



- 1 Polycyclic aromatic hydrocarbon in urban soils of the Eastern European megalopolis: distribution, source
- 2 identification and cancer risk evaluation
- 3 Shamilishvily George Avtandilovich¹ (corresponding author), e-mail: george199207@gmail.com, +79217868236
- 4 Abakumov Evgenii Vasil'evich¹, *e-mail: e_abakumov@mail.ru*
- 5 Gabov Dmitrii Nikolaevich², e-mail: gabov@ib.komisc.ru
- 6 ¹St. Petersburg State University, Dept. of Applied Ecology, St. Petersburg, Russia
- 7 ²Komi Biological Institute of the Russian Academy of Sciences, Syktyvkar, Russia

8 Abstract

9 The study explores qualitative and quantitative composition of 15 priority PAHs in urban soils of some 10 parkland, residential and industrial areas of the large industrial center in the Eastern Europe on example of Saint-11 Petersburg (Russian Federation). Aim of the study was to test the hypothesis on the PAH loading differences 12 between urban territories with different land use scenarios. Qualitative and quantitative determination of PAHs in 13 soils was carried out by reverse-phase high-performance liquid chromatography (HPLC). Benzo(a)pyrene toxic equivalency factors (TEFs) were used to estimate benzo(a)pyrene equivalent (BaPeq) concentrations in order to 14 15 evaluate carcinogenic risk of soil contamination with PAHs. Results of the study demonstrated that soils within 16 urban areas with different land utilization types are characterized by common loads of PAHs generally attributed to high traffic activity in the city. Considerable levels of soil contamination with PAHs were noted. Total PAH 17 concentrations ranged from 0.33 to 8.10 mg·kg⁻¹ and showed no significant differences between land utilization 18 19 types. The common tendency in PAH distribution patterns between investigated sites clearly indicates the common 20 source of PAHs in urban soils. A larger portion of high molecular weight PAHs along with determined molecular 21 ratios suggest the predominance of pyrogenic sources, mainly attributed to combustion of gasoline, diesel and oil. 22 Petrogenic sources of PAHs have a significant portion as well defining the predominance of petroleum associated 23 low molecular weight PAHs such as phenanthrene. Derived concentrations of 7 carcinogenic PAHs as well as 24 calculated BaP total potency equivalents were multiple times higher than reported in a number of other studies, 25 indicating a significant risk for human health in case of direct contact. The obtained BaPea concentrations of the sum of 15 PAHs ranged from 0.05 to 1.39 mg·kg⁻¹. A vast majority of examined samples showed concentrations above 26 27 the safe value of 0.6 mg·kg⁻¹ (CCME, 2010). One-way ANOVA results showed significant differences in levels of 28 pyrene, fluoranthene and phenanthrene - the most abundant individual PAHs in examined sampled, between 29 parkland, residential and industrial land uses, suggesting the influence of land use factor on distribution of these 30 pollutants. Exposure to these soils through direct contact probably poses a significant risk to human health from 31 carcinogenic effects of PAHs, even in urban parklands.

1. Introduction

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33 The quantity of toxic organic substances is extremely high, but in the world practice the evaluation of 34 contamination levels of certain areas is produced mostly for polycyclic aromatic hydrocarbons (PAHs), an 35 ubiquitous organic pollutants in environments, particularly in soils and sediments. PAHs are a large group of 36 persistent organic compounds (POPs) consisting of several hundred individual homologues and isomers containing 37 at least two condensed aromatic rings. Their input to the environment has both natural and anthropogenic origins. 38 Natural sources includes releases from vegetation fires, diagenetic processes and volcanic exhalations (ATSDR 39 1995; Wilcke 2000). In turn, anthropogenic PAHs occur from pyrolytic processes, especially incomplete combustion 40 of organic during industrial activities, domestic heating, waste incineration, transportation and power generation 41 (ATSDR 1995; Wilcke 2000; Dyke et al. 2003). It is believed that by far most PAHs are released into environment 42 by anthropogenic combustion of wood and fossil fuels (Wilcke 2000). Sign of anthropogenic contamination of soil 43 with PAHs are even detected in such remote places as Antarctic Stations, which origin is doubted, whether it has 44 natural sources, i.e. decomposition of plant and guano materials, or comes from anthropogenic sources, such as fuel 45 combustion, petroleum products and long range transport with atmospheric solid particles (Abakumov et al. 2014; 46 Abakumov et al. 2015). Some PAHs are of the most environmental importance because of the established 47 carcinogenic, mutagenic and teratogenic effects in living organisms and in humans particularly (Menzie et al. 1992; 48 Shaw and Connell 1994; Perera 1997; Yu 2002; Guo et al. 2013). A number of 16 PAHs have been listed as priority 49 contaminants by both the US Environment Protection Agency (US EPA) and European Union (EU). A mong them 50 seven compounds, i.e. benzo(a)anthracene, chrysene, benzo(a)pyrene, benzo(b)fluoranthene, benzo(k)fluoranthene, 51 dibenz(a,h)anthracene and indeno(1,2,3-cd)pyrene are considered as probable human carcinogens (US EPA 2002). 52 In Canada, US and some European countries normalization of soil contamination is provided upon developed soil 53 quality criteria for selected PAHs or their sum. Only a few countries have established comprehensive soil guideline





54 values (SGV) for particular land use at least for the sum 85 of priority PAHs (Σ 7; 10; 15; 16). Generally, the 55 existing soil critical values provides only human health-risk based approaches and don't consider protection of other 56 ecological receptors. In turn, US EPA has developed ecological soil screening levels (Eco-SSLs) for PAHs, which 57 are derived separately for four groups of ecological receptors: plants, soil invertebrates, birds and animals. However 58 these screening levels are intended to evaluate an unacceptable ecological risk to terrestrial receptors, they are not 59 designed to be used as cleanup levels. For this purpose US EPA adopted a human health based Preliminary 60 Remediation Goals for Soil (PRG) using estimates of different routes of exposure. In contrast to this, Russian 61 Federation has not yet developed soil guideline values at least for the sum of priority PAHs; normalization is 62 provided only for soil contamination with benzo(a)pyrene without distinction for particular land use. On top of that, 63 no threshold values are provided for other POPs. A summary of soil guideline values for PAHs set in some countries 64 is presented in supplementary materials 1. Thus studies on soil contamination with PAHs are of the most 65 importance as they provide information that can be further used to delineate special contaminated sites exhibiting 66 high risk to human exposure. Thousands of reports about PAHs concentrations, sources and health risk assessments 67 in urban and semiurban areas from all over the world were published in recent years. Elevated levels of PAHs in 68 urban soils were reported in Hustan, USA (Hwang et al. 2002), Beijing, China (Tang et al. 2005), Glasgow of UK 69 and Torino of Italy (Morillo et al. 2007), and Esbjerg of Denmark (Essumang et al. 2011).

70 St. Petersburg is the largest industrial and transport center in the northwestern region of Russia and is of 71 great interest from the viewpoint of environmental concern. The ecological status of such a large center reflects the 72 whole range of socioeconomic problems resulting in decline of human health under the influence of various 73 chemical, physical and biological factors. The ecological situation in the city is determined by the emissions from 74 more than a thousand industrial enterprises, large railway junction, seaport and the great motor vehicle fleet -1 670 75 794 cars and 207 975 trucks as of 2014 (Belousova et al. 2014). All this transport is served by a huge amount of 76 petrol stations and transport companies: currently in St. Petersburg operate 27 fuel operators and 397 petrol stations. 77 Industrial enterprises of the city include high-capacity, resource- and power-consuming ecologically dangerous 78 works. According to the data collected from the automatic air monitoring system of the city in 2014 total emission 79 into the air from both the stationary sources and vehicles has reached 513 200 t in 2014 of chemicals, including 16 80 903 t of hydrocarbons (CHx), 3000 t of black carbon (BC) and 47 900 t of volatile organic compounds (VOC) 81 (Belousova et al. 2014). The amount of emissions per capita complies 135.9 kg / year, per unit area - 434.5 t / km2 82 (Belousova et al. 2014). At the same time, 91.9% of emissions are accounted to the transport activity. Industrial and 83 transport emissions are the major source of soil contamination with PAHs in urban areas along with inputs from 84 petroleum products. No systematic survey of soil contamination with priority PAHs has been conducted yet in St. 85 Petersburg except for benzo(a)pyrene (Gorkiy and Petrova 2007). Considering this fact and environmental aspects 86 of the territory described above, St. Petersburg affords an excellent location to study geochemical cycles of PAHs.

