



1 **Delamination in Tibet: Deriving constraints from the density of eclogite**

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11

12 **Abstract**

13 Tibet, which is characterized by collisional orogens, has undergone the process of delamination or
14 convective removal. The lower crust and mantle lithosphere appear to have been removed through
15 delamination during orogenic development. Numerical and analog experiments demonstrate that
16 the metamorphic eclogitized oceanic subduction slab or lower crust may promote gravitational
17 instability due to its increased density. The eclogitized oceanic subduction slab or crustal root is
18 believed to be denser than the underlying mantle and tends to sink. However, the density of
19 eclogite under high-pressure and high-temperature conditions and density differences from the
20 surrounding mantle is not precisely constrained. Here, we offer new insights into the derivation
21 of eclogite density with a single experiment to constrain delamination in Tibet. Using *in situ*
22 synchrotron X-ray diffraction combined with diamond anvil cell, experiments focused on minerals



23 (garnet, omphacite, and epidote) of eclogite are conducted under simultaneous high-pressure and
24 high-temperature conditions, which avoids systematic errors. Fitting the
25 pressure-temperature-volume data with the third-order Birch-Murnaghan equation of state, the
26 thermal equation of state (EoS) parameters, including the bulk modulus (K_{T0}), its pressure
27 derivative (K'_{T0}), the temperature derivative ($(\partial K_T / \partial T)_P$), and the thermal expansion coefficient
28 (α_0), are derived. The densities of rock-forming minerals and eclogite are modeled along with the
29 geotherms of two types of delamination. The delamination processes of subduction slab breakoff
30 and the removal of the eclogitized lower crust in Tibet are discussed. The Tibetan eclogite which
31 containing 40-60 vol. % garnet and 37-64% degrees of eclogitization can promote the
32 delamination of slab break-off in Tibet. Our results indicate that eclogite is a major controlling
33 factor in the initiation of delamination. A high abundance of garnet, a high Fe-content, and a high
34 degree of eclogitization are more conducive to instigating the delamination.

35 **Keywords:**

36 Eclogite, Equation of state, Single-crystal X-ray diffraction, Delamination, Tibet

37

38 **1. Introduction**

39 The evolution of orogenesis is characterized by lithospheric removal during rapid surface uplift,
40 mantle upwelling, and postcollisional magmatism, particularly in the Central Andes (e.g. Ehlers
41 and Poulsen, 2009; Schurr et al., 2006), Himalayas (e.g. Jiménez-Munt et al., 2008; Singh and
42 Kumar, 2009), and Dabie orogen (e.g. He et al., 2011; Zhang et al., 2010).

43 It is widely accepted that delamination is the most important mechanism of lithospheric
44 removal. Delamination is induced and accompanied by two major requisites: (a) the density



45 difference caused by the negative buoyancy of the delaminated lithosphere; and (b) the presence
46 of a weak lower crust (lower viscosity) that exists between the strong upper crust and lithospheric
47 mantle. Usually, two types of delamination are believed to occur in orogen development. The first
48 is the conventional definition of delamination proposed by Bird (1978, 1979), which was used to
49 interpret the geodynamic evolution of the Colorado Plateau. In this scenario, mantle lithosphere
50 peels back from the overlying upper crust and is removed entirely, with the rising hot mantle
51 filling the lithospheric removal zone (e.g. Göğüş and Ueda, 2018; Krystopowicz and Currie, 2013;
52 Schott and Schmeling, 1998; Sobolev and Babeyko, 2005). A weak decoupling layer, i.e. the lower
53 crust, is an essential condition in this delamination model, which may be affected by the
54 rheological behavior of the hydration, thermal, and chemical characteristics of the lithosphere (e.g.
55 Burov and Watts, 2006; Morency, 2004; Schott and Schmeling, 1998). In addition to conventional
56 delamination, an alternative delamination mechanism is convective removal based on the
57 Rayleigh-Taylor-type instability model (Houseman et al., 1981), namely, viscous “dripping”. This
58 model postulates that there is sufficient perturbation in the lithospheric mantle, which is ascribed
59 to the strong temperature-dependence of typical mantle rheology, without regard to a specific
60 weak layer (e.g. Conrad and Molnar, 1999; Gorczyk et al., 2012; Houseman and McKenzie, 1982;
61 Schott and Schmeling, 1998).

62 All previous studies attribute the gravitational instability process of lithospheric removal to the
63 negative thermal buoyancy of the cold lithosphere (Conrad and Molnar, 1999; Houseman and
64 McKenzie, 1982) or density contrast between asthenosphere and mantle lithosphere
65 (Elkins-Tanton, 2007; Neil and Houseman, 1999). In any case, the density distribution with
66 lithosphere pressure and temperature (P - T) conditions and chemical composition is of vital



67 importance to understanding the process of lithospheric removal.

