

1       **Thermal equation of state of the main minerals of eclogite: Constraining the**  
2                   **density evolution of eclogite during delamination process in Tibet**

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14

15      **Abstract**

16      Tibet, which is characterized by collisional orogens, has undergone the process of delamination or  
17      convective removal. The lower crust and mantle lithosphere appear to have been removed through  
18      delamination during orogenic development. Numerical and analog experiments demonstrate that  
19      the metamorphic eclogitized oceanic subduction slab or lower crust may promote gravitational  
20      instability due to its increased density. The eclogitized oceanic subduction slab or crustal root is  
21      believed to be denser than the underlying mantle and tends to sink. However, the density of  
22      eclogite under high-pressure and high-temperature conditions and density differences from the

23 surrounding mantle is not precisely constrained. Here, we offer new insights into the derivation  
24 of eclogite density with a single experiment to constrain delamination in Tibet. Using *in situ*  
25 synchrotron X-ray diffraction combined with diamond anvil cell, experiments focused on minerals  
26 (garnet, omphacite, and epidote) of eclogite are conducted under simultaneous high-pressure and  
27 high-temperature conditions, which avoids systematic errors. Fitting the pressure-temperature-  
28 volume data with the third-order Birch-Murnaghan equation of state, the thermal equation of state  
29 (EoS) parameters, including the bulk modulus ( $K_{T0}$ ), its pressure derivative ( $K'_{T0}$ ), and the thermal  
30 expansion coefficient ( $\alpha_0$ ), are derived. The densities of rock-forming minerals and eclogite are  
31 modeled along with the geotherms of two types of delamination. The delamination processes of  
32 subduction slab breakoff and the removal of the eclogitized lower crust in Tibet are discussed. The  
33 Tibetan eclogite which containing 40-60 vol. % garnet and 44-70% degrees of eclogitization can  
34 promote the delamination of slab break-off in Tibet. Our results indicate that eclogite is a major  
35 controlling factor in the initiation of delamination. A high abundance of garnet, a high Fe-content,  
36 and a high degree of eclogitization are more conducive to instigating the delamination.

37 **Keywords:**

38 Eclogite, Equation of state, Single-crystal X-ray diffraction, Delamination, Tibet

39

40 **1. Introduction**

41 The evolution of orogenesis is characterized by lithospheric removal during rapid surface uplift,  
42 mantle upwelling, and postcollisional magmatism, particularly in the Central Andes (e.g. Ehlers  
43 and Poulsen, 2009), Himalayas (e.g. Singh and Kumar, 2009), and Dabie orogen (e.g. He et al.,  
44 2011).

45 It is widely accepted that delamination is the most important mechanism of lithospheric  
46 removal. Delamination is induced and accompanied by two major requisites: (a) the density  
47 difference caused by the negative buoyancy of the delaminated lithosphere; and (b) the presence  
48 of a weak lower crust (lower viscosity) that exists between the strong upper crust and lithospheric  
49 mantle. Usually, two types of delamination are believed to occur in orogen development. The first  
50 is the conventional definition of delamination proposed by Bird (1978, 1979), which was used to  
51 interpret the geodynamic evolution of the Colorado Plateau. In this scenario, mantle lithosphere  
52 peels back from the overlying upper crust and is removed entirely, with the rising hot mantle  
53 filling the lithospheric removal zone (e.g. Göğüş and Ueda, 2018; Krystopowicz and Currie, 2013).  
54 A weak decoupling layer, i.e, the lower crust, is an essential condition in this delamination model,  
55 which may be affected by the rheological behavior of the hydration, thermal, and chemical  
56 characteristics of the lithosphere (e.g. Morency, 2004). In addition to conventional delamination,  
57 an alternative delamination mechanism is convective removal based on the Rayleigh-Taylor-type  
58 instability model (Houseman et al., 1981), namely, viscous “dripping”. This model postulates that  
59 there is sufficient perturbation in the lithospheric mantle, which is ascribed to the strong  
60 temperature-dependence of typical mantle rheology, without regard to a specific weak layer (e.g.  
61 Conrad and Molnar, 1999; Houseman and McKenzie, 1982).

62 All previous studies attribute the gravitational instability process of lithospheric removal to the  
63 negative thermal buoyancy of the cold lithosphere (Conrad and Molnar, 1999; Houseman and  
64 McKenzie, 1982) or density contrast between asthenosphere and mantle lithosphere. In any case,  
65 the density distribution with lithosphere pressure and temperature ( $P$ - $T$ ) conditions and chemical  
66 composition is of vital importance to understanding the process of lithospheric removal.

67 The Tibetan Plateau is the most representative and prominent collisional orogens. Two types of  
68 delamination are proposed to proceed (e.g. Chung et al., 2005; Houseman et al., 1981):  
69 lithospheric mantle removal and thickened eclogitized crust removal. The Neo-Tethyan oceanic  
70 subduction, India-Asia collision, and Indian continental subduction could be further considered  
71 responsible for the abnormal thinning of the mantle lithosphere under Tibet (Chung et al., 2005; Li  
72 et al., 2019). The lithospheric removal event in Tibet corresponds to Neo-Tethyan oceanic slab  
73 break-off. The mechanism is primarily based on density contrasts between the denser mantle  
74 lithosphere and the lighter underlying mantle. Some models reveal that lithospheric removal is  
75 induced by the retreating high-density eclogitized lithosphere detached from overlying low-  
76 density crust (Faccenda et al., 2009; Li et al., 2016). Other alternative models indicate that  
77 thickened eclogitized crust is a potential factor deriving lithospheric removal because the  
78 eclogitized crustal root is denser than the underlying mantle and tends to sink (Krystopowicz and  
79 Currie, 2013). Regardless of the above types of delamination, the density of eclogite is closely  
80 related to delamination. Therefore, Tibet provides an excellent opportunity to understand the role  
81 of eclogite density in the process of delamination.

82 An immense amount of concrete research has focused on the origin and appearance of  
83 lithospheric mantle removal from different angles, such as geophysical (Ren and Shen, 2008),  
84 geological (Chung et al., 2005), petrological (Chung et al., 2005; Turner et al., 1993), numerical  
85 and analog experiments (Göğüş and Pysklywec, 2008; Morency, 2004). In particular, numerical  
86 and analog experiments are used as prominent methods to simulate the dynamics of delamination  
87 (Göğüş and Ueda, 2018). Of these studies, the density behavior occurring during the delamination  
88 process has also been investigated intensively following thermodynamic (Semprich et al., 2010),

89 seismic/tomography (Matchette-Downes et al., 2019), and numerical simulations (Li et al., 2016).  
90 However, few studies have systematically illuminated the issue of delamination from the  
91 perspective of eclogite density. Here, we attempt to offer new insights into the derivation of rock  
92 density through the mineral physics method to constrain delamination in Tibet (Ye et al., 2021).  
93 Conducting a single experiment under high-pressure and high-temperature conditions, we obtain  
94 the equation of state (EoS) of the main minerals of eclogite with fewer systematic errors in the  
95 experiment. Furthermore, the newly derived EoS of the main minerals of eclogite, combined with  
96 the published EoSs of the main minerals of peridotite (Ye et al., 2021), geothermal lines, and  
97 collected eclogite mineral compositions, are further used to elucidate a density evolution model  
98 during the delamination process in Tibet. We argue that the EoSs of minerals could be used in a  
99 straightforward manner as new constraints on the construction of the density model. Using a  
100 simplistic calculation setup, in this study, this density evolution model will shed light on the  
101 possibility of delamination during the orogen process.

102

## 103 **2. Geological background**

104 The Tibetan Plateau is composed of four terranes from south to north: the Himalaya, Lhasa,  
105 Qiangtang, and Songpan-Ganzi terranes (Fig. 1). The birth of the Himalayas and Tibetan Plateau  
106 is a consequence of the Indo-Asian collision, which began in the early Cenozoic (Hodges et al.,  
107 2001; Wang et al., 2008). The Neo-Tethyan oceanic slab is proposed to have detached from the  
108 Indian lithosphere, and the onset of the Indo-Asian collision (DeCelles et al., 2002) particularly  
109 occurred in the lower part of the Indian and Lhasa lithospheres. The tectonic evolutionary history  
110 of the Lhasa terrane and Tethys Himalayas is essential for revealing the origin of the Himalayan-

111 Tibetan orogen. The subducting Neo-Tethyan slab was thrust into southern Tibet approximately  
112 70-65 Ma (Fig. 1b). With the closure of the Neo Tethyan Ocean, the India-Asia continent collision  
113 caused compressional deformation in southern Tibet, and a series of collision breakoff events were  
114 delineated spanning from 65 Ma to 42 Ma (Chung et al., 2005, 2009; Ma et al., 2014; Zhu et al.,  
115 2015). During this period, the Indian continental lithosphere might have dragged down to deeper  
116 depths during subduction. Meanwhile, slab rollback accompanied by the southward migration of  
117 asthenospheric convection in Tibet changed the thermal structure of the mantle wedge. The  
118 breakoff of the oceanic Neo-Tethyan slab from the more buoyant Indian continental lithosphere  
119 indicated by the eruption of early Eocene Linzizong volcanic rocks in the Gangdese arc or the  
120 cessation of Gangdese arc magmatism occurred at ~45 Ma (DeCelles et al., 2002), which opened a  
121 channel for the upwelling asthenosphere (Chung et al., 2009; Ma et al., 2014; Zhu et al., 2015).  
122 Additionally, geophysical evidence of longitudinal wave ( $V_p$ ) tomography is interpreted for the  
123 north-dipping high-speed anomaly, which is ascribed to the deep Indian mantle lithosphere (Li et  
124 al., 2008). Subsequently, the subduction of the Indian continental margin continues at a low  
125 subduction angle beneath the Lhasa terrane (Guillot et al., 2008).