87 Therefore this study is aimed to test the hypothesis on the PAH loading differences between urban
 88 territories with different land use scenarios. The results of this study would contribute to the knowledge about PAH
 89 distribution in urban soils of Eastern European region and may be used by decision makers during land
 90 management.

91 Objectives of the study were to: 1) explore qualitative and quantitative composition of 15 priority PAHs in 92 urban soils in some parkland, residential and industrial areas of St. Petersburg: 2) compare with existing data on the 93 PAHs distribution in urban soils; 3) distinguish between PAHs sources using PAH molecular ratios; and 4) evaluate 94 cancer risks associated with soil contamination with PAHs within selected areas.

95 2. Materials and methods

96 2.1 Study site description

97 Choice of the study area, namely Primorsky, Vasileostrovsky and Kirovsky administrative Districts of St. 98 Petersbrg was done in order of increasing of location density of potential stationary sources of contamination with 99 PAHs, population density and traffic activity. Detailed characteristics about each chosen area are given in Table 1. 100 Certain areas of recreational, residential and industrial land use scenarios within each chosen District were subjected 101 under the study. Information on the land use scenario of each chosen area was obtained using online map service 102 "Regional Geoinformational System RGIS" developed with the support of the Committee for land resources and 103 land management of St. Petersburg (Fig. 1). Potential sources of PAH contamination affecting PAH levels in soil 104 here are high traffic activity (Western highway and Primorsky prospect), steel and chemical industries (Kirovsky 105 engineering plant, Baltiysky shipyard plant, varnish factory "Kronos"), thermal-power-stations ("North-Western").





106 Climate is moderately continental, significantly affected by the Baltic Sea. The average annual amount of 107 precipitates varies 565-635 mm. Humidity factor - 1.1-1.3. The territory represents an almost flat plain with 108 altitudes below 20 m above the sea level (Neva Lowland). Natural soil formation usually occurs on ancient lake-109 marine littoral sands, sandy loams, loams (less) depleted in calcium (Gagarina et al. 2008). Urban soils are formed 110 on the bulk deposits ranging from 0.9 to 4 m of thickness (Matinyan et al. 2005; Dashko et al. 2011). Soils are 111 strongly disturbed by anthropogenic activities (buried, sealed and/or contaminated), with small relatively intact 112 islands in natural and seminatural areas on the north, north-west and north-east of the City. Soils of the historical 113 center are presented by anthropogenic soil-like formations called in national soil classification systems as 114 "urbanozems" (Stroganova e. al. 1992) or "urbiquazizems" (Shishov et al. 2006) and generally characterized by light 115 grain size, modified soil profiles, with abundant inclusions of anthropogenic artefacts in the form of debris, domestic 116 wastes and remains of communications, neutral to alkaline pH, high humus, nitrogen and phosphorus content, 117 humate and fulvic-humate types of humus and traces of chemical contamination (Rusakov et al. 2005; Matinyan et 118 al. 2005; Ufimtseva et al. 2011). Investigated urban soils were classified as Technosols according to the World 119 Reference Base for soil resources (Michéli et al. 2006).

120 2.2 Sampling strategy and procedure

121 Sampling was conducted in September 2013 at 9 urban sites, in dry and clear weather conditions according 122 to international standard protocol ISO 10381-1 (2002) and national sampling standard GOST 17.4.4.02-84 (1984). 123 Soil samples were taken from 0-20 cm topsoil layer. A total of 135 grab soil samples were collected diagonally from 124 25 m² sampling plots were combined into 27 composite samples of 0.7 kg each one. Location of the sampling sites 125 was defined according to proximity to residential areas and potential pollution sources (Fig.1).

126 Sampling strategy responds to the study objectives and is aimed to provide comprehensive characterization 127 of the selected sites suspected to be contaminated with PAHs. Quantity of grab samples to be collected depended on 128 the size of sampling sites, e.g. 15 grab samples per 0.8 ha site collected from sampling plots (S = 25 m^2). Soil depth 129 selected for sampling is a function of exposure routes (e.g. soil ingestion, dermal contact with soil and dust, 130 inhalation of contaminated dust, inhalation of volatile compounds). Sampling pattern represents both the purposive 131 and judgement sampling techniques, delineating sample locations that assumed to be representative of the whole site 132 and most contaminated. Instruments for sample derivation included stainless scoop and stainless knife prewashed 133 with acetone. The representativeness of collected samples was provided thorough mixing and taking an average 134 sample by quartering method.

135	Fig. 1	. Location	of the soil	sampling	sites.
100	rig. 1	• LOCATION	or the son	sampling	sucs.

]	Description to fig. 1	
Land use	a - Primorskij District	b - Vasileostrovskij District	c - Kirovskij District
Parkland	1 - The park of the 300th anniversary of St. Petersburg, Primorskij prospect, 157. 59°59'2" N, 30°11'33" E.	4 - AkademicheskijGarden, 2d Line of Vasilyevskij Island, 2A. 59°56'19.8" N, 30°17'18.3" E. Rumyantsevskij Garden, Rumyantsevskaya square, 7. 59°56'18.4" N, 30°17'33.1" E.	7 - The park of 9th January, Stachek prospect, 19. 59°53'31.1" N, 30°16'25.5" E.
Residential	2 - Intersection of Yahtennaya street and Optikov street. 59°59'55.7" N, 30°13'22.9" E.	5 - Korablestroiteley street, 20, 19/2. 59°56'37.3" N, 30°12'48.3" E. 59°56'38.0" N, 30°13'05.2" E.	8 - Korneev street, 4. 59°53'06.9" N, 30°16'03.8" E.
Industrial	3 - Vicinity of the Bus depot №2, Avtobusnaya street, 12A. 60°01'46.6" N, 30°15'34.7" E.	6 - Vicinity of the Baltic shipyard, Detskaya street, 3. 59°55'36.1" N, 30°15'13.1" E.	9 - Vicinity of the Kirovskij engineering plant, Stachek prospect, 47. 59°53'09.3" N, 30°15'48.1" E.

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Collected samples were packed in labeled sterile plastic bags, kept in cool condition and transported to the 137 laboratory. Once in laboratory, soil samples were dispersed on the sterile glass plates and air-dried at the room 138 temperature for 5 days, cleaned from the organic and inorganic debris, grounded in laboratory vibrating cup mill, 139 sieved through 0.25 mm caprone sieve and finally stored in the dark glass containers prewashed with acetone until 140 analysis. This technique enables to prevent cross-contamination as well as losses of PAHs due to environmental 141 factors (Berset et al. 1999).

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143 2.3 HPLC, PAH source identification and risk evaluation

14 15 PAHs were analyzed, including naphthalene (NAP), acenaphthene (ANA), fluorene (FLU),
145 phenanthrene (PHE), anthracene (ANT), fluoranthene (FLT), pyrene (PYR), benzo(a)anthracene (BaA), chrysene
146 (CHR), benzo(b)fluoranthene (BbF), benzo(k)fluoranthene (BkF), benzo(a)pyrene (BaP), dibenz(ah)anthracene
147 (DBA), benzo(g,h,i)perylene (BPE), indeno(1,2,3-cd)pyrene (IPY) (Fig. 2).

148 Fig, 2. Structures of the studied PAH compounds

149 PAHs content in samples were determined on the basis of US EPA method 8310 (1996a), national standard 150 method PND F 16.1:2:2.2:3.62-09 (2009), and Gabov (2007; 2008). Extraction of the PAHs was carried out at room 151 temperature with methylene chloride (high purity grade) and ultrasonic treatment via Branson 5510 ultrasonic bath 152 (USA, power 469 W, working frequency 42 kHz) following the US EPA method 3550b (1996b). Solvent removal 153 (evaporation) was carried out with Kuderna-Danish concentrator (Supelco). PAHs fractions were purified by 154 consecutive chromatography in columns filled with aluminum oxide (Brockman activity grade 2-3, Neva Reaktiv) 155 and silica gel (Fluka) according to the US EPA purification method 3660c (1996c). The purity was controlled by the 156 absence of peaks in the blank chromatogram. A standard mixture of 15 PAHs (Supelco) with the concentrations of 157 each component in the range of $100-2000 \ \mu g/cm^3$ was used to prepare the standard PAH solutions. Qualitative and 158 quantitative determination of PAHs in soils was carried out by reverse-phase high-performance liquid 159 chromatography (HPLC) in gradient mode with spectrofluorimetric detection via chromatograph "Lyumahrom" 160 ("Lumex", Russia). Chromatography was performed at 30°C on a column Supelcosil ™ LC-PAH n5 µm (25 cm × 161 2.1 mm). Mobile phase was provided with acetonitrile-water gradient. Samples of 10 µl volume were injected using 162 injection valve. Individual PAHs were identified by the time of retention and comparison of fluorescence spectra of 163 the components coming from the column with spectra of the standard PAHs. Quantitative analysis of PAHs was 164 performed by external standard method. For the quality assurance purposes Standard reference materials® 1944 165 New York/New Jersey Waterway Sediment (National Institute of Standards and Technologies NIST, USA) 166 containing a mixture of 15 PAHs were subjected to the procedure described above. The error of measuring the 167 PAHs (benz[a]pyrene) in the soils was 35% in the range of 5-40 ng/g and 25% in the range of 40-2000 ng/g with a 168 confidence probability of P = 0.95.

