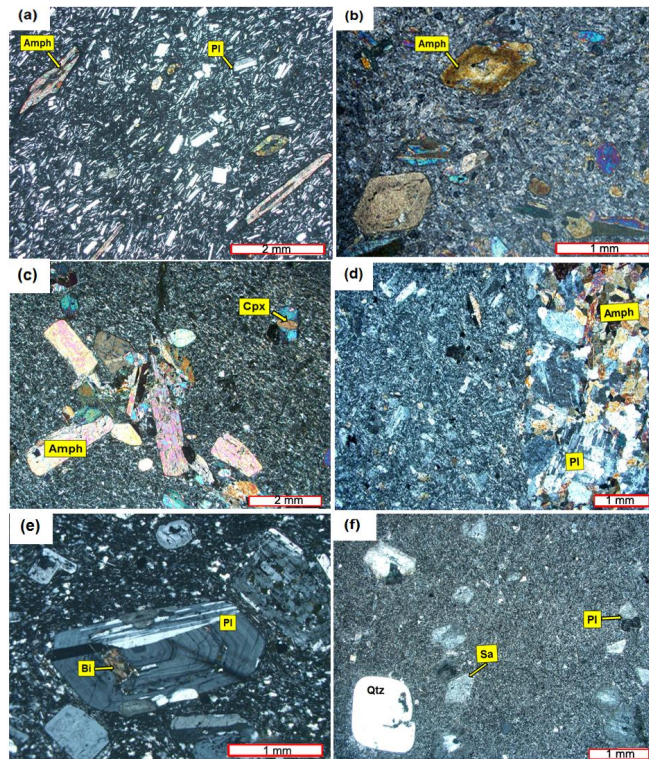
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**Figure 4.** Representative photomicrographs (XPL). **(a)** Flow texture and presence of acicular amphibole and plagioclase in andesite. **(b)** Euhedral to subhedral amphiboles with compositional zoning in trachyandesite. **(c)** association of amphibole aggregates and clinopyroxene in andesite. **(d)** Photomicrograph showing contact between crystal clot containing plagioclase and amphibole and host trachyandesite rock. **(e)** Zoned plagioclase phenocryst in decite. **(f)** Embayment and rounded quartz, altered sanidine and plagioclase in rhyolite.