126 In addition, 25 Ma to 0 Ma is another period considered to contain either the occurrence of slab  
127 breakoff (Miller et al., 1999) or lithospheric mantle removal following slab breakoff (Chung et al.,  
128 2005; Nomade et al., 2004). Previous studies suggested that the hotter asthenosphere considerably  
129 raised the geothermal conditions during this period (Chung et al., 2005). Magmatism of the  
130 ultrapotassic, shoshonitic, and calc-alkaline was widespread, which was potentially due to the  
131 partial melts of the metasomatized lithospheric mantle and eclogitized lower crust. An adopted  
132 model of convective lithospheric removal below Lhasa is widely followed (Miller et al., 1999).

133 The lithospheric removal-related mantle upwelling process has been supported by geological,  
134 geophysical, and petrological studies (Chung et al., 2005; Ren and Shen, 2008; Turner et al., 1993).

135 Here, slab breakoff and convective lithospheric removal under Tibet are adopted as the  
136 background in this study to discuss the possibility of the delamination process.

137

### 138 **3. Materials and methods**

#### 139 **3.1 Starting material**

140 Natural garnet, omphacite, and epidote samples are collected from eclogite in the Dabie-Sulu  
141 [ultra-high pressure metamorphic](#) (UHPM) belt. The compositions of each mineral are determined  
142 to be  $\text{Prp}_{21}\text{Alm}_{47}\text{Grs}_{31}\text{Sps}_1$  (Prp = pyrope, Alm = almandine, Grs = grossular, and Sps =  
143 spessartine) for garnet,  $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$  (Quad = Ferrosilite + enstatite + wollastonite, Jd = jadeite,  
144 Ae = aegirine) for omphacite, and  $\text{Ca}_{2.02}\text{Fe}_{0.75}\text{Al}_{2.32}\text{Si}_{10.16}[\text{SiO}_4][\text{Si}_2\text{O}_7]\text{O}(\text{OH})$  for epidote. The  
145 compositions of garnet and omphacite are shown in Figure 2 and are within the range of natural  
146 mineral compositions of eclogite from Tibet. The chemical composition of representative epidote  
147 minerals in Tibet shows that the Fe content of epidote exposed in eclogite is in the range of 0.13-  
148 0.25 ( $X_{\text{Fe}} = \text{Fe}^{3+}/(\text{Fe}^{3+} + \text{Al}^{3+})$ ) (Huang et al., 2015; Li et al., 2017; Liu et al., 2016), while the Fe  
149 content of epidote in this study is 0.24, which is within the Fe content range of natural epidote.  
150 The samples used in this study are representative of garnet, omphacite, and epidote minerals in  
151 natural eclogites from Tibet. The garnet, omphacite, and epidote with high-quality grains are  
152 separated from the eclogite specimens. The above three samples are crushed into  $30 \times 40 \mu\text{m}^2$  chips  
153 with a single crystal thickness of  $15 \mu\text{m}$  in our experiment.

#### 154 **3.2 Synchrotron X-ray diffraction**

155 The high-pressure and high-temperature experiment is conducted by a BX90 externally-heated  
156 diamond anvil cell (EHDAC) with  $\pm 15^\circ$  opening angles. The above three single crystals are  
157 loaded into the BX90 EHDAC equipped with a pair of 500  $\mu\text{m}$  culet-size diamond anvils and  
158 tungsten carbide (WC) seats (Figure S1). The rhenium (Re) gasket is pre-indented to a thickness  
159 of  $\sim 60 \mu\text{m}$ , and a cylindrical hole with a diameter of 360  $\mu\text{m}$  is drilled as a sample chamber. Gold  
160 powder is also loaded as the pressure calibrant (Fei et al., 2007), and neon is loaded as the  
161 pressure transmitting medium through the GeoSoilEnviroCARS (GSECARS) gas loading system  
162 (Rivers et al., 2008). The quasi-hydrostatic condition in the sample chamber can be maintained up  
163 to  $\sim 20$  GPa using the neon pressure transmitting medium (Finkelstein et al., 2017). On the other  
164 hand, high temperature can significantly decrease the deviatoric stress conditions in the sample  
165 chamber. Moreover, previous studies demonstrate that the deviatoric stress disappears at the  
166 temperatures of 650 K with neon as the pressure transmitting medium (Klotz et al., 2009; Meng et  
167 al., 1993). Therefore, the hydrostatic/quasi-hydrostatic conditions can be maintained within the  $P$ -  
168  $T$  range of our experiment ( $\sim 700$  K, 25 GPa). An automated pressure-driven membrane system is  
169 utilized to generate increasing pressure up to 25.6 GPa. High-temperature conditions up to 700 K  
170 are provided by the heating resistor. Before collecting data, the temperature in the sample chamber  
171 will be stabilized for 5 minutes and the temperature fluctuation is less than 1 K. Setup details for  
172 the employed thermocouples and heaters can be found in our previous articles (Xu et al., 2019,  
173 2020b; Ye et al., 2021).

174 *In-situ* synchrotron single-crystal X-ray diffraction (XRD) experiments were performed at  
175 experimental station 13-BM-C of the Advanced Photon Source, Argonne National Laboratory. The  
176 detailed experimental process and associated parameters can be seen in our previous studies (Xu et



177 al., 2017, 2018, 2020a; Zhang et al., 2017a). The diffraction images and the lattice parameters  
 178 were analyzed by the Bruker APEX3 software package (Dera et al., 2013). The representative  
 179 single-crystal X-ray diffraction patterns are shown in Figure S2. The specific unit-cell parameters  
 180 of the above three samples at each  $P$ - $T$  condition can be found in Table S1.

181

## 182 **4 Results and discussions**

### 183 **4.1 EoS of main minerals for eclogite**

184 The pressure-volume-temperature ( $P$ - $V$ - $T$ ) data in this study are fitted by the third-order Birch-  
 185 Murnaghan-EoS (BM3-EoS) (Birch, 1947) in combination with the Holland-Powell thermal-  
 186 pressure EoS (Holland and Powell, 2011) to obtain the thermal EoS parameters. The volume is  
 187 calculated in  $P$ - $T$  space starting with an isothermal compression and followed by a path along an  
 188 isochor curve to the final temperature. The pressure at a given volume and temperature consists of  
 189 the following two parts:

$$190 \quad P(V, T) = P(V, T_0) + P_{th}(V, T) \quad (1)$$

191 The first term corresponds to the pressure calculated by the BM3-EoS for compression at room  
 192 temperature ( $T_0$ ). The zero-pressure volume ( $V_{T_0}$ ), the isothermal bulk modulus ( $K_{T_0}$ ), and its  
 193 pressure derivate ( $K_{T_0}'$ ) with the following form:

$$194 \quad P(V, T_0) = (3/2)K_{T_0} \left[ (V_{T_0}/V)^{7/3} - (V_{T_0}/V)^{5/3} \right] \times \left\{ 1 + (3/4)(K_{T_0}' - 4) \left[ (V_{T_0}/V)^{2/3} - 1 \right] \right\}$$

195 (2)

196 The second term is the additional pressure generated by heating along an isochor. The thermal-  
 197 pressure is following the equation:

$$198 \quad P_{th}(V, T) = \alpha_{v,0} K_{T_0} \left( \frac{\theta_E}{\xi_0} \right) \left( \frac{1}{\exp(\theta_E/T) - 1} - \frac{1}{\exp(\theta_E/T_0) - 1} \right) \quad (3)$$

199 where  $\xi = \frac{(\theta_E / T)^2 \exp(\theta_E / T)}{((\theta_E / T) - 1)^2}$ ,  $\xi_0$  is the value of  $\xi$  at the reference temperature  $T_0$  and  $\alpha_{V_0}$  is

200 the thermal expansion coefficient at room temperature. The Einstein temperature  $\theta_E$  in this study  
201 are selected and recalculated from the literature (Faccincani et al., 2021; Gottschalk, 2004).