PAH molecular markers and ratios were used to determine PAH sources (Pandey et al. 1999; Yunker et al.
2002; Hwang et al. 2003). Sum of combustion PAHs (CombPAH/15PAH) was used as tracer of pyrogenic sources.
CombPAH/15PAH marker indicates portion of the sum of combustion specific compounds in total PAH content,
which are Fluoranthene, Pyrene, Chrysene, Benzo(a)anthracene, Benzo(k)fluoranthene, Benzo(b)fluoranthene,
Benzo(a)pyrene, Benzo(g,h,i)perylene and Indeno(1,2,3-cd)pyrene (Prahl and Carpenter, 1983). Applied PAH
molecular markers and ratios as well as their ranges are given in supplementary materials 2.

Since benzo[a]pyrene (BaP) is the most studied PAH, the carcinogenic potential of other PAHs is generally
assessed referring it to that of BaP ("toxicity equivalence factors" (TEFs), in similar way to the "toxic equivalents"
(TEQ) used in the evaluation of the toxicity of dioxins and furans. Benzo[a]pyrene Potency Equivalence Approach
is a major approach used by regulatory agencies such as the US EPA (1993; 1999), California EPA (OEHHA 1992),
Netherlands (Verbruggen et al. 2001), the UK (Duggan and Strehlow 1995), or Provinces of British Columbia and
Ontario for assessing the human health risks of PAH-containing mixtures.

181 2.4 Soil properties analysis and statistical treatment

182 Total organic carbon (TOC) was determined using a "Leco" CHN-628 elemental analyzer (USA, 183 combustion temperature 1030 °C, oxygen boost time 28 s). Inorganic carbonates were removed before analysis by 184 acidification in situ of the grounded samples with 1 M hydrochloric acid in order to avoid uncertainty in TOC 185 determination. Clay content was determined with laser diffractometer "Shimadzu" SALD-2201 (Japan). All 186 measurements were done in triplicate. All measurements were converted to absolutely dry sample. Data on analyzed 187 properties of the studied soils is presented in Table 2.

188 Measured TOC concentrations in studied samples ranged between 3.82 to 6.41% with a median value of 4,80%. Numerous studies suggested that soil organic matter (SOM) content plays an important role in retention of PAH in soil (Conte et al. 2001; Chung and Alexander 2002). In simple terms the higher SOM concentrations are, than the higher amount of PAHs can be absorbed (Karickhof and Brown, 1979; Wilcke, 2000). Entering the soil from the atmosphere PAHs are preferentially sorbed to aggregate surfaces (Wilcke, 1996). The close association of PAHs with SOM results in differentiation of organic contaminants pools among particle-size fractions (Guggenberger et al. 1996). A significant increase of PAH concentrations in finer fractions is shown in a number of





studies (Wilcke, 1996). Clay content in studied soils ranges between 1.87 and 8.50 %. A correlation coefficients
were calculated in present study in order to reveal relationship between levels of PAH in soil and analyzed soil
parameters. A strong positive correlation was found between sum of 15 PAH in soil and clay content (r = 0.91; n =
27; p = 0.95), however, no correlation of total PAH and TOC concentrations in soil was detected.

199 Statistical treatment of the data was carried out with STATISTICA 10.0 software. One-way ANOVA was 200 applied in order to test statistical significance of differences between obtained data. The essence of the method is 201 based on estimation of the significance of averages differences between three or more independent groups of data 202 combined by one feature (factor). The null hypothesis of the averages equality is tested during the analysis 203 suggesting the provisions on the equality or inequality of variances. In case of rejection of null hypothesis basic 204 analysis is not applicable. If the variances are equal, F-test Fisher criterion is used for evaluation of intergroup and 205 intragroup variability. If F-statistics exceeds the critical value, the null hypothesis is rejected considering inequality 206 of averages. Post-hoc-test (Fisher LSD) was used to provide detailed evaluation of averages differences between 207 analyzed groups of data. A feature of post-hoc-test is application of intra-group mean squares for the assessment of 208 any pair averages. Differences were considered to be significant at the 95% confidence level. All calculations were 209 carried out via STATISTICA 10.0 software. PAH concentrations were analyzed at least in triplicate. Calculated 210 mean concentrations were provided with standard deviations $(a \pm b)$.

211 **3.** Results and discussion

212 **3.1** PAH concentrations in studied soils

213 The levels of 15 individual PAH compounds analyzed in soils are shown in Table 3. The sum of 15 PAH 214 and the sum of 7 compounds included in the group of probable human carcinogens (B2) by the US EPA (1993) are given additionally. Total PAH concentrations in studied soils were found to range from traces to $8.06 \text{ mg} \cdot \text{kg}^{-1}$ (sum 215 216 of 15 priority PAH, hereafter referred to 15PAH). The vast majority of samples were characterized by concentrations of more than 1 mg·kg⁻¹, which is set as a guide level for total PAH content in soil by a number of 217 countries. The highest 15PAH levels were observed in soil samples collected from residential and industrial sites 218 219 reaching an average of 4.19 and 4.01 mg kg⁻¹ respectively with a maximal value of 8.06 mg kg⁻¹ for industrial site in 220 Kirovsky district (hereafter - K.D.) Concentrations found in parkland areas were substantially lower than those of 221 residential and industrial, with an average value of 1.08 mg·kg⁻¹

222 Distribution of the sum of the 7 carcinogenic PAH (7PAH) in soils of the studied urban sites is generally 223 characterized by the same pattern as the total PAH content in soils. The highest 7PAH levels were measured in soil 224 samples taken from residential sites (1.94 mg kg^{-1}) with an absolute value of 3.47 mg kg^{-1} in technosol of K.D. 225 residential area. 7PAH levels in parkland areas tend to be at lower range respectively to distribution of 15PAH. All 226 sampling sites were located in a proximity of less than 250 m to the highways (Korablestroiteley street, Stachek 227 prospect, Optikov prospect, University embankment, Bolshoi prospect V.O. and others) showing heavy traffic. The 228 portion of 7PAH to the 15PAH in all tested samples ranged between 41 % and 46 %, which evidently shows that the 229 soils may represent considerable health risk for human.

The bar chart showing the contribution of PAH with different ring numbers to the sum of PAH in soils is depicted in **Fig. 3.** The sum of organic pollutants is mostly dominated by heavy molecular weight PAH with 4-5 rings. Portion of 4-ringed PAH compounds in soil of residential and industrial sites accounts for 50% of the sum decreasing to 34% in parkland soils. 5 ringed PAH including such compounds as BaP, BbF, BkF, and DBA contribute up to 31 % of the sum of PAH insignificantly varying between studied areas. The rest portion is accounted for the 6-ringed (10-14%) and low molecular weight PAHs with 2 or 3 rings in structure (11-17%).

236 Fig 3. Distribution pattern of PAHs with different ring numbers in studied soils

237 The pie chart illustrating composition of PAH mixtures in soils is depicted in Fig. 4. The obvious equality 238 in PAH distribution patterns in all studied sites clearly indicates the common source of PAHs. Pyrene and 239 Fluoranthene (4-ring PAHs) are the most abundant compounds in examined samples, portion of which accounts for 240 16-18 % of 15PAH. The following predominant compounds are 5-ring PAH benzo(b)fluoranthene (10-11%) and 241 benzo(a)pyrene (8-11%). The rest portion of the sum is represented by lighter weight PAHs (2-3-ring PAHs) and is 242 generally dominated by Phenanthrene (6-9%). Domination of 4 and 5-ring PAHs, mainly PYR, FLT, BbF and BaP, 243 in studied soils is indicative of elevated diesel fuel consumption activity on the territory. Estimated diesel 244 consumption in St. Petersburg reaches 38% of the total fuel use for transportation (Belousova et al. 2014). As known 245 emission rate of heavyweight PAH fraction due to diesel combustion is several times higher (Sjogren et al. 1996; 246 Marr et al. 1999).