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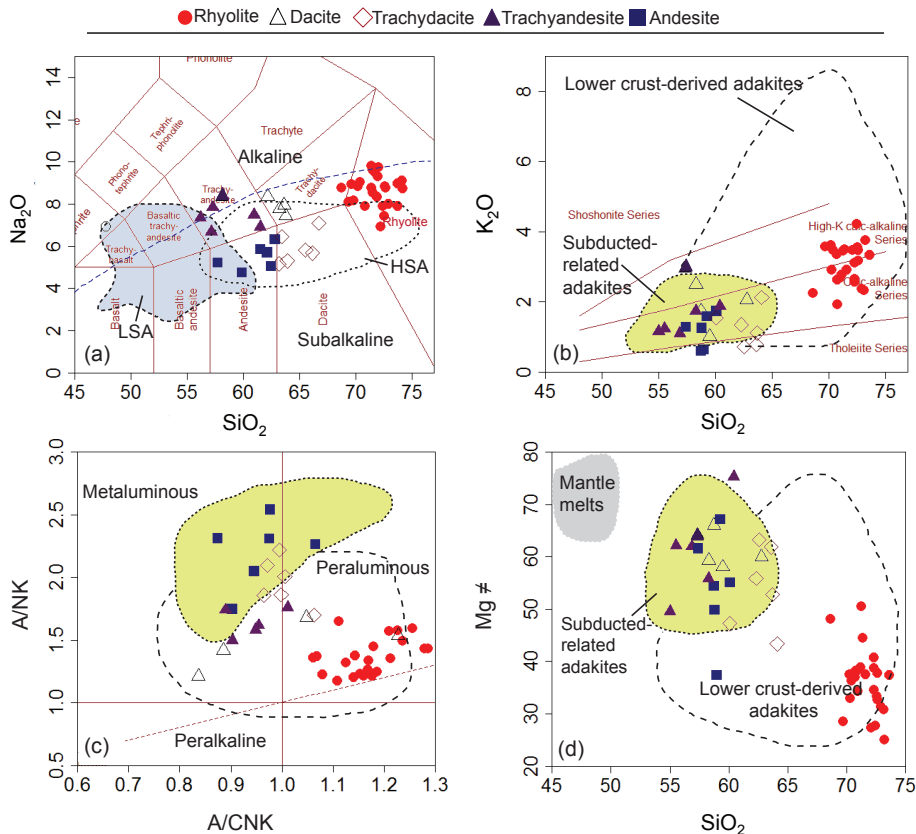
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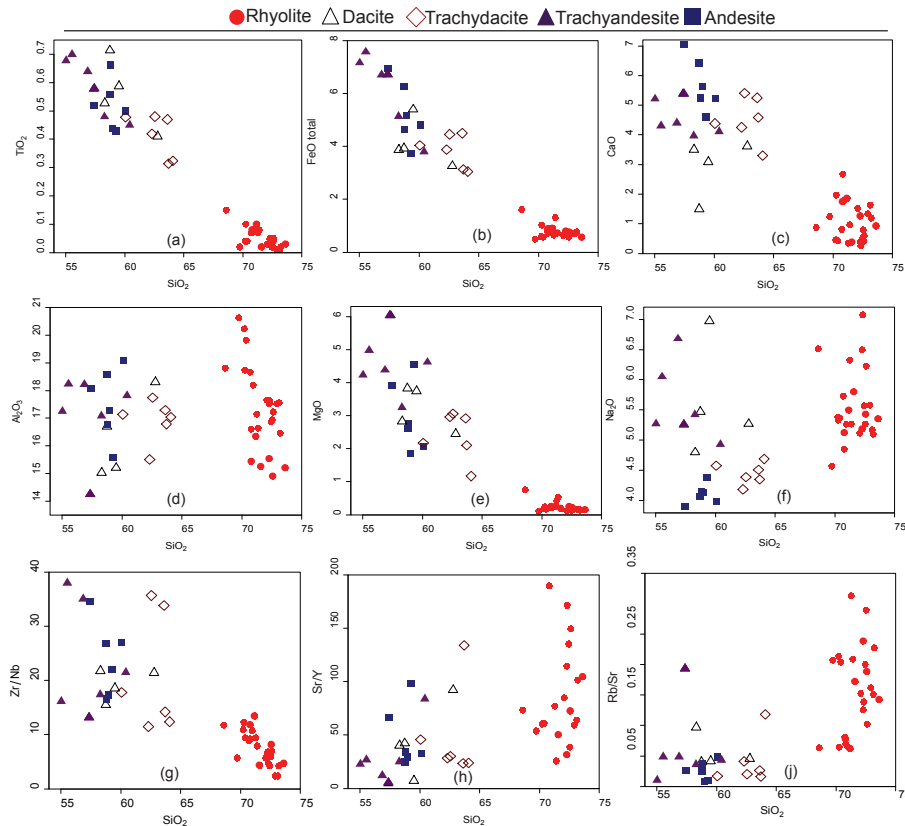
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**Figure 5.** Selected major element plots for the subvolcanic post-ophiolite rocks. **(a)** Total alkalis vs. silica diagram after Le Bas et al. (1986). The boundary between alkaline and subalkaline series is after Irvine and Baragar (1971), and the fields of low-silica adakites (LSA) and high-silica adakites (HSA) are after Martin et al. (2005). **(b)** K<sub>2</sub>O vs. SiO<sub>2</sub> diagram (after Rollinson, 1993); **(c)** A/CNK vs. A/NK diagram (Maniar and Piccoli, 1989); **(d)** Mg# vs. SiO<sub>2</sub> diagram. The fields of subduction-related adakites and lower crust-derived adakites are from Guan et al. (2012).