202 The thermal EoS parameters are derived using the EoSFit program at high-pressure and room-  
203 temperature and high-pressure and high-temperature conditions (Angel et al., 2014) and are shown  
204 in Table S2. Under ambient pressure and temperature conditions, the measured  $V_0$  values of garnet,  
205 omphacite, and epidote are 1566.05(25) Å<sup>3</sup>,  $V_0 = 423.48(24)$  Å<sup>3</sup>, and  $V_0 = 461.57(23)$  Å<sup>3</sup>,  
206 respectively. The fitting parameters under high-pressure and room-temperature yield  $K_{T_0} = 170$  (1)  
207 GPa,  $K_{T_0}' = 3.74$  (22) for garnet,  $K_{T_0} = 121$  (2) GPa,  $K_{T_0}' = 3.90$  (35) for omphacite, and  $K_{T_0} = 122$   
208 (1) GPa,  $K_{T_0}' = 2.51$  (16) for epidote, respectively.

209 To evaluate the quality of BM3-EoS fitting in this study, the relationship between the Eulerian  
210 strain ( $f_E = \left[ (V_0 / V)^{2/3} - 1 \right]$ ) and the normalized pressure ( $F_E = P / \left[ 3f_E (2f_E + 1)^{5/2} \right]$ ) of  
211 the main minerals for eclogite is plotted in Figure S3. Linear fitting of the three sets of data  
212 exhibited a negative slope, indicating that the pressure derivative of the bulk modulus ( $K_{T_0}'$ ) is less  
213 than 4, which is consistent with our BM3-EoS fittings. The intercept value was obtained by  
214 weighted linear regression of the data points, showing that  $F_E(0) = 171$  (2) GPa for garnet,  
215  $F_E(0) = 123$  (2) for omphacite, and  $F_E(0) = 122$  (1) for epidote, respectively. The results are  
216 consistent with the fitted isothermal bulk modulus ( $K_{T_0} = 170$  (1) GPa for garnet,  $K_{T_0} = 121$  (2)  
217 GPa for omphacite, and  $K_{T_0} = 122$  (1) GPa for epidote, respectively) within the error range.  
218 Accordingly, the  $K_{T_0}$  and  $K_{T_0}'$  obtained by the BM3-EoS fitting are reasonable. Using the  $V_0$  fixed  
219 at ambient conditions to fit third-order Birch-Murnaghan and Holland-Powell thermal-pressure  
220 EoS (BM3-HP-EoS), the available EoS parameters,  $K_{T_0} = 170$  (1) GPa,  $K_{T_0}' = 3.82$  (14), and  $\alpha_0 =$

221  $2.71 (5) \times 10^{-5} \text{ K}^{-1}$  for garnet;  $K_{T0} = 121 (3) \text{ GPa}$ ,  $K_{T0}' = 3.97 (34)$ , and  $\alpha_0 = 3.73 (20) \times 10^{-5} \text{ K}^{-1}$  for  
222 omphacite; and  $K_{T0} = 124 (2) \text{ GPa}$ ,  $K_{T0}' = 2.04 (15)$ , and  $\alpha_0 = 3.04 (13) \times 10^{-5} \text{ K}^{-1}$  for epidote are  
223 derived. The  $P$ - $V$ - $T$  data fitted through the BM3-HP-EoS model are shown in Figure 3.

## 224 **4.2 Comparison with previous studies**

### 225 **4.2.1 Garnet**

226 The thermal EoS parameters of garnet are obtained by fitting the  $P$ - $V$ - $T$  data to the BM3-EoS. We  
227 compare our results with those of previous studies (Arimoto et al., 2015; Gréaux and Yamada,  
228 2014; Lu et al., 2013; Milani et al., 2015, 2017; Xu et al., 2019; Zou et al., 2012). The  $K_{T0}$  of end-  
229 member garnet, pyrope, almandine, grossular, and spessartine crystals is between 158 and 179  
230 GPa, and the bulk modulus of almandine is the largest among the above (Table S3). From Table  
231 S3, it can be seen that bulk modulus of powder XRD (Arimoto et al., 2015; Gréaux and Yamada,  
232 2014; Pavese et al., 2001; Zou et al., 2012) are larger than those of single-crystal XRD (Milani et  
233 al., 2015, 2017) with the same composition. The  $K_{T0}$  of solid solution garnets (Beyer et al., 2021;  
234 Jiang et al., 2004; Lu et al., 2013; Xu et al., 2019) is also between 158 and 179 GPa mentioned  
235 above and will be affected by the end-member components. The  $K_{T0}=170 (1) \text{ GPa}$  in this study is  
236 reasonable within this range. The obtained  $K_{T0}'=3.82 (14)$  in this study is slightly lower than that  
237 in previous studies. The Eulerian strain and the normalized pressure of the garnet shown in Figure  
238 S3(a) exhibit a negative slope, which indicating  $K_{T0}'$  is less than 4. Moreover, compared with the  
239 previous results, the obtained value of  $K_{T0}'$  in this study is within the error range (Supporting  
240 Information Text S1). However, there is no obvious correlation between the fitted  $K_{T0}$  and  $K_{T0}'$  for  
241 minerals of different compositions (Fig. S4); hence, the  $K_{T0}$  may not be precise when  $K_{T0}'$  is fixed.  
242 For the  $\alpha_0$ , the andradite has the largest value ( $3.16 (2) \times 10^{-5} \text{ K}^{-1}$ ), and the grossular has the

243 smallest value  $(2.09 (2) \times 10^{-5} \text{ K}^{-1})$  among the end-member garnets. The thermal expansion  
244 coefficient of  $\text{Prp}_{21}\text{Alm}_{47}\text{Grs}_{31}\text{Sp}_{S1}$   $(2.71 (5) \times 10^{-5} \text{ K}^{-1})$  in this study is comparable with previous  
245 studies, but the influence of composition still needs to be considered.

#### 246 **4.2.2 Omphacite**

247 Many studies have focused on the thermoelastic properties of omphacite (Hao et al., 2019;  
248 Nishihara et al., 2003; Pandolfo et al., 2012b, 2012a; Xu et al., 2019; Zhang et al., 2016) (Table  
249 S3). Most of the results are obtained by the single-crystal XRD method, except for the result of  
250 Nishihara et al. (2003), which was obtained from powder XRD.  $K_{T0}'$  shows a higher value of 6.9  
251 (12) in the study of Nishihara et al. (2003), while in others,  $K_{T0}'$  is between 4 and 5.7, and the  
252 result of  $K_{T0}'$  (3.97) in this study is slightly lower than the above values. Additionally, according to  
253 the results shown in Table S3, the bulk moduli of omphacite are in the range of 115-123 GPa. In  
254 the study of Xu et al. (2019), an increase in the iron content would decrease  $K_{T0}$ , and they also  
255 discussed the reasons for the discrepancy in  $K_{T0}$  in detail, such as the effective ionic radius,  
256 pressure transmitting medium, and experimental pressure range. Comparing our results with Xu et  
257 al. (2019), we conclude that the incorporation of Fe would reduce the bulk modulus. However,  
258 except for Fe content, there does not seem to be a significant correlation between the other  
259 components and the bulk modulus of omphacite. The  $\alpha_0$  of the Di-Jd solid solution is similar  $(2.64$   
260  $(2) \times 10^{-5} \text{ K}^{-1}$ - $2.8 (3) \times 10^{-5} \text{ K}^{-1})$  but less than that of  $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$   $(3.73 (20) \times 10^{-5} \text{ K}^{-1})$  and  
261  $\text{Quad}_{53}\text{Jd}_{27}\text{Ae}_{20}$   $(3.4 (4) \times 10^{-5} \text{ K}^{-1})$ . It may be inferred that the Ae contents affect thermal expansion.