68 The Tibetan Plateau is the most representative and prominent collisional orogens. Two types of
69 delamination are proposed to proceed (e.g. Chung et al., 2005; Houseman et al., 1981; Molnar et
70 al., 1993; Platt and England, 1994; Sun et al., 2020): lithospheric mantle removal and thickened
71 eclogitized crust removal. The Neo-Tethyan oceanic subduction, India-Asia collision, and Indian
72 continental subduction could be further considered responsible for the abnormal thinning of the
73 mantle lithosphere under Tibet (Chung et al., 2005; DeCelles et al., 2011; Li et al., 2019; Ma et al.,
74 2017; Xu et al., 2008; Zhao et al., 2020). The lithospheric removal event in Tibet corresponds to
75 Neo-Tethyan oceanic slab break-off. The mechanism is primarily based on density contrasts
76 between the denser mantle lithosphere and the lighter underlying mantle. Some models reveal that
77 lithospheric removal is induced by the retreating high-density eclogitized lithosphere detached
78 from overlying low-density crust (Faccenda et al., 2009; Li et al., 2016; Ueda et al., 2012). Other
79 alternative models indicate that thickened eclogitized crust is a potential factor deriving
80 lithospheric removal because the eclogitized crustal root is denser than the underlying mantle and
81 tends to sink (Krystopowicz and Currie, 2013). Regardless of the above types of delamination, the
82 density of eclogite is closely related to delamination. Therefore, Tibet provides an excellent
83 opportunity to understand the role of eclogite density in the process of delamination.

84 An immense amount of concrete research has focused on the origin and appearance of
85 lithospheric mantle removal from different angles, such as geophysical (Ren and Shen, 2008;
86 Tilmann, 2003), geological (Chung et al., 2005; Molnar et al., 1993), petrological (Chung et al.,
87 2005; Turner et al., 1993), numerical and analog experiments (Bajolet et al., 2012; Göğüş and
88 Pysklywec, 2008; Morency, 2004; Valera et al., 2011). In particular, numerical and analog



89 experiments are used as prominent methods to simulate the dynamics of delamination (Göğüş and
90 Ueda, 2018). Of these studies, the density behavior occurring during the delamination process has
91 also been investigated intensively following thermodynamic (Duesterhoeft et al., 2012; Semprich
92 et al., 2010), seismic/tomography (Li and Fang, 2017; Matchette-Downes et al., 2019), and
93 numerical simulations (Gerya et al., 2004; Li et al., 2016; Sobolev and Babeyko, 2005). However,
94 few studies have systematically illuminated the issue of delamination from the perspective of
95 eclogite density. Here, we attempt to offer new insights into the derivation of rock density through
96 the mineral physics method to constrain delamination in Tibet (Ye et al., 2021). Conducting a
97 single experiment under high-pressure and high-temperature conditions, we obtain the equation of
98 state (EoS) of the main minerals of eclogite with fewer systematic errors in the experiment.
99 Furthermore, the newly derived EoS of the main minerals of eclogite, combined with the
100 published EoSs of the main minerals of peridotite (Ye et al., 2021), geothermal lines, and collected
101 eclogite mineral compositions, are further used to elucidate a density evolution model during the
102 delamination process in Tibet. We argue that the EoSs of minerals could be used in a
103 straightforward manner as new constraints on the construction of the density model. Using a
104 simplistic calculation setup, in this study, this density evolution model will shed light on the
105 possibility of delamination during the orogen process.

106

107 **2. Geological background**

108 The Tibetan Plateau is composed of four terranes from south to north: the Himalaya, Lhasa,
109 Qiangtang, and Songpan-Ganzi terranes (Fig. 1). The birth of the Himalayas and Tibetan Plateau
110 is a consequence of the Indo-Asian collision, which began in the early Cenozoic (Hodges et al.,



111 2001; Rowley, 1998; Wang et al., 2008). The Neo-Tethyan oceanic slab is proposed to have
112 detached from the Indian lithosphere, and the onset of the Indo-Asian collision (DeCelles et al.,
113 2002; Kohn and Parkinson, 2002) particularly occurred in the lower part of the Indian and Lhasa
114 lithospheres. The tectonic evolutionary history of the Lhasa terrane and Tethys Himalayas is
115 essential for revealing the origin of the Himalayan-Tibetan orogen. The subducting Neo-Tethyan
116 slab was thrust into southern Tibet approximately 70-65 Ma (Fig. 1b). With the closure of the Neo
117 Tethyan Ocean, the India-Asia continent collision caused compressional deformation in southern
118 Tibet (Ding et al., 2003), and a series of collision breakoff events were delineated spanning from
119 65 Ma to 42 Ma (Chung et al., 2005, 2009; Lee et al., 2009; Leech et al., 2005; Ma et al., 2014;
120 Zhu et al., 2011, 2015). During this period, the Indian continental lithosphere might have dragged
121 down to deeper depths during subduction (Chemenda et al., 2000). Meanwhile, slab rollback
122 accompanied by the southward migration of asthenospheric convection in Tibet changed the
123 thermal structure of the mantle wedge. The breakoff of the oceanic Neo-Tethyan slab from the
124 more buoyant Indian continental lithosphere indicated by the eruption of early Eocene Linzizong
125 volcanic rocks in the Gangdese arc or the cessation of Gangdese arc magmatism occurred at ~45
126 Ma (DeCelles et al., 2002), which opened a channel for the upwelling asthenosphere (Chung et al.,
127 2009; Ma et al., 2014; Zhu et al., 2015). Additionally, geophysical evidence of longitudinal wave
128 (V_p) tomography is interpreted for the north-dipping high-speed anomaly, which is ascribed to the
129 deep Indian mantle lithosphere (Li et al., 2008; Liang et al., 2016; Tilmann, 2003). Subsequently,
130 the subduction of the Indian continental margin continues at a low subduction angle beneath the
131 Lhasa terrane (Guillot et al., 2008).