247 Fig 4. Composition of PAH mixtures in studied soils

248 Obtained data are nearly consistent with data from Lodygin et al. (2008) exploring PAH levels (sum of 11 249 PAHs) in soils of Vasilievsky Island in St. Petersburg). The main anthropogenic impact on soils of residential area 250 of the island was exerted by light polyarens, including 2-4 ring substances (as stated by the author), the portion of 251 which in the total content of PAHs was more than 50%. Maximum concentrations of PAHs were detected in soils 252 along highways with intense traffic and considerable emission of combustion gases. The reported total PAH content 253 ranged from 0.197 to 8.20 mg·kg⁻¹ between different land utilization types. The described distribution patterns of 254 individual PAHs are similar to those of this study: the most abundant are 4-5 ring PAHs, particularly Pyrene (17%), 255 Fluoranthene (17%), Benzo(g,h,i)perylene (13%), Benzo(b)fluoranthene (12%) and Benzo(a)pyrene (12%). Several 256 samples were noticed to exhibit higher contents of heavy polyarens of natural origin, as both of the samples were 257 represented by fresh organic material (peat) which is used as amendment in soils of residential areas and roadsides. 258 Thus the findings of above mentioned study suggest that spatial distribution of PAHs is mainly dictated by the 259 closeness to highways and by the artificial input of peat material in the urban soils.

260 There is still a lack of information about PAHs concentrations in soils of St. Petersburg, so the data on the 261 pollutants distribution in water sediments obtained from environmental monitoring systems may be applied in 262 discussion for evaluation of the PAH loads. Comparative PAH levels were detected in bottom sediments in different 263 parts of Neva Bay (Gulf of Finland) and along the Niva river waterway. Reported total PAHs concentrations ranged 264 between 0.01 to 14.5 mg·kg⁻¹ (HELCOM 2014). Benzo(a)pyrene was detected in 96% of sediment samples taken 265 with and average concentration of 0.09 mg·kg⁻¹.

266 Total PAH concentrations in soils of urban and industrial sites from a number of investigations set in other 267 countries are summarized in Table 4. Tang et al. (2005) reported a sum of 16 PAHs of 27.82 mg kg⁻¹ in roadside soils of Beijing, China. Hwang et al. (2002) found a total PAH concentration of 0.20-2.20 mg·kg⁻¹ in urban and 268 suburban soils in Huston, Texas, USA. Notable PAH concentrations were observed by Mielke et al. (2001) in New 269 Orleans urban soils (USA), fluctuating around a medium level of 3.73 mg·kg⁻¹. Nadal et al. (2004) reported 270 271 relatively lower 16 PAHs levels in soils of the vicinity of the chemical and petrochemical industries and 272 urban/residential sites in Tarragona County (Catalonia, Spain), ranging 284 between 0.11 and 1.0 mg·kg⁻¹. 273 Comparable findings were announced by Bucheli et al. (2004) for soils of urban and semiurban areas in Switzerland, 274 containing 0.05-0.62 mg·kg⁻¹ of the sum of 16 PAHs. In general terms, the predominance of 3-5 ring PAHs is noted, 275 which is mainly attributed with influence of the anthropogenic activities on the studied territories.

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3.2 Determination of the PAH sources and statistics

277 While a domination of high molecular weight PAH fraction indicates a combustion origin (pyrogenic), 278 enrichment of low molecular weight PAHs is common in fresh fuels (petrogenic) (Masclet et al. 1987, Budzinski et. 279 al. 1997). Special molecular markers and ratios, proposed by Yunker et al. (2002) and a total combustion PAHs 280 index, reported by Hwang et al. (2003) were applied for PAH sources apportionment. Obtained meanings of applied 281 PAH molecular ratios are listed in Table 5. Applied markers allow to distinguish between pyrogenic and petrogenic 282 as well as traffic and non-traffic sources of PAHs, namely: ANT/(ANT+PHE), FLT/(FLT+PYR), BaA/(BaA+CHR), 283 IPY/(IPY+BPE), CombPAH/15PAH and BaP/BPE. Calculated ratios for samples taken from residential and 284 industrial exhibited numbers that point to a domination of pyrogenically formed PAHs. The cross -plots of the PAH 285 ratios is depicted in Fig. 5

286 Fig. 5 PAH source apportionment

287 Several markers are indicative of certain combustion sources of PAHs, appointing to gasoline, diesel, crude 288 oil or grass, coal and wood combustion origins, namely: FLT/(FLT+PYR), BaA/(BaA+CHR), IPY/(IPY+BPE) and 289 BaP/BPE. The calculated FLT/(FLT+PYR) (0.49-0.51), IPY/(IPY+BPE) (0.30-40) and BaP/BPE (1.20-1.64) values 290 point to a domination of gasoline, diesel and oil combustion. However, obtained values of FLT/(FLT+PYR) and 291 BaA/(BaA+CHR) ratios suggested that coal and wood combustion have a certain role in PAHs origination as well. It 292 is important to note that the shift of heavy and low molecular PAHs ratio towards the heavy ones cannot be 293 explained only by anthropogenic factor, the degradation of lighter PAHs due to environmental factors such as 294 photolysis under the direct sun rays in the topsoil layers, as well as thermal degradation, biological uptake and 295 biodegradation may play a significant role as well (Behymer and Hites 1985; Wild and Jones 1995; Wang 1998; 296 Johnsen 2005; Choi et al. 2010). These processes are predetermined by physical and chemical properties of the 297 lighter fraction PAHs such as low molecular weight, high vapor pressure and high volatility rate (Mackay and 298 Hickie 2000). Volatilization was proved to play the most significant role in the global degradation of the 2- and 3-





299 ringed PAHs especially. Park et al. (1990) reported that approximately 30% loss of Naphthalene accounts for 300 volatilization, while for the remaining compounds this process was insignificant. Heavy weight PAHs, i.e. 4-6-ring 301 compounds, have low solubility in water and low volatility, strong affinity to particulates (BC and SOM, fine 302 fractions), are less accessible for biological uptake and degradation and thus are more persistent in the environment 303 (Johnsen 2005; Haritash 2009). It has been proven that PAHs may form nonextractable [¹⁴C]PAH residues in soil 304 under the stimulation of microbial activity, which obviously leads to unexpectable lower results while analyzing the 305 concentrations of Naphthalene, Anthracene, Pyrene and Benzo(a)pyrene in soil samples (Eschenbach et al. 1994; 306 Eschenbach et al. 1998).

307Obtained probabilities for One-way ANOVA revealed no statistically significant differences of total PAH308concentrations in soils between different land uses (P < 0.05). Plot of LS Means is depicted in Fig. 6.

309 Fig. 6. LS Means plot, differences of PAH levels in soil between land uses.

310 The differences in levels of individual PAH compounds were tested using Post-hoc Fisher LSD test. The 311 results showed significant differences of FLT, PYR and PHE concentrations between parkland, residential and 312 industrial areas (p = 0.01-0.03). The tested hypothesis suggested that PAH levels in urban soil may differ between 313 areas with different land utilization type, following the order: industrial, residential, parkland. Thus the results of the 314 study did not prove the tested hypothesis, suggesting the argument of equal PAHs load on the urban soils. The land 315 use factor is expressed only in distribution of the dominant individual PAHs, particularly FLT and PYR. These 316 compounds are known to be a part of the PAHs mixtures isolated from the exhaust gases and industrial emissions 317 (Fernandes et al. 1997; Rehwagen et al. 2005). So not too surprising, that elevated levels of these pollutants are 318 expected primarily in industrial and transport areas along with surroundings, where maximum input of black carbon 319 from air pollution sources is noted. PHE representing low molecular weight PAH is a thermodynamically stable tri-320 aromatic compound arising from petroleum-hydrocarbon-based releases. Distribution of this contaminant follows 321 the scheme of potential sources of contamination with petroleum products allocation (Fig 7).

Fig. 7. Scale of potential sources of contamination with petroleum products (units per square km) with PHEdistribution plots

324 **3.3** Risk evaluation of PAHs in soils

Health risk associated with soil contamination with PAHs was assessed using benzo(a)pyrene total potency
 equivalents approach (BaP_{eq}). The BaP_{eq} for a soil sample is simply calculated by multiplying the concentration of
 each PAH in the sample by its benzo(a)pyrene toxic equivalency factor (TEF), given in Table 6.