#### 262 **4.2.3 Epidote**

263 The thermal EoS parameters of epidote in this study are compared with those reported in previous  
264 studies (Fan et al., 2014; Gatta et al., 2011; Holland et al., 1996; Li et al., 2020; Qin et al., 2016)

265 (Table S3). Although the bulk modulus appears to be related to the  $\text{Fe}^{3+}$  content, it does not show a  
266 good correlation. Increasing the content of  $\text{Fe}^{3+}$  can enhance the bulk modulus, but the result in  
267 Holland et al. (1996) shows an abnormally large value of 162 (4) GPa, which is much higher than  
268 the 111-133 GPa resulting from other studies. This may be attributed to the fixed  $K_{T0'}$  at 4 and  
269 powder XRD methods used in the study of Holland et al. (1996). Furthermore, the  $K_{T0'}$  obtained  
270 from powder XRD (Fan et al., 2014; Gatta et al., 2011) is also larger than that from single-crystal  
271 XRD (Qin et al., 2016). The possible reasons for these discrepancies are complicated. Li et al.  
272 (2020) conducted a detailed study on this topic. Previous studies on  $\alpha_0$  and  $(\partial K_T/\partial T)_P$  of epidote  
273 are limited. **The  $\alpha_0$  ( $3.04 (13) \times 10^{-5} \text{ K}^{-1}$ ) in this study is lower than that of Gatta et al. (2011) ( $5.1$   
274  $(2) \times 10^{-5} \text{ K}^{-1}$ ) and Li et al. (2020) ( $3.8 (5) \times 10^{-5} \text{ K}^{-1}$ ).**

275

## 276 **5 Implications**

277 In the Himalayan-Tibetan system, lithospheric removal is proposed to occur in either the breakoff  
278 of the subducted slab of the Indian continental lithosphere (Chung et al., 2005; Turner et al., 1993)  
279 or convective removal of the thickened lower part of the lithosphere (Miller et al., 1999). The  
280 metamorphic eclogitization taking place in the subducted slab and the lowermost crust has been  
281 deduced as the possible cause of subducted slab break-off and the convective removal of the lower  
282 crust (Krystopowicz and Currie, 2013). Increased density in the eclogitized subducted slab and the  
283 lower crust will promote the above two lithospheric removal modes if the lower crust is weak  
284 enough for the negative buoyancy of the mantle lithosphere to be detached. Therefore, to better  
285 consider the role of eclogite density variations in the process of lithospheric removal, we model  
286 the density of minerals and eclogite aggregates along with the geotherms of Tibet and discuss the

287 effects of the degree of eclogitization on lithospheric removal.

288 The eclogite chemical data collected in Tibet and examined in our study come from a great  
289 number of eclogite samples collected in previous studies (e.g. Chan et al., 2009; Liu et al., 2019;  
290 Yang et al., 2009; Zhai et al., 2011a). The eclogite samples consist of garnet, omphacite, epidote,  
291 amphibole, zoisite, symplectite along with minor phengite, quartz, rutile, and rare apatite, ilmenite,  
292 and titanite as accessory minerals. Since the eclogite samples have suffered retrograde  
293 metamorphism, we assume that is largely composed of garnet and omphacite plus slight epidote  
294 before retrograde metamorphism. The accessory phases observed in natural eclogite are excluded  
295 because of their minimal abundance of less than 5%. Based on the mineral composition data of  
296 exposed eclogite in Tibet (Fig. S6) (e.g. Cheng et al., 2015; Dong et al., 2018; Huang et al., 2015;  
297 Jin et al., 2019; Li et al., 2017; Zhai et al., 2011b, 2011a), the components of eclogite are 50  
298 vol. % garnet + 45 vol. % omphacite + 5 vol. % epidote (parameterized as a value out of 100)  
299 using the normal distribution.

300 We take into account two different delamination modes, namely, delamination caused by the  
301 separation of the Neo-Tethyan slab (detachment of the subducted Neo-Tethyan oceanic slab) in the  
302 Paleozoic and convective removal of the lower crust of the subducted Indian continent beneath the  
303 Lhasa terrane during the Cenozoic. The temperature and pressure conditions of exposed eclogites  
304 in the Paleozoic and Cenozoic are somewhat consistent with the geothermal lines provided by  
305 previous studies (Fig. S7). The two different delamination modes reflect relatively cold geotherms  
306 and hot geotherms, respectively. Therefore, these geothermal lines are used in our models. The  
307 thermal EoS parameters of eclogitic garnet, omphacite, and epidote are derived through the **BM3-**  
308 **HP-EoS** shown in supporting information Table S2.

## 309 **5.1 The density of main minerals for eclogite along the geothermal profile in Tibet**

310 Tibetan eclogite is mainly composed of garnet, and omphacite, with a few epidotes. As shown in  
311 Figure 2, the exposed minerals differ in composition. The specific composition of minerals  
312 constrains the density. Therefore, we refer to the thermoelastic parameters of Xu et al. (2019) and  
313 Nishihara et al. (2003) to depict the density distribution of different components (Fe content) of  
314 garnet and omphacite under Tibetan geothermal lines, respectively. The corresponding  
315 thermoelastic parameters can be seen in Table S3. The mineral compositions of previous studies  
316 are within the range of the Tibetan constituents collected in this study (Fig. 2).

317 The density distribution of minerals along with relatively cold Tibetan geothermal conditions is  
318 shown in Figure 4 (the results along with hot geotherms can be seen in supporting information Fig.  
319 S8). The result clearly shows that the density of garnet is linked with the iron content. The density  
320 of garnet ( $\text{Prp}_{21}\text{Alm}_{47}\text{Grs}_{31}\text{Sps}_1$ , with 47 mol. % almandine) in this study is higher than that of  
321 low-Fe garnet ( $\text{Prp}_{28}\text{Alm}_{38}\text{Grs}_{33}\text{Sps}_1$ , with 38 mol. % almandine) (Xu et al., 2019) by 2.22% but  
322 lower than that of high-Fe garnet ( $\text{Prp}_{14}\text{Alm}_{62}\text{Grs}_{19}\text{Adr}_3\text{Sps}_2$ , with 62 mol. % almandine) (Xu et al.,  
323 2019) by 3.82% at ~80 km (Fig. 4a). With increasing depth, the density of high-Fe garnet  
324 increases by a larger amplitude. This discrepancy may be caused by its smaller degree of thermal  
325 expansion ( $2.56(44) \times 10^{-5} \text{ K}^{-1}$ ). Accordingly, the influence of pressure on the density is greater  
326 than that of temperature, which leads to faster increases in density with depth. The density of  
327 omphacite does not show obvious characteristics related to its composition. The density of  
328 omphacite ( $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$ , with 7 mol. % aegirine) in this study is lower than that of high-Fe  
329 omphacite ( $\text{Quad}_{53}\text{Jd}_{27}\text{Ae}_{20}$ , with 20 mol. % aegirine) (Xu et al., 2019),  $\text{Quad}_{72}\text{Jd}_{28}$  (Nishihara et  
330 al., 2003), and  $\text{Quad}_{57}\text{Jd}_{42}\text{Ae}_1$  (with 1 mol. % aegirine) (Xu et al., 2019) by 1.95%, 1.47%, and

331 0.83%, respectively, at ~80 km (Fig. 4b). The presence of iron in certain quantities does increase  
332 the density of omphacite, but the density of omphacite is also affected by other elements, such as  
333 calcium and magnesium. Moreover, thermal EoS parameters are also of vital importance to  
334 calculate the density. The relatively low thermal expansion of  $\text{Quad}_{72}\text{Jd}_{28}$  ( $2.7(3)\times 10^{-5} \text{ K}^{-1}$ ) and  
335  $\text{Quad}_{57}\text{Jd}_{42}\text{Ae}_1$  (with 1 mol. % aegirine) ( $2.8(3)\times 10^{-5} \text{ K}^{-1}$ ) may further enhance the increasing rate  
336 of density with depth. It is worth noting that the densities of  $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$  (with 7 mol. %  
337 aegirine) in this study and  $\text{Quad}_{57}\text{Jd}_{42}\text{Ae}_1$  (with 1 mol. % aegirine) of Xu et al. (2019) are the same  
338 under ambient conditions but inconsistent under high-pressure and high-temperature conditions.  
339 Therefore, the  $K_{T0}$  and  $K_{T0}'$  of the two omphacites are somewhat consistent with each other, while  
340 the thermal expansion and  $(\partial K_T/\partial T)_P$  are different. Collectively, the thermal EoS parameters are of  
341 the essence in the derivation of the mineral density.