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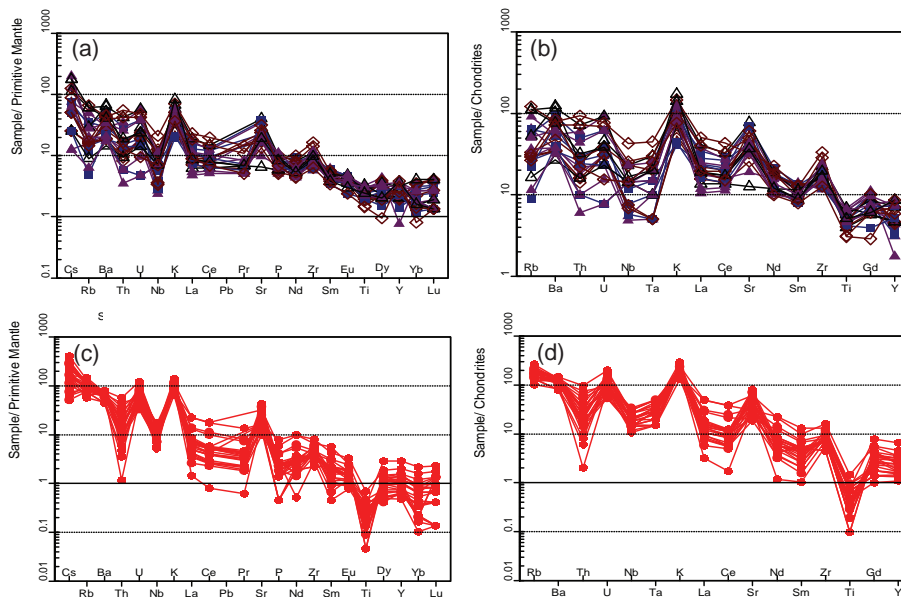


**Figure 6.** Harker diagrams of the major oxides (wt.%) and incompatible element ratios vs. SiO<sub>2</sub> for the studied rocks.



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**Figure 7.** Normalized REE and trace element patterns for the northern (**a** and **b**) and southern (**c** and **d**) subvolcanic rocks. Primitive-mantle and chondrite data are from Sun and McDonough (1989).

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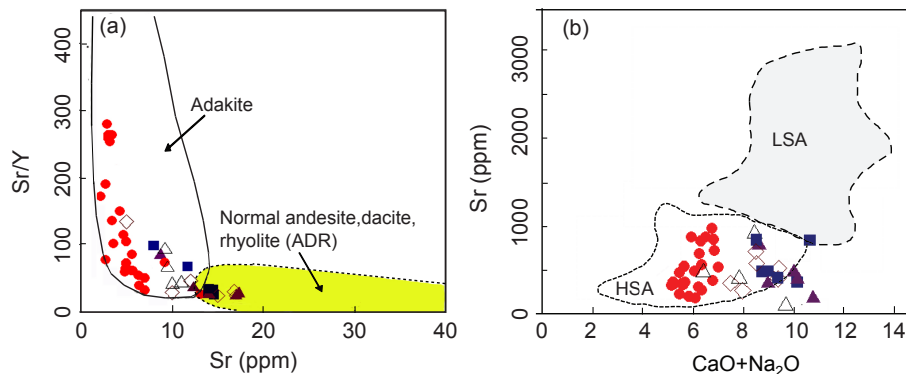
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**Figure 8.** (a) Plot of Sr/Y vs. Y for the studied rocks. Fields of adakite and arc normal rocks are from Petford and Atherton (1996) and Defant and Drummond (1990). (b) Plot of Sr vs. CaO + Na<sub>2</sub>O (wt.%) showing the chemical differences between low silica adakites (dashed field) and high silica adakites (dotted field) (after Castillo, 2012).