132 In addition, 25 Ma to 0 Ma is another period considered to contain either the occurrence of slab



133 breakoff (Jiang et al., 2012; Miller et al., 1999) or lithospheric mantle removal following slab
134 breakoff (Chung et al., 2005; Nomade et al., 2004). Previous studies suggested that the hotter
135 asthenosphere considerably raised the geothermal conditions during this period (Chung et al.,
136 2005; Hou et al., 2004; Williams et al., 2001). Magmatism of the ultrapotassic, shoshonitic, and
137 calc-alkaline was widespread, which was potentially due to the partial melts of the metasomatized
138 lithospheric mantle and eclogitized lower crust. An adopted model of convective lithospheric
139 removal below Lhasa is widely followed (Miller et al., 1999; Molnar et al., 1993; Platt and
140 England, 1994). The lithospheric removal-related mantle upwelling process has been supported by
141 geological, geophysical, and petrological studies (Chung et al., 2005; Molnar et al., 1993; Ren and
142 Shen, 2008; Turner et al., 1993).

143 Here, slab breakoff and convective lithospheric removal under Tibet are adopted as the
144 background in this study to discuss the possibility of the delamination process.

145

146 **3. Materials and methods**

147 **3.1 Starting material**

148 Natural garnet, omphacite, and epidote samples are collected from eclogite in the Dabie-Sulu
149 UHPM belt. The compositions of each mineral are determined to be $\text{Prp}_{21}\text{Alm}_{47}\text{Grs}_{31}\text{Sps}_1$ (Prp =
150 pyrope, Alm = almandine, Grs = grossular, and Sps = spessartine) for garnet, $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$
151 (Quad = Ferrosilite + enstatite + wollastonite, Jd = jadeite, Ae = aegirine) for omphacite, and
152 $\text{Ca}_{2.02}\text{Fe}_{0.75}\text{Al}_{2.32}\text{Si}_{0.16}[\text{SiO}_4][\text{Si}_2\text{O}_7]\text{O}(\text{OH})$ for epidote. The compositions of garnet and omphacite
153 are shown in Figure 2 and are within the range of natural mineral compositions of eclogite from
154 Tibet. The chemical composition of representative epidote minerals in Tibet shows that the Fe



155 content of epidote exposed in eclogite is in the range of 0.13-0.25 ($X_{\text{Fe}}=\text{Fe}^{3+}/(\text{Fe}^{3+}+\text{Al}^{3+})$) (Huang
156 et al., 2015; Li et al., 2017; Liu et al., 2016; Ma et al., 2017; Song et al., 2003; Weller et al., 2016;
157 Yang et al., 2014), while the Fe content of epidote in this study is 0.24, which is within the Fe
158 content range of natural epidote. The samples used in this study are representative of garnet,
159 omphacite, and epidote minerals in natural eclogites from Tibet. The garnet, omphacite, and
160 epidote with high-quality grains are separated from the eclogite specimens. The above three
161 samples are crushed into $30\times 40\ \mu\text{m}^2$ chips with a single crystal thickness of $15\ \mu\text{m}$ in our
162 experiment.

163 **3.2 Synchrotron X-ray diffraction**

164 The high-pressure and high-temperature experiment is conducted by a BX90 externally-heated
165 diamond anvil cell (EHDAC) with $\pm 15^\circ$ opening angles. The above three single crystals are
166 loaded into the BX90 EHDAC equipped with a pair of $500\ \mu\text{m}$ culet-size diamond anvils and WC
167 seats (Figure S1). The rhenium (Re) gasket is pre-indented to a thickness of $\sim 60\ \mu\text{m}$, and a
168 cylindrical hole with a diameter of $360\ \mu\text{m}$ is drilled as a sample chamber. Gold powder is also
169 loaded as the pressure calibrant (Fei et al., 2007), and neon is loaded as the pressure transmitting
170 medium through the GeoSoilEnviroCARS (GSECARS) gas loading system (Rivers et al., 2008).
171 The quasi-hydrostatic condition in the sample chamber can be maintained up to ~ 20 GPa using the
172 neon pressure transmitting medium (Finkelstein et al., 2017). On the other hand, high temperature
173 can significantly decrease the deviatoric stress conditions in the sample chamber. Moreover,
174 previous studies demonstrate that the deviatoric stress disappears at the temperatures of 650 K
175 with neon as the pressure transmitting medium (Klotz et al., 2009; Meng et al., 1993). Therefore,
176 the hydrostatic/quasi-hydrostatic conditions can be maintained within the P - T range of our



177 experiment (~700 K, 25 GPa). An automated pressure-driven membrane system is utilized to
178 generate increasing pressure up to 25.6 GPa. High-temperature conditions up to 700 K are
179 provided by the heating resistor. Setup details for the employed thermocouples and heaters can be
180 found in our previous articles (Xu et al., 2019, 2020b; Ye et al., 2021).