328 The calculated BaPeq on the average concentration of 15PAH (here and after referred to BaPeq-15PAH) varied between 0.44 to 0.66 mg·kg⁻¹ dry soil. The highest BaPeq-15PAH mean concentrations were found in 329 residential and industrial areas - 0.66 and 0.55 mg·kg⁻¹ respectively. Parkland areas are characterized by the lower 330 but still considerable levels of BaPeq-15PAH (mean 0.44 mg·kg⁻¹). It is to be noted that one single sample taken 331 332 from the Kirovskij parkland exhibited a total BaP_{eq} concentration of 1.84 mg·kg⁻¹ (The park of 9th January), which 333 evidently shows that parkland land uses are subjected under a high load of PAHs as well as other land uses. 334 Obtained values are several times higher than reported total PAHs carcinogenic potencies in a number of studies $(BaP_{eq} \text{ of total PAHs}): 0.02 \text{ mg} \cdot \text{kg}^{-1}$ in soils of Viseu and 0.23 mg \cdot \text{kg}^{-1} in Lisbon, Portugal (Cachada et al. 2012); Nadal et al. (2004) reported BaP_{eq} concentrations varying between 0.02 to 0.12 mg \cdot \text{kg}^{-1} in soils of Tarragona 335 336 County, Spain; 0.18 mg kg⁻¹ in soils of Beijing and 0.24 mg kg⁻¹ in Shanghai, China (Liu et al. 2010; Wang et al. 337 338 2013).

Finally, obtained BaP total potency equivalents of PAHs were compared with Soil Quality Guideline values for the direct contact with contaminated soil in respect to particular land use (CCME 2010), setting out the acceptable level of incremental lifetime cancer risk (ILCR) of 1×10^{-6} for BaP_{eq} concentration in soil above the 0.6 to 5.3 mg·kg⁻¹ (for each land use). The reported BaP_{eq} for mean total PAH concentrations were above the safe level of 0.6 mg·kg⁻¹. Exposure to these soils through direct contact probably poses a significant risk to human health from carcinogenic effects of PAHs, even in urban parklands.

345 4. Conclusions

Results of the study demonstrated that soils within urban areas with different land utilization types are characterized by common loads of PAHs generally attributed to high traffic density of the city. Considerable levels of soil contamination with PAHs were noted. The common tendency in PAH distribution patterns between investigated sites clearly indicates the common source of PAHs in urban soils. A larger portion of high molecular weight PAHs along with determined molecular ratios suggest the predominance of pyrogenic sources, mainly





351 attributed to combustion of gasoline, diesel and oil. Petrogenic sources of PAHs have a significant portion as well 352 defining the predominance of petroleum associated low molecular weight PAHs such as phenanthrene. Derived 353 concentrations of 7 carcinogenic PAHs as well as calculated BaP total potency equivalents were multiple times 354 higher than reported in a number of other studies, indicating a significant risk for human health in case of direct 355 contact. One-way ANOVA results showed significant differences in levels of pyrene, fluoranthene and 356 phenanthrene - the most abundant individual PAHs in examined sampled, between parkland, residential and 357 industrial land uses, suggesting the influence of land use factor on distribution of these pollutants. Further study with 358 an application of complex statistical methods is needed such as principal component analysis which would 359 contribute to precision of PAHs sources allocation.

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362 References

- Abakumov EV, Lodygin ED, Gabov DA, Krylenkov VA (2014) Polycyclic aromatic hydrocarbons content in Antarctica soils as exemplified by the Russian polar stations. Gigiena i sanitariia 1:31-35.
- Abakumov EV, Parnikoza IY, Lupachev AV, Lodygin ED, Gabov DN, Kunakh VA (2015) Content of polycyclic aromatic hydrocarbons in soils of Antarcti stations regions. Gigiena i saniataria 7:2025.
- ATSDR (1995) Polycyclic Aromatic Hydrocarbons. Agency for Toxic Substances and Disease Registry. http://www.atdsr.cdc.gov/toxprofiles/phs69.html. Accessed 24 August 2015.
- BBodSchV (1999) Bundes Bodenschutz- und Altlastenverordnung (Bundesbodenschutzverordnung -BBodSchV). http://www.gesetze-im-internet.de/bundesrecht/bbodschv/gesamt.pdf. Accessed 6 September
 2015.
- 372 5. Behymer TD, Hites RA (1985) Photolysis of polycyclic aromatic hydrocarbons adsorbed on simulated
 373 atmospheric particulates. Environ Sci Techol 19(10):1004-1006.
- Belousova et al (2015) Report on the environmental situation in St. Petersburg in 2014. The Committee for Nature Use, Environmental Protection, and Ecological Safety of St. Petersburg, St. Petersburg, Russia.
 <u>http://gov.spb.ru/static/writable/ckeditor/uploads/2015/06/19/doklad_2014_SWipmNU.pdf</u>. Accessed 11
 September 2015.
- Berset JD, Ejem M, Holzer R, Lischer P (1999) Comparison of different drying, extraction and detection techniques for the determination of priority polycyclic aromatic hydrocarbons in background contaminated soil samples. Anal Chim Acta 383(3):263-275.
- Bucheli TD, Blum F, Desaules A, Gustafsson Ö (2004) Polycyclic aromatic hydrocarbons, black carbon, and molecular markers in soils of Switzerland. Chemosphere 56(11):1061-1076.
- Budzinski H, Jones I, Bellocq J, Pierard C, Garrigues PH (1997) Evaluation of sediment contamination by polycyclic aromatic hydrocarbons in the Gironde estuary. Mar Chem 58(1):85-97.
- 10. Cachada A, Pato P, Rocha-Santos T, da Silva EF, Duarte AC (2012) Levels, sources and potential human
 health risks of organic pollutants in urban soils. Sci Total Environ 430:184-192.
- 11. CCME (2010) Polycyclic aromatic hydrocarbons. Canadian soil quality guidelines for protection of
 environmental and human health. Canadian Council of Ministers of the Environment. http://ceqg rcqe.ccme.ca/en/index.html. Accessed 20 September 2015.
 - Choi HG, Moon HB, Choi M, Yu J, Kim SS (2010) Mussel watch program for organic contaminants along the Korean coast, 2001–2007. Environ Monit Assess 169(1-4):473-4.
- 392 13. Chung N, Alexander M. Differences in sequestration and bioavailability of organic compounds aged in dissimilar soils. Environ Sci Technol 1998;32:855- 60.
- 14. Conte, P., Zena, A., Pilidis, G., & Piccolo, A. (2001). Increased retention of polycyclic aromatic
 hydrocarbons in soils induced by soil treatment with humic substances. Environmental pollution, 112(1),
 27-31.
- 15. Dashko RE, Aleksandrova OU, Kotyukov PV, Shidlovskaya AV (2011) Features of the engineering geological conditions of St. Petersburg. Journal of Urban development and Geotechnical Engineering
 13:25-71. http://urban-development.ru/2011/2.pdf. Accessed 12 September 2015.
- 400 16. Duggan M, Strehlow CD (1995) Contaminants in Soil: Collation of Toxicological Data and Intake Values
 401 for Humans: Benzo[a]pyrene. Department for Environment, Food and Rural Affairs and the Environment
 402 Agency, London.