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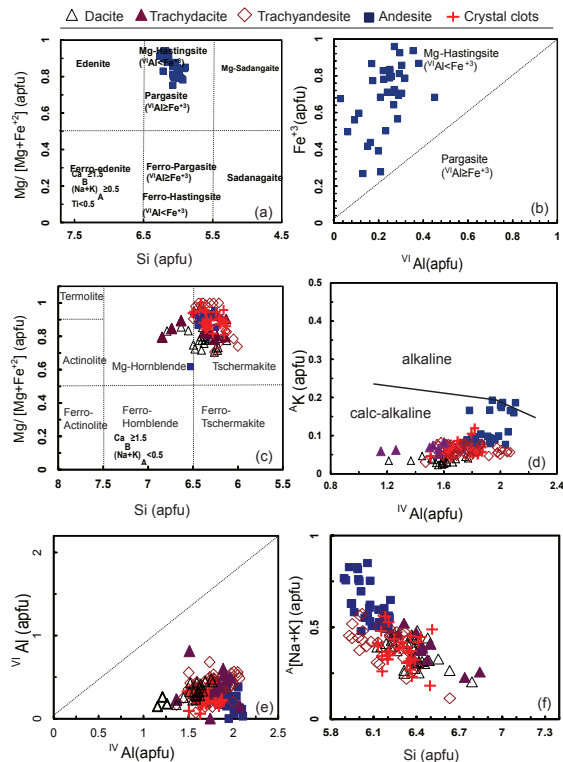
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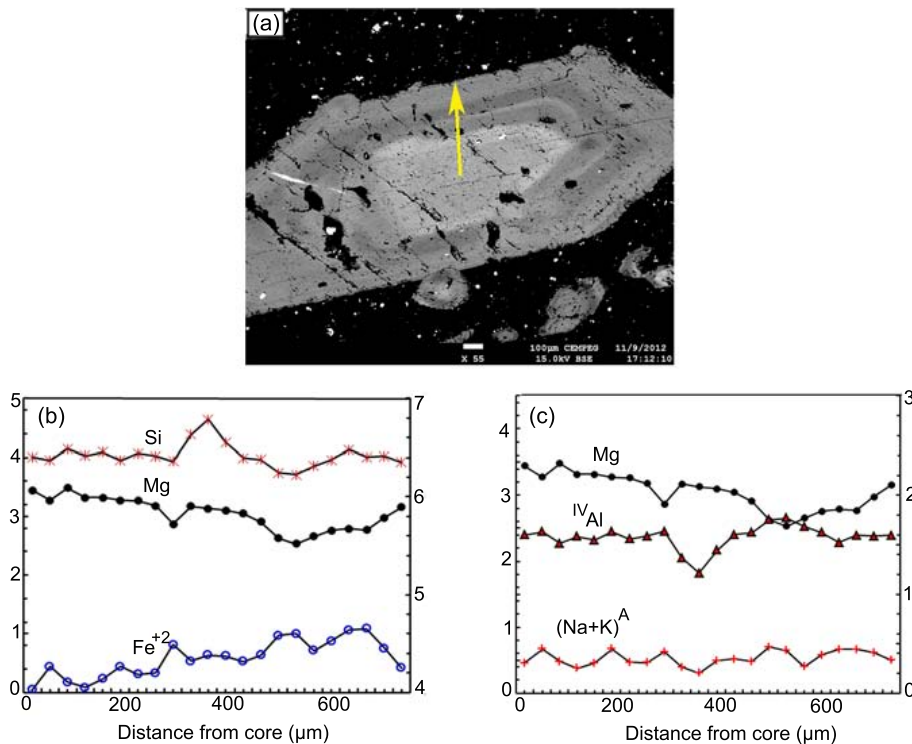
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**Figure 9.** Major element classification diagrams for amphiboles. **(a)** Amphiboles in the Mg-andesite samples. **(b)** Amphiboles in andesites with  $^{VI}Al < Fe^{+3}$  fall in the Mg-hastingsite field. **(c)** Amphiboles in the trachyandesite, trachydacite and dacite samples and crystal clots on the Mg/(Mg + Fe<sup>2+</sup>) vs. Si classification diagrams (after Leake, 1978; Leake et al., 1997, 2004). **(d)** Plot of AK vs. IVAI after Ridolfi and Renzulli (2012). Our calcic amphiboles located in the calc-alkaline field. **(e)**  $^{VI}Al$  plotted against  $^{IV}Al$  per formula unit and **(f)**  $(Na + K)^A$  vs. Si per formula unit for amphiboles from all samples. Procedure for assigning Al according to Leake (1978) and Leake et al. (2004). Solid black line in **(e)** indicates slope of 1.

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**Figure 10.** (a) Back-scatter electron microprobe image of a representative amphibole phenocryst from a trachydacite sample that comprises distinct alternating dark and light zones. (b) and (c) Major element traverses in multiple-zoned amphibole phenocryst from (a). Arrow indicates position of rim to core traverse. Scale bars represent 100 μm.