181 *In-situ* synchrotron single-crystal X-ray diffraction (XRD) experiments were performed at
182 experimental station 13-BM-C of the Advanced Photon Source, Argonne National Laboratory. The
183 detailed experimental process and associated parameters can be seen in our previous studies (Xu et
184 al., 2017, 2018, 2020a; Zhang et al., 2017a). The diffraction images and the lattice parameters
185 were analyzed by the Bruker APEX3 software package (Dera et al., 2013). The specific unit-cell
186 parameters of the above three samples at each P - T condition can be found in Table S1.

187

188 4 Results and discussions

189 4.1 EoS of main minerals for eclogite

190 The pressure-volume-temperature (P - V - T) data in this study are fitted by the high-temperature
191 third-order Birch-Murnaghan-EoS (HT-BM3-EoS) (Birch, 1947) to obtain the thermal EoS
192 parameters including the zero-pressure volume (V_{T0}), the isothermal bulk modulus (K_{T0}), and its
193 pressure derivate (K'_{T0}) with the following form:

$$194 \quad P = (3/2)K_{T0} \left[(V_{T0}/V)^{7/3} - (V_{T0}/V)^{5/3} \right] \times \left\{ 1 + (3/4)(K'_{T0} - 4) \left[(V_{T0}/V)^{2/3} - 1 \right] \right\} \quad (1)$$

195 where V_{T0} and K_{T0} at different isotherms are expressed by the following equations:

$$196 \quad V_{T0} = V_0 \exp \int_{300}^T \alpha_T dT \quad (2)$$

$$197 \quad K_{T0} = K_0 + (\partial K_T / \partial T)_P (T - 300) \quad (3)$$

198 where $(\partial K_T / \partial T)_P$ is the temperature derivative of the bulk modulus and α_T ($\alpha_T = \alpha_0 + \alpha_1 T + \alpha_2 T^2$) is
199 the thermal expansion coefficient at room pressure. Considering the limited high temperature



200 experimental data in this study, we fixed α_1 and α_2 as 0 in the fitting.

201 The thermal EoS parameters are derived using the EoSFit program without any constraints at
202 high-pressure and room-temperature and high-pressure and high-temperature conditions (Angel et
203 al., 2014) and are shown in Table S2. Under ambient pressure and temperature conditions, the
204 measured V_0 values of garnet, omphacite, and epidote are 1565.8 (4) \AA^3 , $V_0 = 423.3$ (4) \AA^3 , and V_0
205 $= 461.2$ (2) \AA^3 , respectively. The fitting parameters under high-pressure and room-temperature
206 yield $K_{T0} = 172$ (2) GPa, $K_{T0}' = 3.6$ (2) for garnet, $K_{T0} = 124$ (2) GPa, $K_{T0}' = 3.7$ (4) for omphacite,
207 and $K_{T0} = 122$ (1) GPa, $K_{T0}' = 2.5$ (2) for epidote, respectively.

208 To evaluate the quality of BM3-EoS fitting in this study, the relationship between the Eulerian
209 strain ($f_E = [(V_0/V)^{2/3} - 1]$) and the normalized pressure ($F_E = P / [3f_E(2f_E + 1)^{5/2}]$) of
210 the main minerals for eclogite is plotted in Figure S2. Linear fitting of the three sets of data
211 exhibited a negative slope, indicating that the pressure derivative of the bulk modulus (K_{T0}') is less
212 than 4, which is consistent with our BM3-EoS fittings. The intercept value was obtained by
213 weighted linear regression of the data points, showing that $F_E(0) = 171$ (2) GPa for garnet,
214 $F_E(0) = 123$ (2) for omphacite, and $F_E(0) = 122$ (1) for epidote, respectively. The results are
215 consistent with the fitted isothermal bulk modulus ($K_{T0} = 172$ (2) GPa for garnet, $K_{T0} = 124$ (2)
216 GPa for omphacite, and $K_{T0} = 122$ (1) GPa for epidote, respectively) within the error range.
217 Accordingly, the K_{T0} and K_{T0}' obtained by the BM3-EoS fitting are reasonable. Using the V_0 fixed
218 at ambient conditions to fit HT-BM3-EoS, the available EoS parameters, $K_{T0} = 171.4$ (8) GPa, K_{T0}'
219 $= 3.5$ (1), $(\partial K_T / \partial T)_P = -0.010$ (3) GPaK^{-1} , and $\alpha_0 = 2.86$ (9) $\times 10^{-5}$ K^{-1} for garnet; $K_{T0} = 122$ (2) GPa,
220 $K_{T0}' = 4.1$ (3), $(\partial K_T / \partial T)_P = -0.025$ (6) GPaK^{-1} , and $\alpha_0 = 4.7$ (4) $\times 10^{-5}$ K^{-1} for omphacite; and $K_{T0} =$
221 122.7 (6) GPa, $K_{T0}' = 2.49$ (8), $(\partial K_T / \partial T)_P = -0.029$ (2) GPaK^{-1} , and $\alpha_0 = 4.7$ (1) $\times 10^{-5}$ K^{-1} for