403	17.	Dyke PH, Foan C, Fiedler H (2003) PCB and PAH releases from power stations and waste incineration
404	10	processes in the UK. Chemosphere 50(4):469-480.
405	18.	Eschenbach A, Kastner M, Bierl R, Schaefer G, Mahro B (1994) Evaluation of a new, effective method to
400	10	Extract polycyclic aromatic hydrocarbons from soil samples. Chemosphere 28(4)/:085-692.
407	19.	Eschenbach A, wienberg R, Manro B (1998) Fate and stability of nonextractable residues of [14C]PAH in
408	20	contaminated soils under environmental stress conditions. Environ Sci Technol 32(17):2585-2590.
409	20.	Essumang DK, Kowalski K, Sogaard EG. Levels, distribution and source characterization of polycyclic
410		aromatic hydrocarbons (PAHs) in topsous and roadside soils in Esbjerg, Denmark. Bull Environ Contam
411	01	10x1c01 2011;86;438–43.
412	21.	Fernandes MB, Sicre M-A, Boireau A, Ironczynski J (1997) Polyaromatic hydrocarbon (PAH)
415	22	Calera DN, Demosilera MA, Kandersenka DM (2007). Debaudio respective badressed and in background
414 /15	22.	Cabov DN, Beznosikov VA, Kondratenko BM (2007) Polycyclic aromatic nydrocarbons in background
415	22	Cohoy DN, Domosilou VA, Kondrotanko DM, Velovlava EV (2008) Ecomption of nelvovalia anomatic
410	23.	Cabov DN, Bezhosikov VA, Kolutateliko BN, Takovleva EV (2008) Formation of polycyclic aromatic
417	24	Involutions in normen and model taga solis. Editasian Soli Sci+ $41(11)$. 160-1160.
410	24.	Caganna El, Kastvolova OG, Schastnaya ES et al (2008) Sons of the Russian plant natural zones, a
419	25	Corkiv AV Detrova EA (2007) Contamination of Saint Detersburg with organic toxic substances. Dussian
420	25.	Consequences and the substances
421	26	COST 17.4.4.02.84 (1084) Natura protection. Soils. Matheds for sampling and propagation of soils for
422	20.	chamical bacterialogical helmintological analysis Moscow (in Pussian)
425	27	Guo W. He MC. Vang ZE. Zhang HV. Lin CV. Tian ZI (2013) The distribution sources and toxicity risks
424	27.	of polyayalia aromatic hydrogerhous and n alkanes in rivaring and actuaring agree and toxicity lisks
425		Diver wetershed Environ Earth Sci 68(7):2015 2024
420	28	Havitash AK Kaushik (D (2000) Biodegradation aspects of polycyclic aromatic hydrocarbons (DAHs): a
428	20.	review I Hazard Mater 169(1):1-15
429	29	Havelcová M. Melegy A. Rapant S. (2014) Geochemical distribution of polycyclic aromatic hydrocarbons
430	27.	in soils and sediments of El-Tabhin Found Chemosphere 95:63-74
431	30.	HELCOM (2014) BASE project 2012-2014: Preparation of biodiversity and hazardous substances
432		indicators with targets that reflect good environmental status for HELCOM (including the HELCOM
433		CORESET project) and improvement of Russian capacity to participate in operationalization of those
434		indicators.
435		http://helcom.fi/Lists/Publications/INDICATORS Russian%20capacity%20to%20participate%20in%20op
436		erationalization% 20of% 20CORESET% 20indicators.pdf. Accessed June 2016.
437	31.	Hwang HM, Wade TL, Sericano JL (2003) Concentrations and source characterization of polycyclic
438		aromatic hydrocarbons in pine needles from Korea, Mexico, and United States. Atmos Environ
439		37(16):2259-2267.
440	32.	Hwang S, Cutright TJ (2002) Biodegradability of aged pyrene and phenanthrene in a natural soil.
441		Chemosphere 47(9):891-899.
442	33.	ISO 10381-1 (2002) Soil quality. Sampling. Part 1: Guidance on the design of sampling programmes.
443	34.	Johnsen AR, Wick LY, Harms H (2005) Principles of microbial PAH-degradation in soil. Environ Pollut
444		133(1):71-84.
445	35.	Kalf DF, Crommentuijn T, van de Plassche EJ (1997) Environmental quality objectives for 10 polycyclic
446		aromatic hydrocarbons (PAHs). Ecotox Environ Safe 36(1):89-97.
447	36.	Liu S, Xia X, Yang L, Shen M, Liu R (2010) Polycyclic aromatic hydrocarbons in urban soils of different
448		land uses in Beijing, China: distribution, sources and their correlation with the city's urbanization history. J
449		Hazard Mater 177(1):1085-1092.
450	37.	Lodygin ED, Chukov SN, Beznosikov VA, Gabov DN (2008) Polycyclic aromatic hydrocarbons in soils of
451		Vasilievsky Island (St. Petersburg). Eurasian Soil Sci+ 41(12):1321-1326.
452	38.	Mackay D, Hickie B (2000) Mass balance model of source apportionment, transport 482 and fate of PAHs
453		in Lac Saint Louis, Quebec. Chemosphere 41(5):681-692.
454	39.	Marr LC, Kirchstetter TW, Harley RA, Miguel AH, Hering SV, Hammond SK (1999) Characterization of
455		polycyclic aromatic hydrocarbons in motor vehicle fuels and exhaust emissions. Environ Sci Technol
456		33(18):3091-3099.





457	40.	Masclet P, Bresson MA, Mouvier G (1987) Polycyclic aromatic hydrocarbons emitted by power stations,
458		and influence of combustion conditions. Fuel 66(4):556-562.
459	41.	Menzie CA, Potocki BB, Santodonato J (1992) Exposure to carcinogenic PAHs in the environment.
460	40	Environ Sci. Technol 26(7):1278-1284. Micháli E. Sahad D. Szazzaran O. Dant D. Nachtarzaela E (2006) World reference hass for sail resources.
401	42.	Micheli E, Schad P, Spaargaren O, Dent D, Nachtergaele F (2006) world reference base for soil resources:
402	12	2000: a framework for international classification, correlation and communication (FAO).
405	45.	DA He in urban soile: a comparison of three European cities. L'Environ Monit 2007: 0:1001-8
404	11	Nadal M. Schuhmacher M. Domingo II. (2004) Levels of PAHs in soil and vegetation samples from
466		Tarragona County Snain Environ Pollut 132(1)-1-11
467	45	Nishet IC LaGov PK (1992) Toxic equivalency factors (TFFs) for polycyclic aromatic hydrocarbons
468		(PAHs) Regul Toxicol Pharmacol 16(3):290-300
469	46.	OEHHA (Office of Environmental Health Hazard Assessment) (1992) Expedited Cancer Potency Factors
470		and Proposed Regulatory Levels for Certain Proposition 65 Carcinogens. Air Resources Board and
471		OEHHA, California EPA, Sacramento, CA.
472	47.	Pandey PK., Patel KS, Lenicek J (1999) Polycyclic aromatic hydrocarbons: need for assessment of health
473		risks in India. Study of an urban-industrial location in India. Environ Monit Assess 59(3):287-319.
474	48.	Park KS, Sims RC, Dupont RR (1990) Transformation of PAHs in soil systems. J Environ 522 Eng
475		116(3):632-6.
476	49.	Perera FP (1997) Environment and cancer: who are susceptible? Science 278(5340):1068-1073.
477	50.	PND F 16.1:2:2.2:3.62-09 (2009) Quantitative chemical analysis of soil. Methods of measurement of the
478		mass fraction of polycyclic aromatic hydrocarbons in soil, sediments, sewage sludge and industrial wastes
479		by HPLC, Moscow, Russia. 23 pp (in Russian).
480	51.	Prahl FG, Carpenter R (1983) Polycyclic aromatic hydrocarbon (PAH)-phase associations in Washington
481	50	coastal sediment. Geochim Cosmochim Ac 47(6):1013-1023.
482	52.	Rehwagen M, Muller A, Massolo L, Herbarth O, Ronco A (2005) Polycyclic aromatic hydrocarbons
485	52	associated with particles in ambient air from urban and industrial areas. Sci 10tal Environ 348:199–210.
404 185	35.	the historic center Proceedings of the scientific conference: Ecology of St. Petersburg and its surroundings
405		Publiching of the St. Deterchurg State University. St. Deterchurg Russia, np. 80-82 (in Russian)
487	54	Shaw GR Connell DW (1994) Prediction and monitoring of the carcinogenicity of polycyclic aromatic
488	51.	compounds (PACs). In: Reviews of environmental contamination and toxicology. Springer New York, pp
489		1-62.
490	55.	Shishov LL, Tonkonogov VD, Lebedeva II, Gerasimova MI (2006) Russian soil classification
491		and diagnostics. Smolensk: Oikumena, pp 342 (in Russian).
492	56.	Stroganova MN, Agarkova MG (1992) Urban Soils: Experience of Study and Systematics (by
493		Example of Soils of Southwestern Part of Moscow). Soil Sci 7:16-24.
494	57.	Tang L. Tang XY. Zhu YG. Zheng MH. Miao OL (2005) Contamination of polycyclic aromatic
495		hydrocarbons (PAHs) in urban soils in Beijing. China. Environ Int 31(6):822-828.
196	58	Tranido M (1999) Polycyclic aromatic hydrocarbons in Estonian soil: contamination and profiles
490	50.	Finition Pollut 105(1):67-74
100	50	U.S. Department of Health and Human Services (1003) Hazardous Substances Data Bank
490	39.	(USDD, online, detelose). National Torrisology, Information, Dragnom, National Library, of
499 500		(HSDD, online database). National Toxicology Information Flograni, National Library of Madiaira Pathasda MD
500	<i>c</i> 0	Medicine, Demesua, MD. (2011) (2011) (1.11) (1.11) (1.11) (1.11)
501	60.	Utimtseva MD, Tereknina NV, Abakumov EV (2011) Fiziko-knimicneskaya knarakteristika
502		urbanozemov tsentrarnogo ratona Sankt-Peterourga. Vestnik Sankt-Peterourgskogo universiteta.
503	~	/(4):85-9/ (m Kussian).
504	61.	US EPA (1993) Provisional Guidance for Quantitative Risk Assessment of PAH. National
505		Service Center for Environmental Publications (NSCEP) of the US Environmental Protection
506		Agency, Washington DC: Office of Health and Environmental Assessment.
507		http://nepis.epa.gov/Exe/ZyPURL.cgi?Dockey=30002TUA.txt. Accessed 20 April 2015.
508	62.	US EPA (1996 ^a) Method 8310: Polynuclear Aromatic Hydrocarbons. In: Test Methods for
509		Evaluating Solid Waste, Physical/Chemical Methods; Third Edition; Final Update 3-A. National