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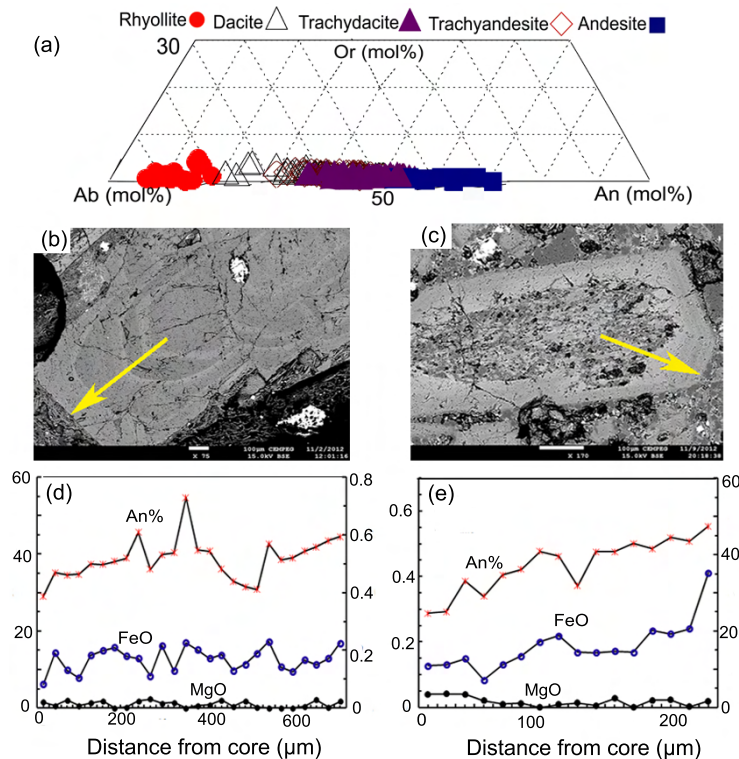
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**Figure 11.** (a) Composition of all analyzed plagioclase data points ( $n = 212$ ) in the feldspars composition triangle. Plagioclase composition in the samples of the northern sector vary between  $An_{20}$  and  $An_{70}$ . Plagioclase in rhyolites from the south part range from  $An_{20}$  to  $An_8$ . The BSE images of selected plagioclase phenocrysts show (b) oscillatory zoned plagioclase with abrupt increasing in An contents (up to 15 mol %) toward the rim and (c) sieve-textured plagioclase mantled by high An rim. (d) and (e) zoning profiles showing variations in  $X_{An}$  and MgO and FeO for the oscillatory zoned plagioclase and the sieve-textured plagioclase in (b) and (c), respectively. Yellow arrows indicate core to rim profile.

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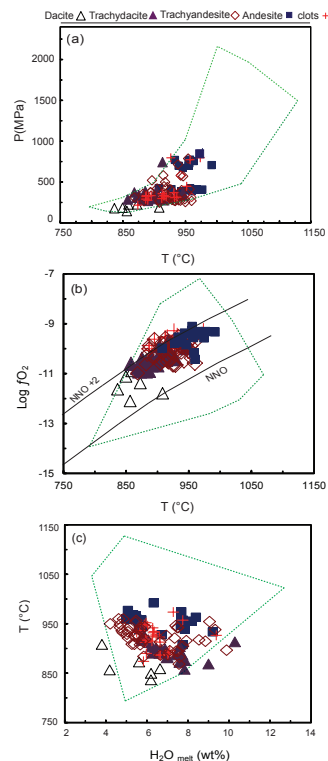
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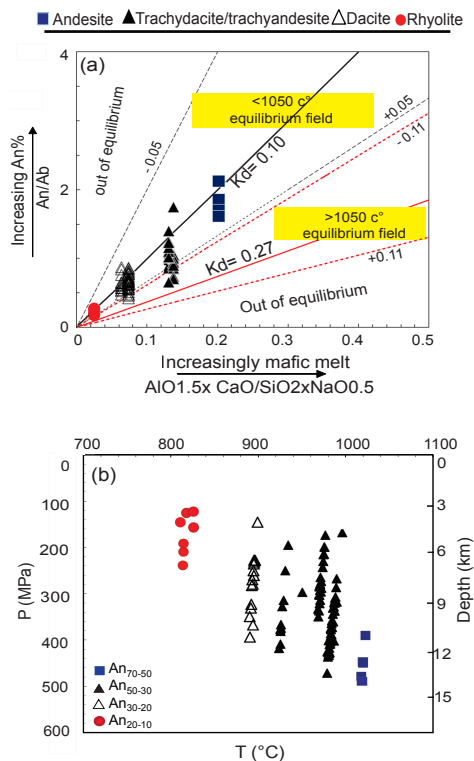
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**Figure 12.** (a)  $P$ - $T$  diagram for calcic experimental amphiboles after the Ridolfi and Renzulli (2012) with low uncertainties ( $T \pm 23.5^\circ\text{C}$ ,  $P \pm 11.5\%$ ). Our amphiboles plot in the field of experimentally re-produced amphiboles (after Ridolfi and Renzulli, 2012). (b)  $\text{Log}(f\text{O}_2)$ - $T$  and (c)  $T$ - $\text{H}_2\text{O}_{\text{melt}}$  diagrams for the studied amphiboles as obtained by amphibole thermobarometry after Ridolfi and Renzulli (2012). This method indicates typical uncertainties for  $\text{log } f\text{O}_2$  errors ( $\pm 0.2$  log unit) and  $\text{H}_2\text{O}_{\text{melt}}$  ( $\pm 0.8$  wt.%). The NNO and NNO+2 curves are from O'Neill and Pownceby (1993).