222 epidote are derived. The P - V - T data fitted through the HT-BM3-EoS model are shown in Figure 3.

223 **4.2 Comparison with previous studies**

224 **4.2.1 Garnet**

225 The thermal EoS parameters of garnet are obtained by fitting the P - V - T data to the BM3-EoS. We
226 compare our results with those of previous studies (Arimoto et al., 2015; Gréaux and Yamada,
227 2014; Lu et al., 2013; Milani et al., 2015, 2017; Xu et al., 2019; Zou et al., 2012). The K_{T0} of
228 end-member garnet, pyrope, almandine, grossular, and spessartine crystals is between 158 and 179
229 GPa, and the bulk modulus of almandine is the largest among the above (Table S3). From Table
230 S3, it can be seen that the bulk modulus of powder XRD (Arimoto et al., 2015; Gréaux and
231 Yamada, 2014; Pavese et al., 2001; Zou et al., 2012) are larger than those of single-crystal XRD
232 (Milani et al., 2015, 2017) with the same composition. The K_{T0} of solid solution garnets (Beyer et
233 al., 2021; Jiang et al., 2004; Lu et al., 2013; Xu et al., 2019) is also between 158 and 179 GPa
234 mentioned above and will be affected by the end-member components. The $K_{T0}=171.4$ (8) GPa in
235 this study is reasonable within this range. The obtained $K_{T0}'=3.5$ (1) in this study is slightly lower
236 than that in previous studies. The Eulerian strain and the normalized pressure of the garnet shown
237 in Figure S2(a) exhibit a negative slope, which indicating K_{T0}' is less than 4. Moreover, compared
238 with the previous results, the obtained value of K_{T0}' in this study is within the error range
239 (Supporting Information Text S1). However, there is no obvious correlation between the fitted K_{T0}
240 and K_{T0}' for minerals of different compositions (Fig. S3); hence, the K_{T0} may not be precise when
241 K_{T0}' is fixed. The value of $(\partial K_T/\partial T)_P = -0.010$ (3) GPa/K in this study is close to that of Xu et al.
242 (Xu et al., 2019) obtained through single-crystal XRD experiments, which reflects that the
243 compositional effect on $(\partial K_T/\partial T)_P$ is minor, but $(\partial K_T/\partial T)_P$ is smaller than that of end-member



244 garnets obtained from energy-dispersive XRD experiments (Table S3). For the α_0 , the andradite
245 has the largest value ($3.16(2) \times 10^{-5} \text{ K}^{-1}$), and the grossular has the smallest value ($2.09(2) \times 10^{-5} \text{ K}^{-1}$)
246 among the end-member garnets. The thermal expansion coefficient of $\text{Prp}_{21}\text{Alm}_{47}\text{Gr}_{53}\text{Sp}_{81}$ (2.86
247 $(9) \times 10^{-5} \text{ K}^{-1}$) in this study is comparable with previous studies, but the influence of composition
248 still needs to be considered (Supporting Information Text S2).

249 **4.2.2 Omphacite**

250 Many studies have focused on the thermoelastic properties of omphacite (Hao et al., 2019;
251 Nishihara et al., 2003; Pandolfo et al., 2012b, 2012a; Xu et al., 2019; Zhang et al., 2016) (Table
252 S3). Most of the results are obtained by the single-crystal XRD method, except for the result of
253 Nishihara et al. (2003), which was obtained from powder XRD. K_{T0}' shows a higher value of 6.9
254 (12) in the study of Nishihara et al. (2003), while in others, K_{T0}' is between 4 and 5.7, and the
255 result of K_{T0}' (4.1) in this study is exactly between the above values. Additionally, according to the
256 results shown in Table S3, the bulk moduli of omphacite are in the range of 115-123 GPa. In the
257 study of Xu et al. (2019), an increase in the iron content would decrease K_{T0} , and they also
258 discussed the reasons for the discrepancy in K_{T0} in detail, such as the effective ionic radius,
259 pressure transmitting medium, and experimental pressure range. Comparing our results with Xu et
260 al. (2019), we conclude that the incorporation of Fe would reduce the bulk modulus. However,
261 except for Fe content, there does not seem to be a significant correlation between the other
262 components and the bulk modulus of omphacite. The α_0 of the Di-Jd solid solution is similar (2.64
263 $(2) \times 10^{-5} \text{ K}^{-1}$ - $2.8(3) \times 10^{-5} \text{ K}^{-1}$) but less than that of $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$ ($4.7(4) \times 10^{-5} \text{ K}^{-1}$) and
264 $\text{Quad}_{53}\text{Jd}_{27}\text{Ae}_{20}$ ($3.4(4) \times 10^{-5} \text{ K}^{-1}$). It may be inferred that the Ae contents affect thermal expansion.
265 The $(\partial K_T / \partial T)_P$ of $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$ in this study is $-0.025(6) \text{ GPa/K}$, which is larger than that of