510 511 512		Service Center for Environmental Publications (NSCEP) of the US Environmental Protection Agency, Washington DC: Office of Health and Environmental Assessment. Revision 0. http://nepis.epa.gov/Exe/ZvPURL.cgi?Dockev=50000U6E.txt_Accessed 20 April 2015
512	63	US EPA (1006 ^b) Method 3550b; Ultrasonic extraction In: Test Methods for Evaluating Solid
514	05.	Waste Physical/Chemical Methods: Third Edition: Final Undate 3-A National Service Center for
514		Environmental Dublications (NSCED) of the US Environmental Drotaction A genery Washington
515		DC: Office of Health and Environmental Assessment Pavision 2
517		bttp://pepis.epa.gov/Eve/ZvPUPL . cgi2Dockey=5000016E tyt_Accessed 20 April 2015
517 E10	61	LIS EDA (1006 ^c) Mathad 2620a; Silica Cal Chamun Jr. Tast Mathada for Evaluating Solid
510	04.	US EPA (1990) Method 5050C: Sinca Ger Cleanup. In: Test Methods for Evaluating Solid Waste Physical/Chemical Methods: Third Edition: Final Undate 3-A. National Service Center for
520		Environmental Publications (NSCED) of the US Environmental Protection Agency Washington
520		DC: Office of Health and Environmental Assessment Pavision 3
522		bttp://www3.apa.gov/apawasta/bazard/tastmathods/sw8/6/pdfs/3630c.pdf Accassed 20 April
522		nup//wwws.epa.gov/epawaste/nazaru/testinetilous/swo40/puis/5050c.pui. Accessed 20 April
525	65	2015. US EDA (2002) Delvavalia Organia Matter In: US EDA official web site. US Environmental
524	05.	Distriction Access Westington DC: Office of Health and Environmental Accessment
525		Protection Agency, wasnington DC: Office of Health and Environmental Assessment.
520	"	Intp://www.s.epa.gov/un/atw//inten/polycycl.intin. Accessed 50 August 2015.
527	00.	USEPA (1999) Guidance for Conducting Health Risk Assessment of Chemical Mixtures. In: Risk
528		Assessment Forum Technical Panel Report (External Scientific Peer Review Drait). National
529		Center for Environmental Assessment (NCEA) of the US Environmental Protection Agency.
530		nttp://ctpub.epa.gov/si/si_public_tile_download.ctm?p_download_id=36583. Accessed 10
531	<i>(</i> 7	September 2015.
532	67.	veroruggen Elvij, Postnumus R, van wezel AP (2001) Ecotoxicological Serious Risk
533		Concentrations for soil, sediment and (ground) water: updated proposals for first series of
534		compounds. In: RIVM report /11/01 20. National institute of Public Health and the
535		Environment, Netherlands. http://www.poi.ni/sites/default/files/cms/publicaties//11/01020.pdf.
536	<i>c</i> 0	Accessed 10 September 2015.
537	68.	Wang R, Cadman P (1998) Soot and PAH production from spray combustion of different
538	60	hydrocarbons behind reflected shock waves. Combust Flame 112(3):359-370.
539	69.	Wang XT, Miao Y, Zhang Y, Li YC, Wu MH, Yu G (2013) Polycyclic aromatic hydrocarbons
540		(PAHs) in urban soils of the megacity Shanghai: occurrence, source apportionment and potential
541		human health risk. Sci. Total. Environ. 447:80-89.
542	70.	Wilcke, W., Zech, W., & Kobža, J. (1996). PAH-pools in soils along a PAH-deposition gradient.
543		Environmental Pollution, 92(3), 307-313.
544	71.	Wilcke W (2000) Synopsis polycyclic aromatic hydrocarbons (PAHs) in soil—a review. J Plant
545		Nutr Soil Sci. 163(3):229-248.
546	72.	Wild SR, Jones KC (1995) Polynuclear aromatic hydrocarbons in the United Kingdom
547		environment: a preliminary source inventory and budget. Environ. Pollut. 88(1):91-108.
548	73.	Yang HH, Lai SO, Hsieh LT, Hsueh HJ, Chi TW (2002) Profiles of PAH emission from steel and
549		iron industries. Chemosphere 48(10):1061-1074.
550	74.	Yu H (2002) Environmental carcinogenic polycyclic aromatic hydrocarbons: photochemistry and
551		phototoxicity. J. Environ. Sci. Heal. C. 20(2):149-183.
552	75.	Yunker MB, Macdonald RW, Vingarzan R, Mitchell RH, Goyette D, Sylvestre S (2002) PAHs in
553		the Fraser River basin: a critical appraisal of PAH ratios as indicators of PAH source and
554		composition. Org. Geochem. 33(4):489-515.
555		





556 Table 1. Description of the study area

Characteristics	Units	Primorsky District	Vasileostrovsky District	Kirovsky District
S	km ²	109.87	17.1	47.10
Population	-	534 646	211 048	334746
Industries	units	250	350	70
Number of potential contamination sources with petroleum products	units	14	7	10
Density of potential contamination sources	units per km ²	0.13	0.41	0.21
CH_x emissions from stationary sources in 2014	thousand tons	0.556	0.034	0.708
VOC emissions from stationary sources in 2014	thousand tons	0.153	0.099	0.545
BC emissions from stationary sources in 2014	thousand tons	0.237	0.037	0.174

Table 2. Physicochemical properties of the studied soils

District	Landuse	Soil type FAO	TOC	Clay
District	Land use	Soli type. IAO		%
:5	Parkland	Mollic Technosol	4.10 ± 0.01	5.83 ± 0.21
norski	Residential	Urbic Technosol	3.82 ± 0.03	7.43 ± 0.06
Pri	Industrial	Urbic Technosol	5.49 ± 0.02	8.50 ± 0.10
škij	Parkland	Mollic Technosol	5.39 ± 0.01	7.3 ± 0.20
leostrovs	Residential	Urbic Technosol	6.41 ± 0.02	1.87 ± 0.12
Vasil	Industrial	Urbic Technosol	5.28 ± 0.02	3.27 ± 0.15
ij	Parkland	Mollic Technosol	4.19 ± 0.03	7.5 ± 0.10
irovsł	Residential	Urbic Technosol	4.80 ± 0.03	3.27 ± 0.15
Ki	Industrial	Urbic Technosol	3.09 ± 0.02	7.67 ± 0.06

Table 3. PAH mean concentrations in soils of St. Petersburg $(mg \cdot kg^{-1})$.