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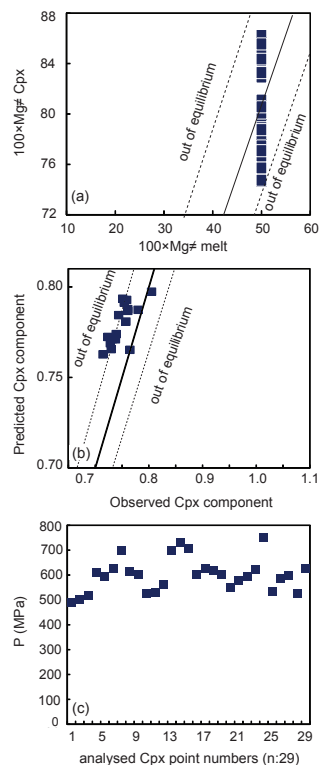
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**Figure 13.** (a) Equilibrium test for plagioclase and four possible melts. The andesite samples appear to be in equilibrium with high anorthite plagioclase (An<sub>70–50</sub>), while the trachyandesite/trachydacite and dacite are in equilibrium with (An<sub>50–30</sub>) and (An<sub>30–20</sub>), respectively. Rhyolite, in turn, is in equilibrium with low anorthite plagioclase (An<sub>20–10</sub>). (b) Results of plagioclase-melt thermobarometry after (Putirka, 2008). SEE for the plagioclase-melt thermobarometer are  $\pm 36^\circ\text{C}$  and  $\pm 247\text{ MPa}$ .