266 Quad₅₇Jd₄₂Ae₁ and Quad₅₃Jd₂₇Ae₂₀ in the Xu et al. (2019) study.

267 **4.2.3 Epidote**

268 The thermal EoS parameters of epidote in this study are compared with those reported in previous
269 studies (Fan et al., 2014; Gatta et al., 2011; Holland et al., 1996; Li et al., 2020; Qin et al., 2016)
270 (Table S3). Although the bulk modulus appears to be related to the Fe³⁺ content, it does not show a
271 good correlation. Increasing the content of Fe³⁺ can enhance the bulk modulus, but the result in
272 Holland et al. (1996) shows an abnormally large value of 162 (4) GPa, which is much higher than
273 the 111-133 GPa resulting from other studies. This may be attributed to the fixed K_{T0}' at 4 and
274 powder XRD methods used in the study of Holland et al. (1996). Furthermore, the K_{T0}' obtained
275 from powder XRD (Fan et al., 2014; Gatta et al., 2011) is also larger than that from single-crystal
276 XRD (Qin et al., 2016). The possible reasons for these discrepancies are complicated. Li et al.
277 (2020) conducted a detailed study on this topic. Previous studies on α_0 and $(\partial K_T/\partial T)_P$ of epidote
278 are limited. The α_0 ($4.7 (1) \times 10^{-5} \text{ K}^{-1}$) in this study is similar to that of Gatta et al. ($5.1 (2) \times 10^{-5}$
279 K^{-1}) (Gatta et al., 2011) and slightly larger than that of Li et al. ($3.8 (5) \times 10^{-5} \text{ K}^{-1}$) (Li et al., 2020).
280 In previous studies, only Li et al. (2020) derived the value of $(\partial K_T/\partial T)_P$ ($-0.004 (1) \text{ GPa/K}$), which
281 is much smaller than the absolute value produced in this study ($-0.029 (2) \text{ GPa/K}$).

282

283 **5 Implications**

284 In the Himalayan-Tibetan system, lithospheric removal is proposed to occur in either the breakoff
285 of the subducted slab of the Indian continental lithosphere (Chung et al., 2005; Liu et al., 2014;
286 Turner et al., 1993; Zhao et al., 2009) or convective removal of the thickened lower part of the
287 lithosphere (Husson et al., 2014; Miller et al., 1999; Tian et al., 2017; Zhang et al., 2017b). The



288 metamorphic eclogitization taking place in the subducted slab and the lowermost crust has been
289 deduced as the possible cause of subducted slab break-off and the convective removal of the lower
290 crust (Kind, 2002; Krystopowicz and Currie, 2013; Shi et al., 2015). Increased density in the
291 eclogitized subducted slab and the lower crust will promote the above two lithospheric removal
292 modes if the lower crust is weak enough for the negative buoyancy of the mantle lithosphere to be
293 detached. Therefore, to better consider the role of eclogite density variations in the process of
294 lithospheric removal, we model the density of minerals and eclogite aggregates along with the
295 geotherms of Tibet and discuss the effects of the degree of eclogitization on lithospheric removal.

296 The eclogite chemical data collected in Tibet and examined in our study come from a great
297 number of eclogite samples collected in previous studies (e.g. Chan et al., 2009; Liu et al., 2019;
298 Song et al., 2003; Weller et al., 2016; Yang et al., 2009; Zhai et al., 2011a). The eclogite samples
299 consist of garnet, omphacite, epidote, amphibole, zoisite, symplectite along with minor phengite,
300 quartz, rutile, and rare apatite, ilmenite, and titanite as accessory minerals. Since the eclogite
301 samples have suffered retrograde metamorphism, we assume that is largely composed of garnet
302 and omphacite plus slight epidote before retrograde metamorphism. The accessory phases
303 observed in natural eclogite are excluded because of their minimal abundance of less than 5%.
304 Based on the mineral composition data of exposed eclogite in Tibet (Fig. S3) (e.g. Cheng et al.,
305 2015; Dong et al., 2018; Huang et al., 2015; Jin et al., 2019; Li et al., 2017; Yang et al., 2014; Zhai
306 et al., 2011b, 2011a), the components of eclogite are 50 vol. % garnet + 45 vol. % omphacite + 5
307 vol. % epidote (parameterized as a value out of 100) using the normal distribution.