## 342 **5.2 The density of eclogite in Tibet**

343 Eclogitized crust and lithospheric mantle may be potential factors causing delamination (Faccenda  
344 et al., 2009; Krystopowicz and Currie, 2013). The density of eclogite and peridotite can provide  
345 new constraints to control the breakoff of the subducted slab and convective removal of the  
346 lithosphere in the process of delamination. Therefore, we plot the density distribution of eclogite  
347 with different garnet contents and peridotite along the Paleozoic and Cenozoic Tibetan geotherms,  
348 as shown in Figure 5. In our model, the mineral composition of Tibetan eclogite is in the range of  
349 40 vol. % garnet + 55 vol. % omphacite + 5 vol. % epidote to 60 vol. % garnet + 35 vol. %  
350 omphacite + 5 vol. % epidote based on the exposed eclogite in Tibet (the composition of epidote is  
351 only 5 vol. % default due to its low content in this study). The composition of surrounding  
352 peridotite consists of 70 vol. % olivine + 25 vol. % orthopyroxene + 3 vol. % clinopyroxene + 2



353 vol. % spinel (Yang et al., 2019; Zhao et al., 2021). The densities of eclogite and peridotite  
 354 aggregates are obtained considering their arithmetic mean. The densities of each mineral under  
 355 specific temperature and pressure conditions are derived by the following formula:

$$356 \quad \rho(T, P) = \frac{V(T, 0)}{V(T, P)} \times \frac{Z \times M}{N_a \times V_0} \quad (4)$$

357 where  $V_0$  is the reference unit cell volume at ambient conditions,  $M$  is molecular weight,  $Z$  is the  
 358 number of formula units in the unit cell and  $N_a$  is the Avogadro number.

359 Most changes in the deep conditions of the Earth are progressing slowly, so there is adequate  
 360 time for recrystallization to relieve the maximum stress point (Robertson, 1988; Skinner, 1966).  
 361 Here, we assume that the elastic-plastic interaction among different minerals and possible  
 362 deviations from hydrostatic conditions are ignored and the density of the eclogite aggregate can be  
 363 obtained by the arithmetic mean as follows:

$$364 \quad \bar{\rho} = \sum \lambda_i \rho_i(T, P) \quad (5)$$

365 where the subscript  $i$  denotes the  $i$ th mineral of the upper mantle, and  $\lambda$  is the volume proportion  
 366 of each mineral.

367 The densities of Tibetan eclogite (with the garnet composition of  $\text{Prp}_{21}\text{Alm}_{47}\text{Grs}_{31}\text{Sps}_1$ , the  
 368 omphacite composition of  $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$ , and the epidote composition of  
 369  $\text{Ca}_{2.02}\text{Fe}_{0.75}\text{Al}_{2.32}\text{Si}_{0.16}(\text{SiO}_4)(\text{Si}_2\text{O}_7)\text{O}(\text{OH})$ ) and peridotite (with the olivine composition of  
 370  $\text{Fo}_{89.9}\text{Fa}_{10.1}$ , the orthopyroxene composition of  $\text{En}_{89.6}\text{Fs}_{9.7}\text{Wo}_{0.7}$ , the clinopyroxene composition of  
 371  $\text{Quad}_{88.5}\text{Jd}_{11.5}$ , and the spinel composition of  
 372  $(\text{Mg}_{0.790}\text{Fe}_{0.204}\text{Ni}_{0.005}\text{Ti}_{0.001})_{1.000}(\text{Al}_{0.821}\text{Cr}_{0.158}\text{Fe}_{0.021})_{2.002}\text{O}_4$ ) in this study along the Paleozoic  
 373 geothermal line are shown in Figure 5a. The results show that the increase in garnet has a  
 374 profound influence on the density of eclogite. For every 10% increase in garnet, the density of

375 eclogite increases by ~1.7%. The garnet content in Tibetan eclogite is estimated to be 40 vol. %-60  
376 vol. % (Fig. S6). The densities of this part of eclogite are 3.54-3.66 g/cm<sup>3</sup>, which is approximately  
377 7.4%-11.2% more than that of peridotite (3.29 g/cm<sup>3</sup>) at ~80 km. The density difference between  
378 eclogite and peridotite is 0.24 g/cm<sup>3</sup>-0.37 g/cm<sup>3</sup> (Fig. 5b). At the same time, we also consider the  
379 density of eclogite aggregates without epidote (Fig. S6). The results show that 5 vol. % epidote  
380 has little effect on the density of eclogite, especially eclogite with garnet contents of 50 vol. %-60  
381 vol. % (Fig. S9). To account for the role of iron, the density distributions of high-Fe  
382 (Prp<sub>14</sub>Alm<sub>62</sub>Grs<sub>19</sub>Adr<sub>3</sub>Sps<sub>2</sub> and Quad<sub>53</sub>Jd<sub>27</sub>Ae<sub>20</sub> and low-Fe eclogite (Prp<sub>28</sub>Alm<sub>38</sub>Grs<sub>33</sub>Sps<sub>1</sub> and  
383 Quad<sub>57</sub>Jd<sub>42</sub>Ae<sub>1</sub>) are plotted to better constrain the range of eclogite density (Fig. S10) (Xu et al.,  
384 2019). For high-Fe and low-Fe eclogites, the densities of eclogite increase by ~1.9% and ~1.4%  
385 for each 10% increase in garnet, respectively. The densities of eclogite are 3.64 g/cm<sup>3</sup>-3.78 g/cm<sup>3</sup>  
386 for high-Fe content and 3.53 g/cm<sup>3</sup>-3.63 g/cm<sup>3</sup> for low-Fe content at ~80 km. Furthermore, the  
387 densities of high-Fe and low-Fe eclogites are 10.6%-14.9% and 7.2%-10.3% higher than the  
388 surrounding peridotite, respectively. For a more straightforward comparison, taking eclogite  
389 containing 50 vol. % garnet as an example (Fig. S11), the densities of high-Fe eclogite, low-Fe  
390 eclogite, and Tibetan eclogite at ~80 km are 3.71 g/cm<sup>3</sup>, 3.58 g/cm<sup>3</sup>, and 3.61 g/cm<sup>3</sup>, respectively.  
391 An increase in the iron content can substantially increase the density of eclogite, although it will  
392 be constrained by the thermal EoS parameters of minerals.

393 Similarly, we also discuss the density profile along the Cenozoic geothermal line, which can be  
394 seen in supporting information Text S2. In any case, the density difference caused by eclogite may  
395 be one of the prominent factors instigating the delamination process.

### 396 **5.3 Influence of the degree of eclogitization on the density of the subducted slab**

397 Eclogite in the mantle, which is believed to be 5%-10% denser than peridotite (Garber et al.,  
398 2018), is responsible for the excess compositional density. Furthermore, some calculations  
399 propose that the degree of eclogitization of the subducted slab is a key factor in the delamination  
400 process (Matchette-Downes et al., 2019). To investigate the influence of the degree of  
401 eclogitization in the delamination process, we plot the density variations with different mineral  
402 compositions under different degrees of eclogitization (Fig. 6). We consider eclogitization in the  
403 lithospheric mantle of the subducted slab, here the degree of eclogitization refers to the amount of  
404 eclogite in the lithospheric mantle. In our preferred model, the 7-km thick subducted oceanic crust  
405 becomes eclogite, while the lithospheric mantle constrains a different amount of eclogite. Since  
406 the subducted Indian oceanic slab might be fragmented into several pieces, the longitudinal size of  
407 the fractured slab is postulated to be 60 km (Peng et al., 2016). Our estimated average density of  
408 the fragmented slab with various degrees of eclogitization is shown in Figure 6a. The results  
409 clearly show that the density increases monotonically with the garnet content and the degree of  
410 eclogitization. The garnet content is of profound importance to the density of eclogite. The higher  
411 the proportion of garnet is, the greater the density increases with increasing degrees of  
412 eclogitization. The garnet content in Tibetan eclogite is estimated to be between 40-60 vol. %.  
413 Taking garnet with an average volume percentage of 50 vol. % in Tibetan eclogite as an example,  
414 the density of eclogitized subducted slabs ranges from 3.35 g/cm<sup>3</sup> with 10% eclogitization to 3.61  
415 g/cm<sup>3</sup> with 100 vol. % eclogitization. For a garnet content of 50 vol.%, the density increases by  
416 0.029 g/cm<sup>3</sup> per 10 vol. % increase in the degree of eclogitization. The density will increase with  
417 increasing garnet contents, from 0.006 g/cm<sup>3</sup> for 10 vol. % to 0.051 g/cm<sup>3</sup> for 90 vol. %. The  
418 densities of high-Fe and low-Fe eclogitized fragmented slabs are also shown in Figure S12. The

419 high-Fe content shows that the density variation increases with the degree of eclogitization from  
420 0.007 g/cm<sup>3</sup> for 10 vol. % to 0.064 g/cm<sup>3</sup> for 90 vol. % garnet, while the low-Fe content shows a  
421 density change from 0.004 g/cm<sup>3</sup> for 10 vol. % garnet to 0.045 g/cm<sup>3</sup> for 90 vol. % garnet.

## 422 **5.4 Delamination in Tibet**

423 The development of delamination is associated with the instability of the lower crust and the  
424 mantle lithosphere. The eclogitization of the subducted slab and lower crust plays a vital role in  
425 the process of delamination due to the high density of eclogite, which makes the formation denser  
426 than the surrounding mantle lithosphere and provides critical negative buoyancy (Göğüş and Ueda,  
427 2018; Krystopowicz and Currie, 2013). The densities of the eclogitic lower crust and mantle  
428 lithosphere during slab subduction and convective removal are sufficiently higher than that of the  
429 asthenosphere and are good candidates for the initiation of destabilization.