Compound	Parkla	und (n = 9)		Reside	ntial (n = 9)		Industr	rial (n = 9)		P One-way ANOVA	
4	$Mean\pm SD$	Max	Min	$Mean \pm SD$	Max	Min	$Mean \pm SD$	Max	Min	(lpha=0.05)	
NAP	0.06 ± 0.08	0.28	0.03	0.05 ± 0.02	0.07	0.00	0.09 ± 0.07	0.21	0.00	0.42	
ANA	0.02 ± 0.06	0.18	0.00	0.00	0.01	0.00	0.00	0.03	0.00	0.41	
FLU	0.10 ± 0.06	0.23	0.05	0.17 ± 0.11	0.40	0.03	0.17 ± 0.11	0.31	0.06	0.28	
PHE	0.16 ± 0.13	0.45	0.05	0.26 ± 0.17	0.47	0.03	0.36 ± 0.22	0.65	0.07	0.10	
ANT	0.06 ± 0.11	0.37	0.01	0.04 ± 0.04	0.11	0.00	0.05 ± 0.03	0.09	0.01	0.87	
FLT	0.18 ± 0.07	0.35	0.09	0.69 ± 0.52	1.49	0.04	0.72 ± 0.48	1.50	0.11	0.02	
PYR	0.18 ± 0.08	0.35	0.09	0.74 ± 0.55	1.67	0.04	0.70 ± 0.46	1.50	0.16	0.02	
BaA	0.19 ± 0.17	0.53	0.04	0.35 ± 0.26	0.64	0.02	0.30 ± 0.20	0.67	0.07	0.32	
CHR	0.15 ± 0.14	0.44	0.01	0.31 ± 0.24	0.69	0.02	0.28 ± 0.18	0.54	0.07	0.24	
BbF	0.23 ± 0.21	0.69	0.05	0.46 ± 0.30	0.84	0.02	0.41 ± 0.30	1.00	0.10	0.21	
BkF	0.15 ± 0.17	0.56	0.02	0.19 ± 0.14	0.36	0.01	0.16 ± 0.11	0.33	0.04	0.82	
BaP	0.22 ± 0.22	0.70	0.04	0.43 ± 0.32	0.87	0.02	0.34 ± 0.23	0.73	0.07	0.30	
DBA	0.03 ± 0.06	0.18	0.00	0.02 ± 0.01	0.04	0.00	0.02 ± 0.03	0.08	0.00	0.93	
BPE	0.17 ± 0.14	0.46	0.04	0.29 ± 0.21	0.52	0.01	0.27 ± 0.20	0.69	0.06	0.39	
IPΥ	0.12 ± 0.15	0.49	0.00	0.17 ± 0.17	0.45	0.01	0.15 ± 0.13	0.38	0.00	0.76	
Σ 15PAH	2.02 ± 1.50	4.78	0.58	4.17 ± 2.91	8.10	0.33	4.02 ± 2.61	8.06	0.86	0.16	
$\overline{\Sigma}$ 7PAH ^a	1.08 ± 1.04	3.18	0.21	1.94 ± 1.36	3.47	0.10	1.66 ± 1.13	3.20	0.36	0.35	
NAP – naphthale	ne,ANA - acenapht	hene,FLU –	fluorene,PH	E - phenanthrene,	ANT – anthrae	cene,FLT –	fluoranthene,PYR	– pyrene,Baz	A – benzo(a	I)anthracene,CHR -	
chrysene,BbF-1	cenzo(b)fluoranthene	s,BkF – benz	zo(k)fluorant	hene,BaP - benzo(;)pyrene,DBA	- dibenz(ah)anthracene,BPE	 benzo(g,h, 	i)perylene,I	PY - indeno(1,2,3-cc	d) pyrene.
^a Carcinogenic P.	AHs: chrysene,benz	o(a)anthrace	ne,benzo(b)	fluoranthene, benzo	(k)fluoranther	ie,benzo(a)	pyrene,indeno(1,2,3	3-cd) pyrene	and dibenz((ah)anthracene.	

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Location	Study area	Concentrations (mg·kg ⁻¹ d.w.)	∑РАН	Reference
Huston, TX, USA	Urban / suburban	0.2-2.2	23	Hwang et al. (2002)
Mexico City, Mexico	Urban / industrial	0.20-1.10	17	Hwang et al. (2003)
Beijing, China	Urban	0.22-27.82	16	Tang et al. (2005)
New Orlean, USA	Urban	3.73 (median)	16	Mielke et al. (2001)
Tarragona County, Catalonia, Spain Swiss soil	Urban / residential / industrial	0.11-1.00	16	Nadal et al. (2004)
monitoring system (NABO), Switzerland	Urban parkland / semiurban	0.05-0.62	16	Bucheli et al. (2004)
Tallinn, Estonia	Urban	2.20±1.40	12	Trapido (1999)
Linz, Austria	Industrial	1.45 (median)	18	Weiss et al. (1994)
Tokushima, Japan	Urban	0.61	13	Yang et al. (2002)
Shanghai, China	Main urban	0.13-8.65 / 0.08-7.22	26 / 16	Wang et al. (2013)
El-Tabbin, Egypt	Urban / industrial	0.05-5.56	16	Havelcová et al. (2014)
Phoenix, Arizona, USA	Urban (highways)	0.06-10.12	20	Marusenko et al. (2010)

Table 4. Reported total concentrations of PAHs in urban soils (mg·kg⁻¹ dry weight) from a number of studies

Table 5. PAH ratios in studied soils

Ratio	Parkland	Indicated source (origin)	Residential	Indicated source (origin)	Industrial	Indicated source (origin)
ANT / (ANT + PHE)	0.19	Pyrogenic	0.09	Petrogenic	0.12	Pyrogenic
FLT / (FLT + PYR)	0.51	Grass. coal and wood combustion	0.49	Gasoline. diesel and crude oil combustion	0.50	Gasoline. diesel and crude oil combustion
BaA / (BaA + CHR)	0.58	Grass. coal and wood combustion	0.52	Grass. coal and wood combustion	0.51	Grass. coal and wood combustion
IPY / (IPY + BPE)	0.30	Liquid fossil fuel combustion	0.40	Liquid fossil fuel combustion	0.34	Liquid fossil fuel combustion
BaP/BPE	1.20	Traffic sources	1.64	Traffic sources	1.31	Traffic sources
CombPAH / ∑PAH	0.79	Combustion dominated source	0.80	Combustion dominated source	0.81	Combustion dominated source

Table 6. PAH concentrations in urban soils, expressed in BaP_{eq} , $mg \cdot kg^{-1}$





Compound		Parkland			Residential			Industrial		TEF ⁴
	$Mean \times TEF$	$Max \times TEF$	$Min \times TEF$	$Mean \times TEF$	$Max \times TEF$	$Min \times TEF$	$Mean \times TEF$	$Max \times TEF$	$\operatorname{Min}\times\operatorname{TEF}$	1
NAP	0.00006	0.00028	0.00003	0.00005	0.00007	0.00	0.0000	0.00021	0.00	0.001
ANA	0.00002	0.00018	0.00	0.00	0.0001	0.00	0.00	0.00003	0.00	0.001
FLU	0.0001	0.00023	0.00005	0.00017	0.0004	0.00003	0.00017	0.00031	0.00006	0.001
PHE	0.00016	0.00045	0.00005	0.00026	0.00047	0.00003	0.00036	0.00065	0.00007	0.001
ANT	0.0006	0.0037	0.0001	0.0004	0.0011	0.00	0.0005	0.0009	0.0001	0.01
FLT	0.00018	0.00035	0:0000	0.00069	0.00149	0.00004	0.00072	0.0015	0.00011	0.001
PYR	0.00018	0.00035	60000.0	0.00074	0.00167	0.00004	0.0007	0.0015	0.00016	0.001
BaA	0.019	0.053	0.004	0.035	0.064	0.002	0.03	0.067	0.007	0.10
CHR	0.0015	0.0044	0.0001	0.0031	0.0069	0.0002	0.0028	0.0054	0.0007	0.01
BbF	0.023	0.069	0.005	0.046	0.084	0.002	0.041	0.10	0.01	0.10
BkF	0.015	0.0560	0.002	0.019	0.036	0.001	0.016	0.033	0.004	0.10
BaP	0.22	0.7	0.04	0.43	0.87	0.02	0.34	0.73	0.07	1.00
DBA	0.15	0.90	0.00	0.10	0.20	0.00	0.10	0.40	0.00	5.00
BPE	0.0017	0.0046	0.0004	0.0029	0.0052	0.0001	0.0027	0.0069	0.0006	0.01
IPΥ	0.012	0.049	0.00	0.017	0.045	0.001	0.015	0.038	0.00	0.10
Σ15PAH	0.4435	1.84154	0.05191	0.65531	1.31631	0.02644	0.55004	1.3854	0.0928	
Σ 7PA H^{a}	0.4405	1.8314	0.0511	0.6501	1.3059	0.0262	0.5448	1.3734	0.0917	

^a Values of the Toxic equivalency factors proposed by Nisbet and Lagoy (1992).

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Fig. 1. Location of the soil sampling sites.



Fig, 2. Structures of the studied PAH compounds.







Fig 3. Distribution pattern of PAHs with different ring numbers in studied soils



Fig 4. Composition of PAH mixtures in studied soils







Fig. 5 PAH source apportionment







Fig. 6. LS Means plot, differences of PAH levels in soil between land uses



Fig. 7. Scale of potential sources of contamination with petroleum products (units per square km) with PHE distribution plots