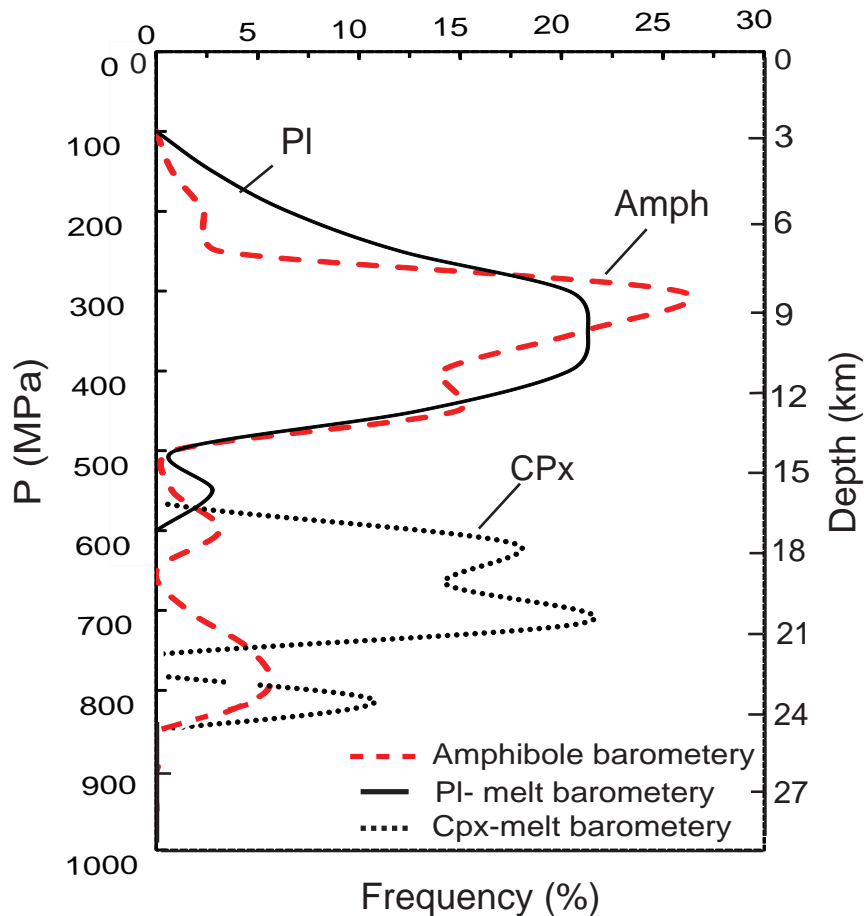
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**Figure 14.** (a) Test for equilibrium using the  $K_d[\text{FeMg}]$  between clinopyroxene and nominal melt (basaltic andesite sample [no. N-1]). The result shows  $K_d[\text{FeMg}]$  values close to the ideal of 0.27 (Putirka, 2008). (b) Test for equilibrium using predicted vs. observed clinopyroxene components of diopside + hedenbergite derived using the nominal equilibrium melt (basaltic andesite dyke). Pressure calculated for selected clinopyroxene using clinopyroxene-melt barometry after Putirka (2008).





**Figure 15.** A comparison of the results of amphibole, plagioclase-melt and clinopyroxene-melt barometries that indicate two distinct magma storage regions in the crust, one around a depth of ~ 6–9 km and the other at ~ 18–27 km depth.

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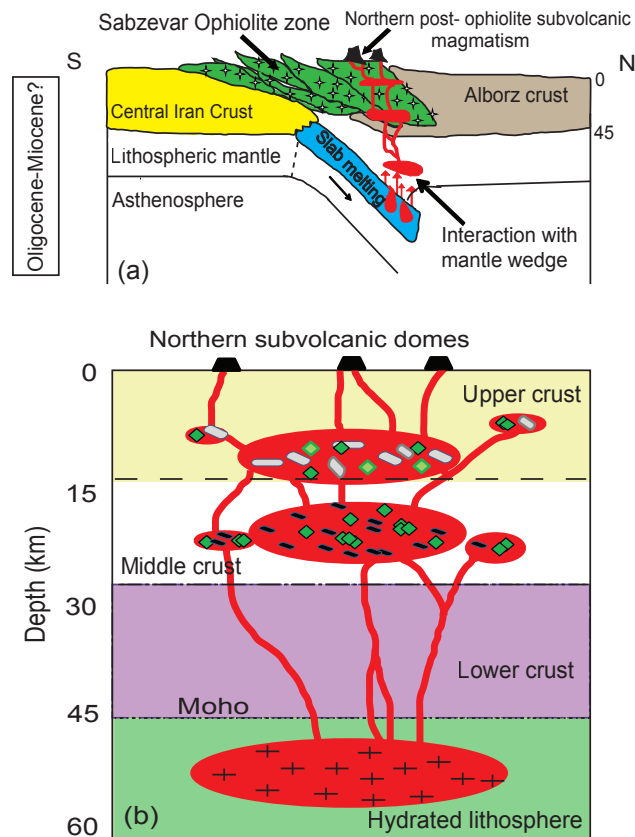
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**Figure 16.** (a) Tectonic framework illustrating northward subduction of Sabzevar oceanic crust (eastern branch of Neo-Tethys) beneath eastern Alborz zone. (b) Schematic illustration of the magma plumbing system for subvolcanic adakite-like rocks in northern part of Sabzevar belt based on the derived thermobarometric data.