### 430 **5.4.1 Subducted slab breakoff**

431 A series of collisional breakoff events is proposed to have occurred throughout 60-45 Ma in Tibet  
432 (Chung et al., 2005, 2009; Ma et al., 2014; Zhu et al., 2015). The formation of eclogite  
433 presumably kick-starts slab breakoff during the subduction of the Indian oceanic plate underthrust  
434 below the southern margin of Tibet. The subducted Indian oceanic slab fragmented into several  
435 pieces, due to what has been identified as a high-velocity anomaly (Peng et al., 2016; Shi et al.,  
436 2020b). The seismological evidence of high density (Hetényi et al., 2007), high  $V_P$  (Schulte-  
437 Pelkum et al., 2005), and low longitudinal/transverse ( $V_P/V_S$ ) ratios (Wittlinger et al., 2009) further  
438 confirms that there may be variable degrees of eclogitization beneath Tibet. Figure 6 shows the  
439 density profile of subducted slabs with different garnet compositions, different degrees of  
440 eclogitization, and variable densities compared with the surrounding peridotite. An increasing

441 degree of eclogitization and an enhanced garnet content in eclogite increases the density difference  
442 between the slab and the surrounding peridotite. Previous studies have made preliminary estimates  
443 of the average density from the isostatic balance and geoid anomalies and postulated that the  
444 density excess could be between 0-0.19 g/cm<sup>3</sup> (Matchette-Downes et al., 2019). For Tibetan  
445 eclogite containing 40 vol. %-60 vol. % garnet, if the lithospheric mantle is a mixture of peridotite  
446 and eclogite with a density anomaly of 0.19 g/cm<sup>3</sup>, our model requires a range of 44%-70%  
447 degrees of eclogitization. If the eclogite is high-Fe, only a 30%-48% degree of eclogitization is  
448 needed to produce the density difference (Fig. S12), while an eclogitization degree is in the range  
449 of 49%-74% is needed for the low-Fe eclogite. However, some seismological data show that the  
450 crust or lithospheric mantle being only ~30% eclogitized might cause gravitational instability in  
451 Tibet (Matchette-Downes et al., 2019; Shi et al., 2020a), which is lower than our estimation. Our  
452 results clearly show that density excess is closely linked with garnet content and eclogitization  
453 degree. If eclogite has a high garnet content, a relatively low degree of eclogitization could  
454 instigate the delamination of slab breakoff.

455 On the other hand, the presence of a weak lower crust and a vertical conduit to accommodate  
456 asthenosphere influx is also necessary for the delamination process. The weak layer between the  
457 residual crustal and downward peeling lithosphere layer (and/or lower crust) (Göğüş and Ueda,  
458 2018) could promote the initiation and propagation of delamination. Therefore, very high  
459 temperatures and relatively low lower-crustal viscosities are also other controlling factors of  
460 delamination (Göğüş and Pysklywec, 2008; Morency, 2004). Here, we assume that the length of  
461 the fractured slab is 60 km, which drops 80 km over 45 Ma and that the viscosity of the  
462 asthenosphere is  $5 \times 10^{20}$  Pa·S (Wang et al., 2019). By using Stokes' Law (Supporting Information

463 Text S3), ignoring the thermal disturbance, and assuming the most ideal conditions, the density  
464 difference caused by eclogite needs to be at least  $0.15 \text{ g/cm}^3$  to produce such delamination. The  
465 result is close to those discussed above in gravity anomalies.

466 In particular, the presence of eclogite with a greater abundance of garnet, a higher-Fe content,  
467 and a greater degree of eclogitization would instigate the delamination process of slab breakoff.

#### 468 **5.4.2 Removal of the eclogitized lower crust**

469 The thickened lower crust undergoes “convective removal” due to gravitational instability, which  
470 is another type of delamination that occurred in Tibet from 25 Ma to 0 Ma (Chung et al., 2005;  
471 Nomade et al., 2004). The convective removal of the lithosphere during delamination corresponds  
472 to higher temperature conditions (Craig et al., 2020). In this circumstance, the density of Tibetan  
473 eclogite is  $6.9\%-10.8\%$  denser than the surrounding peridotite at  $\sim 60 \text{ km}$  (Fig. 5b), which is  
474 analogous to the results in the case of subducted slab detachment. This result is also in ample  
475 agreement with the result obtained by Garber et al. (2018), which noted that eclogite is  $5\%-10\%$   
476 denser than peridotite. The density difference between eclogite and peridotite is  $0.22 \text{ g/cm}^3\text{-}0.35$   
477  $\text{g/cm}^3$  with 40 vol. %-60 vol. % garnet in Tibet (Fig. 5d). During this stage, it is believed that  
478 delamination of the thickened, eclogitized lower crust has occurred. Similarly, Stokes’ law can be  
479 used considering ideal conditions without any thermal disturbance. If the falling block is assumed  
480 to be approximately 30 km in the longitudinal direction and the viscosity of the asthenosphere is  
481  $5 \times 10^{20} \text{ Pa}\cdot\text{S}$ , the falling block can drop by  $70\text{-}110 \text{ km}$  within 25 Ma. For eclogite with a high-Fe  
482 content, a density difference of  $0.35 \text{ g/cm}^3\text{-}0.50 \text{ g/cm}^3$  makes the fragmented block capable of  
483 falling 105-155 km, while the density difference of  $0.24 \text{ g/cm}^3\text{-}0.33 \text{ g/cm}^3$  with a low-Fe content  
484 makes the block able to fall 75-102 km (Fig. S13). The fragmented block with a high-Fe content

485 can fall a larger distance at the same time, indicating that the high-Fe content is more likely to  
486 promote the occurrence of delamination. This result is consistent with the high-velocity  
487 anomalous blocks identified at 100-200 km by seismic tomography (Peng et al., 2016; Shi et al.,  
488 2016, 2020a).

489 In summary, density contrasts can provide a stimulus for the initiation of instability. It is accepted  
490 that eclogite with a high garnet content and a high Fe content and a high proportion of eclogite in  
491 the lithospheric mantle may have strongly promoted delamination during the process of India-Asia  
492 collision from the perspective of density.

493

## 494 **6. Conclusion**

495 The *P-V-T* EoS of the main minerals of eclogite is combined with its mineral composition  
496 and the geothermal line to derive the density of Tibetan eclogite in this study. We offer a new  
497 perspective by obtaining the thermal EoS for the main minerals of eclogite in a single experiment.  
498 The thermal EoS parameters of the main minerals of eclogite are derived by fitting the *P-V-T* data  
499 to the HT-BM-EoS. The density of minerals along the Tibetan geotherm shows that the density is  
500 closely related to its composition and thermal EoS parameters. Increasing iron contents increase  
501 the density of minerals, but if the molecular masses of two minerals are similar, the thermal EoS  
502 parameters play a pivotal role. The garnet content profoundly increases the density of eclogite. For  
503 every 10 vol. % increase in garnet, the density of eclogite increases by approximately 1.7%. The  
504 density of Tibetan eclogite is approximately 7-11% denser than that of the surrounding peridotite.  
505 An increasing proportion of garnet, Fe content, and degree of eclogitization enhance the density  
506 difference to facilitate the delamination process. For Tibetan eclogite containing 40-60 vol. %

507 garnet, 44-70% degrees of eclogitization can produce the same density difference as obtained by  
508 the isostatic balance and the geoid anomaly. According to a rough calculation, the fragmented  
509 block will fall 70-155 km. A high-Fe content is more likely to promote delamination. Eclogite is a  
510 good candidate for the initiation of instability and may be more susceptible to inducing the  
511 breakoff of the subducted slab or the gravitational removal of the lower crust during the process of  
512 the India-Asia collision.

513

#### 514 **Data availability**

515 All the data presented in this paper are available upon request.

516

#### 517 **Author contributions**

518 All authors contributed to the preparation and revision of the manuscript. Z. Ye: Data curation,  
519 Investigation, Formal analysis, Writing-original draft, Writing-review & editing. D. Fan:  
520 Investigation, Conceptualization, Supervision, Methodology, Funding acquisition, Writing-review  
521 & editing. B. Li: Data curation, Writing-review & editing. Q. Tang: Software, Validation, Writing-  
522 review & editing. J. Xu: Investigation, Supervision, Writing-review & editing. D. Zhang: Formal  
523 analysis, Writing-review & editing. W. Zhou: Investigation, Conceptualization, Supervision,  
524 Writing-review & editing.