308 We take into account two different delamination modes, namely, delamination caused by the
309 separation of the Neo-Tethyan slab (detachment of the subducted Neo-Tethyan oceanic slab) in the



310 Paleozoic and convective removal of the lower crust of the subducted Indian continent beneath the
311 Lhasa terrane during the Cenozoic. The temperature and pressure conditions of exposed eclogites
312 in the Paleozoic and Cenozoic are somewhat consistent with the geothermal lines provided by
313 previous studies (Fig. 4). The two different delamination modes reflect relatively cold geotherms
314 and hot geotherms, respectively. Therefore, these geothermal lines are used in our models. The
315 thermal EoS parameters of eclogitic garnet, omphacite, and epidote are derived through the
316 HT-BM3-EoS shown in supporting information Table S2.

317 **5.1 The density of main minerals for eclogite along the geothermal profile in Tibet**

318 Tibetan eclogite is mainly composed of garnet, and omphacite, with a few epidotes. As shown in
319 Figure 2, the exposed minerals differ in composition. The specific composition of minerals
320 constrains the density. Therefore, we refer to the thermoelastic parameters of Xu et al. (2019) and
321 Nishihara et al. (2003) to depict the density distribution of different components (Fe content) of
322 garnet and omphacite under Tibetan geothermal lines, respectively. The corresponding
323 thermoelastic parameters can be seen in Table S3. The mineral compositions of previous studies
324 are within the range of the Tibetan constituents collected in this study (Fig. 2).

325 The density distribution of minerals along with relatively cold Tibetan geothermal conditions is
326 shown in Figure 5 (the results along with hot geotherms can be seen in supporting information Fig.
327 S6). The result clearly shows that the density of garnet is linked with the iron content. The density
328 of garnet ($\text{Prp}_{21}\text{Alm}_{47}\text{Grs}_{31}\text{Sps}_1$, with 47 mol. % almandine) in this study is higher than that of
329 low-Fe garnet ($\text{Prp}_{28}\text{Alm}_{38}\text{Grs}_{33}\text{Sps}_1$, with 38 mol. % almandine) (Xu et al., 2019) by 2.25% but
330 lower than that of high-Fe garnet ($\text{Prp}_{14}\text{Alm}_{62}\text{Grs}_{19}\text{Adr}_3\text{Sps}_2$, with 62 mol. % almandine) (Xu et al.,
331 2019) by 3.74% at ~80 km (Fig. 5a). With increasing depth, the density of high-Fe garnet



332 increases by a larger amplitude. This discrepancy may be caused by its smaller degree of thermal
333 expansion ($2.56(44) \times 10^{-5} \text{ K}^{-1}$). Accordingly, the influence of pressure on the density is greater
334 than that of temperature, which leads to faster increases in density with depth. The density of
335 omphacite does not show obvious characteristics related to its composition. The density of
336 omphacite ($\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$, with 7 mol. % aegirine) in this study is lower than that of high-Fe
337 omphacite ($\text{Quad}_{53}\text{Jd}_{27}\text{Ae}_{20}$, with 20 mol. % aegirine) (Xu et al., 2019), $\text{Quad}_{72}\text{Jd}_{28}$ (Nishihara et
338 al., 2003), and $\text{Quad}_{57}\text{Jd}_{42}\text{Ae}_1$ (with 1 mol. % aegirine) (Xu et al., 2019) by 2.07%, 1.63%, and
339 0.99%, respectively, at ~80 km (Fig. 5b). The presence of iron in certain quantities does increase
340 the density of omphacite, but the density of omphacite is also affected by other elements, such as
341 calcium and magnesium. Moreover, thermal EoS parameters are also of vital importance to
342 calculate the density. The relatively low thermal expansion of $\text{Quad}_{72}\text{Jd}_{28}$ ($2.7(3) \times 10^{-5} \text{ K}^{-1}$) and
343 $\text{Quad}_{57}\text{Jd}_{42}\text{Ae}_1$ (with 1 mol. % aegirine) ($2.8(3) \times 10^{-5} \text{ K}^{-1}$) may further enhance the increasing rate
344 of density with depth. It is worth noting that the densities of $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$ (with 7 mol. %
345 aegirine) in this study and $\text{Quad}_{57}\text{Jd}_{42}\text{Ae}_1$ (with 1 mol. % aegirine) of Xu et al. (2019) are the same
346 under ambient conditions but inconsistent under high-pressure and high-temperature conditions.
347 Therefore, the K_{T0} and K_{T0}' of the two omphacites are somewhat consistent with each other, while
348 the thermal expansion and $(\partial K_T / \partial T)_P$ are different. Collectively, the thermal EoS parameters are of
349 the essence in the derivation of the mineral density.