525

#### 526 **Competing interests**

527 The authors declare that they have no conflict of interest.

528



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532

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540

541 **Supplementary Information.** The supplementary information describes the density profile of  
542 garnet at high temperature, density profile along Cenozoic geothermal line, data of unit-cell  
543 parameters of eclogite minerals, thermal EoS parameters of this study and previous researches,  
544 figures of Eulerian finite strain-normalized pressure ( $F_E - f_E$ ), isothermal bulk modulus ( $K_{T0}$ ) and its  
545 pressure derivative ( $K_{T0}'$ ) plot of garnet and omphacite, normal distribution of eclogite minerals,  
546 density evolution of minerals, and density profile of different Fe-content eclogite.

547

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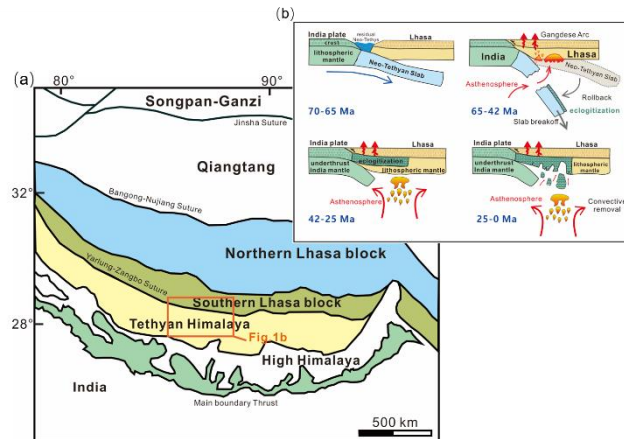
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854 **Figure:**



855

856 **Figure 1.** (a) Schematic geological map of the Tibetan Plateau (modified from Chung et al. 2005

857 and Wang et al. 2010). (b) Interpretive geological cartoon of India-Asia collision evolution. 70-65

858 Ma: The flat Neo-Tethyan oceanic slab subducts beneath Tibet with the closure of the Neo-Tethys

859 Ocean. 65-42 Ma: The rollback of the Neo-Tethyan slab breaks off after densification by

860 eclogitization. 42-25 Ma: The subduction of the Indian continent continued at a low subduction

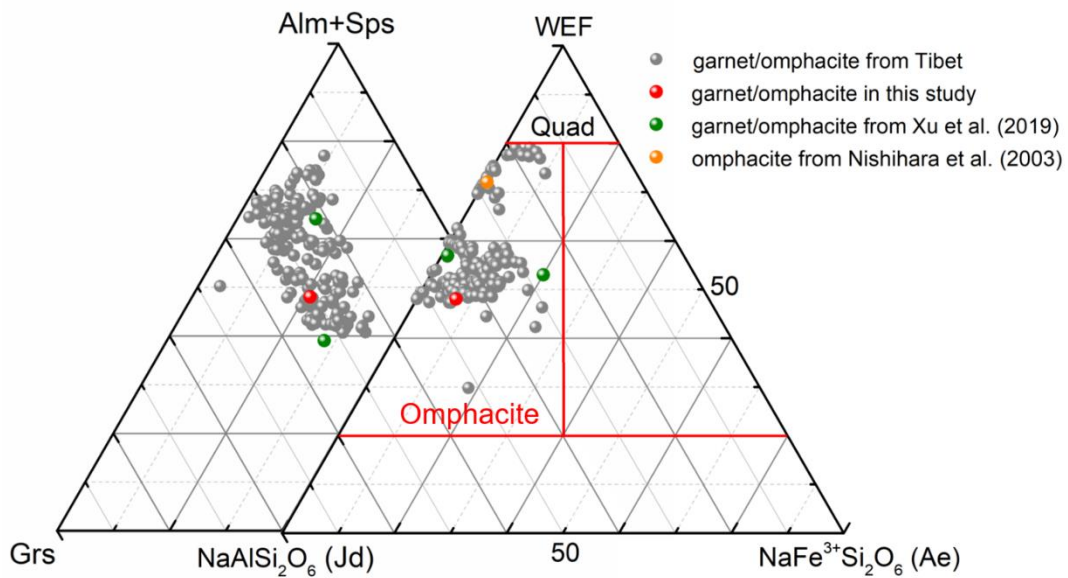
861 angle beneath the Lhasa terrane and was accompanied by heavy thermal perturbation. 25-0 Ma:

862 The thickened eclogitic lower crust undergoes the “convective removal” of delamination due to

863 gravitational instability.

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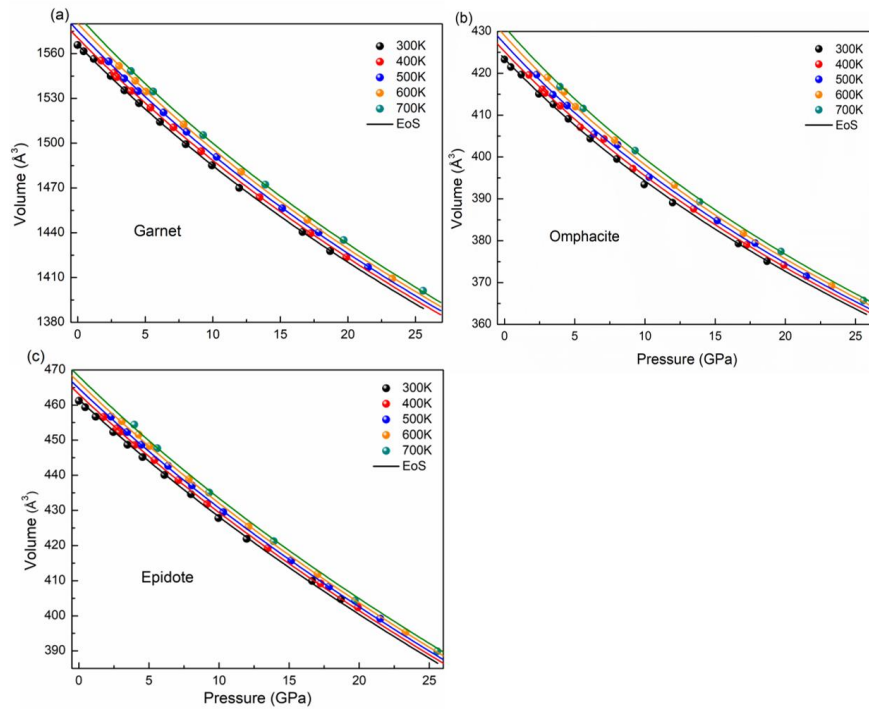




865

866 **Figure 2.** Composition of garnet and omphacite in eclogites from Tibet and this study. The gray  
 867 solid circles represent the components of garnet and omphacite collected from previous studies in  
 868 Tibet (e.g. Chan et al., 2009; Liu et al., 2019; Song et al., 2003; Weller et al., 2016; Yang et al.,  
 869 2009; Zhai et al., 2011a). The green solid circles are garnet and omphacite with different Fe  
 870 contents according to Xu et al. (2019). The orange solid circles are omphacite according to  
 871 Nishihara et al. (2003). The red solid circles are the components of garnet and omphacite in this  
 872 study. Prp = pyrope, Alm = almandine, Grs = grossular, Sps = spessartine, Quad = Ferrosilite +  
 873 enstatite + wollastonite, Jd = jadeite, and Ae = aegirine.

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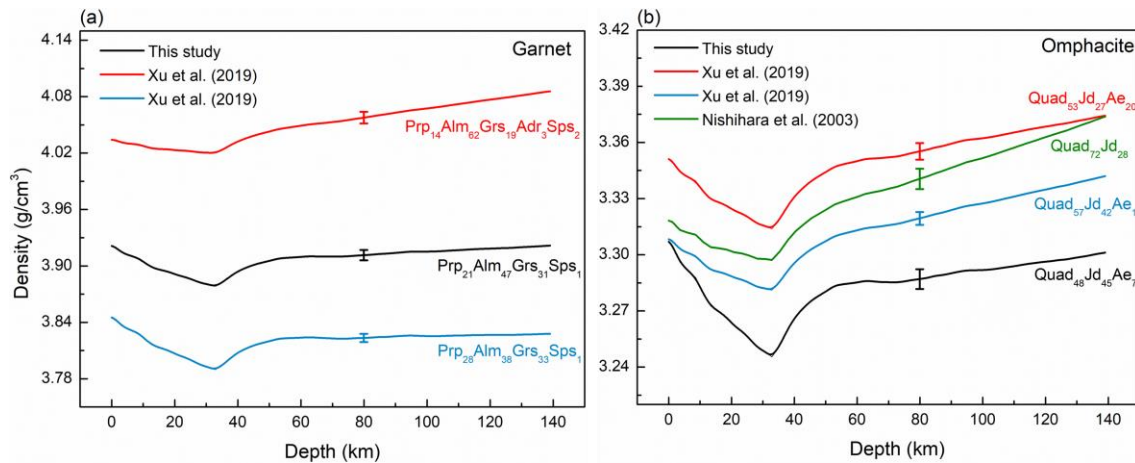


875

876 **Figure 3.** Pressure-volume-temperature relations of garnet (a), omphacite (b), and epidote (c).

877 Isothermal compression curves are calculated by using the thermoelastic parameters obtained in

878 this study.



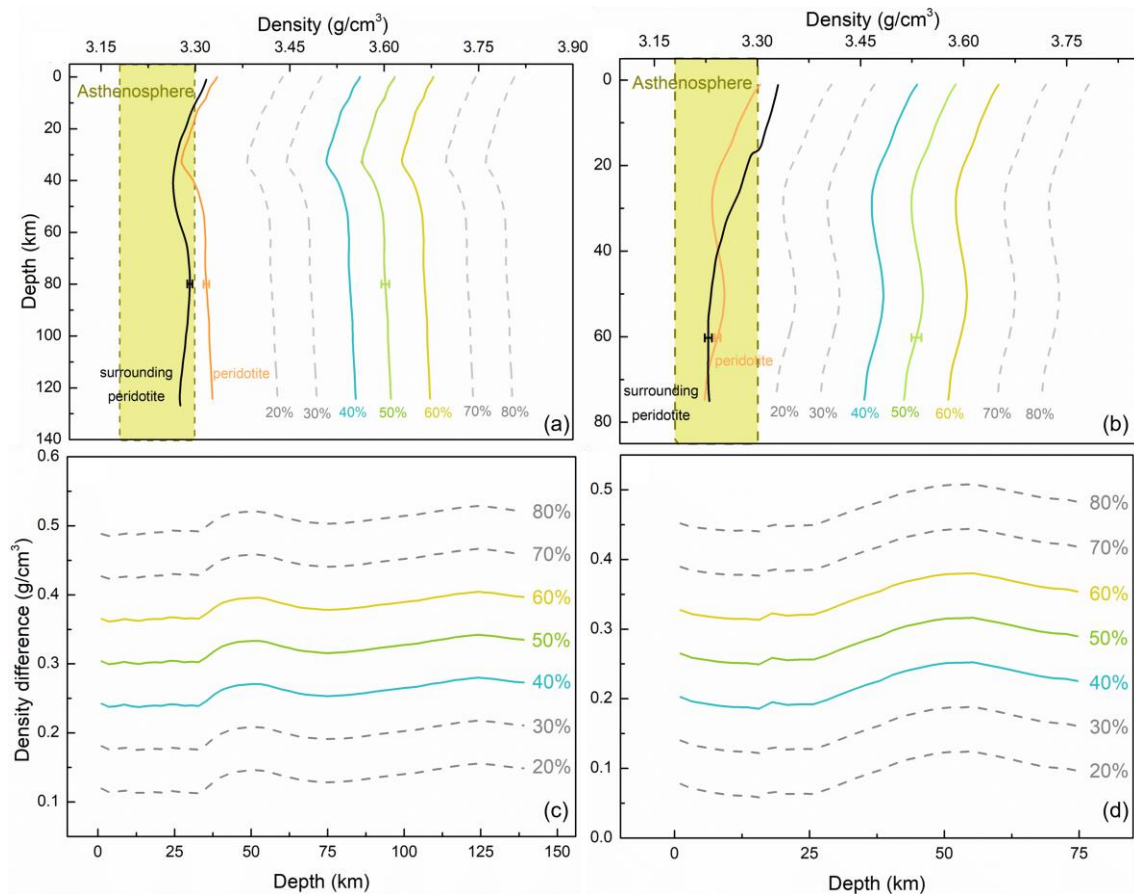
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880 **Figure 4.** Density profiles of garnet (a) and omphacite (b) along with the cold Tibetan geothermal

881 line (Wang et al., 2013). The garnets of Prp<sub>21</sub>Alm<sub>47</sub>GrS<sub>31</sub>Sps<sub>1</sub> and Prp<sub>28</sub>Alm<sub>38</sub>GrS<sub>33</sub>Sps<sub>1</sub> are from

882 Xu et al. (2019). The omphacites of Quad<sub>53</sub>Jd<sub>27</sub>Ae<sub>20</sub> and Quad<sub>57</sub>Jd<sub>42</sub>Ae<sub>1</sub> are from Xu et al. (2019)

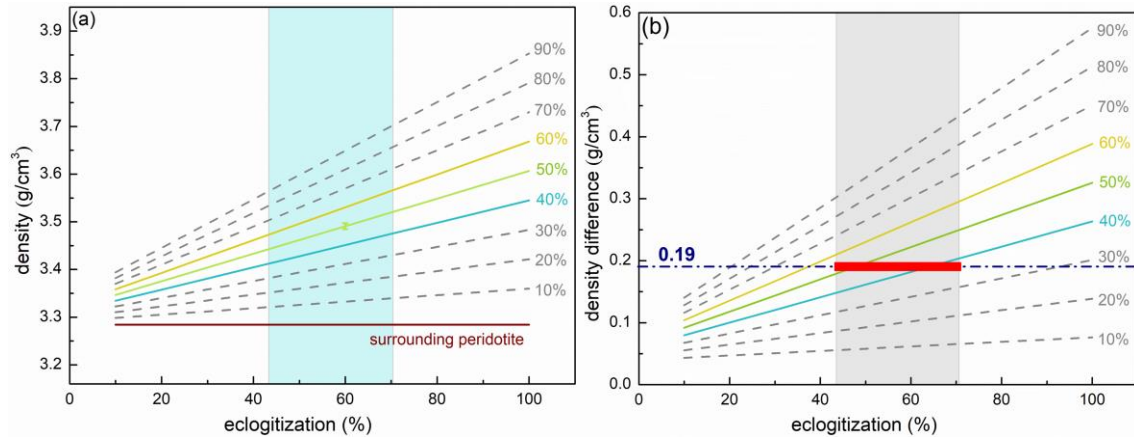
883 and Quad<sub>72</sub>Jd<sub>28</sub> is from Nishihara et al. (2003).



884

885 **Figure 5.** Density profiles of eclogite and peridotite assemblages ((a) and (b)) and density  
 886 difference between eclogite and peridotite ((c) and (d)) in Tibet along the Paleozoic and Cenozoic  
 887 geothermal lines under the conditions of Neo-Tethyan oceanic slab detachment (a) (Wang et al.,  
 888 2013) and subduction of the Indian continental margin beneath the Lhasa terrane (b) (Craig et al.,  
 889 2020). The percentage represents the content of garnet in eclogite, of which epidote accounts for 5  
 890 vol. % by default. The orange curve and black curve show the density profile of peridotite with a  
 891 composition of 70 vol. % olivine + 25 vol. % orthopyroxene + 3 vol. % clinopyroxene + 2 vol. %  
 892 spinel. The orange line shows the density of peridotite in the lithospheric mantle along the  
 893 Paleozoic (a) (Wang et al., 2013) and Cenozoic (b) (Craig et al., 2020) geothermal lines, and the  
 894 black curve indicates that the density of peridotite in the surrounding lithospheric mantle is along  
 895 the Paleozoic (a) (Nábělek and Nábělek, 2014) and Cenozoic (b) (Wang et al., 2013) geothermal

896 lines in Tibet. The shaded region is the density range of the asthenosphere (Chen and Tenzer, 2019;  
 897 Levin, 2006; Panza et al., 2020; Singh and Mahatsente, 2020).  
 898



899  
 900 **Figure 6.** (a) The effect of eclogitization on the density of the subducted slab at ~80 km (2.6 GPa  
 901 and 625 °C) along the Paleozoic geothermal line. The percentage on the right represents the  
 902 content of garnet and the content of epidote is fixed at 5 vol. % by default. The content of garnets  
 903 in Tibet is between 40 vol. % and 60 vol. %. The density represents the average density of the  
 904 subducted slab with the entire eclogitic ocean lower crust and partially eclogitized lithospheric  
 905 mantle, where the degree of eclogitization refers to the lithospheric mantle. The rufous line  
 906 represents the average density of surrounding peridotite in this study. The blue shading indicates  
 907 the possible degree of eclogitization. (b) Density difference between eclogite with different  
 908 degrees of eclogitization and surrounding peridotite. The red dashed solid line represents a density  
 909 excess of 0.19 g/cm<sup>3</sup> from the isostatic balance and the geoid anomaly (Matchette-Downes et al.,  
 910 2019).  
 911