350 **5.2 The density of eclogite in Tibet**

351 Eclogitized crust and lithospheric mantle may be potential factors causing delamination (Faccenda
352 et al., 2009; Krystopowicz and Currie, 2013; Ueda et al., 2012). The density of eclogite and
353 peridotite can provide new constraints to control the breakoff of the subducted slab and convective



354 removal of the lithosphere in the process of delamination. Therefore, we plot the density
355 distribution of eclogite with different garnet contents and peridotite along the Paleozoic and
356 Cenozoic Tibetan geotherms, as shown in Figure 6. In our model, the mineral composition of
357 Tibetan eclogite is in the range of 40 vol. % garnet + 55 vol. % omphacite + 5 vol. % epidote to 60
358 vol. % garnet + 35 vol. % omphacite + 5 vol. % epidote based on the exposed eclogite in Tibet
359 (the composition of epidote is only 5 vol. % default due to its low content in this study). The
360 composition of surrounding peridotite consists of 70 vol. % olivine + 25 vol. % orthopyroxene + 3
361 vol. % clinopyroxene + 2 vol. % spinel (Konstantinovskaia et al., 2003; Yang et al., 2019; Zhao et
362 al., 2021). The densities of eclogite and peridotite aggregates are obtained considering their
363 arithmetic mean. The density of each mineral under a specific temperature and room pressure can
364 be calculated by the following equations:

$$365 \quad \rho(T, 0) = \rho_0 \exp \left[- \int_{T_0}^{T'} \alpha(T) dT \right] \quad (1)$$

$$366 \quad \alpha(T) = \alpha_0 + \alpha_1 T + \alpha_2 T^{-1} + \alpha_3 T^{-2} \quad (2)$$

367 where $\rho(T, 0)$ and ρ_0 are the densities of specific and ambient temperatures, respectively. $\alpha(T)$ is
368 the thermal expansion coefficient, which is a function of temperature. Here, we define $\alpha(T)$ to be
369 a constant (Table S2). The relationship with pressure is obtained according to the third-order
370 Birch-Murnaghan equation of state and Euler finite strain theory (Birch, 1947, 1978):

$$371 \quad P = (3/2)(K_0 + (T - T_0)(\partial K / \partial T)_P) \left[(V_{T_0} / V)^{7/3} - (V_{T_0} / V)^{5/3} \right] \left\{ 1 + (3/4)(K_{T_0}' - 4) \left[(V_{T_0} / V)^{2/3} - 1 \right] \right\} \\ 372 \quad (3)$$

374 where V_{T_0} , K_{T_0} , and K_{T_0}' are the unit cell volume, bulk modulus, and its pressure derivative,
375 respectively, V is the unit cell volume at high pressures, and $(\partial K_T / \partial T)_P$ is the temperature
376 derivative of the bulk modulus. The densities of each mineral under specific temperature and



377 pressure conditions are derived by the following formula:

$$378 \quad \rho(T, P) = \rho(T, 0)V(P, T) \quad (4)$$

379 Most changes in the deep conditions of the Earth are progressing slowly, so there is adequate time
380 for recrystallization to relieve the maximum stress point (Robertson, 1988; Skinner, 1966). Here,
381 we assume that the elastic-plastic interaction among different minerals and possible deviations
382 from hydrostatic conditions are ignored and the density of the eclogite aggregate can be obtained
383 by the arithmetic mean as follows:

$$384 \quad \bar{\rho} = \sum \lambda_i \rho_i(T, P) \quad (5)$$

385 where the subscript i denotes the i th mineral of the upper mantle, and λ is the proportion of each
386 mineral.

387 The densities of Tibetan eclogite (with the garnet composition of $\text{Prp}_{21}\text{Alm}_{47}\text{Grs}_{31}\text{Sps}_1$, the
388 omphacite composition of $\text{Quad}_{48}\text{Jd}_{45}\text{Ae}_7$, and the epidote composition of
389 $\text{Ca}_{2.02}\text{Fe}_{0.75}\text{Al}_{2.32}\text{Si}_{0.16}(\text{SiO}_4)(\text{Si}_2\text{O}_7)\text{O}(\text{OH})$) and peridotite (with the olivine composition of
390 $\text{Fo}_{89.9}\text{Fa}_{10.1}$, the orthopyroxene composition of $\text{En}_{89.6}\text{Fs}_{9.7}\text{Wo}_{0.7}$, the clinopyroxene composition of
391 $\text{Quad}_{88.5}\text{Jd}_{11.5}$, and the spinel composition of
392 $(\text{Mg}_{0.790}\text{Fe}_{0.204}\text{Ni}_{0.005}\text{Ti}_{0.001})_{1.000}(\text{Al}_{0.821}\text{Cr}_{0.158}\text{Fe}_{0.021})_{2.002}\text{O}_4$) in this study along the Paleozoic
393 geothermal line are shown in Figure 6a. The results show that the increase in garnet has a
394 profound influence on the density of eclogite. For every 10% increase in garnet, the density of
395 eclogite increases by ~1.7%. The garnet content in Tibetan eclogite is estimated to be 40 vol. % -60
396 vol. % (Fig. S7). The densities of this part of eclogite are 3.55-3.67 g/cm³, which is approximately
397 7.7%-11.5% more than that of peridotite (3.29 g/cm³) at ~80 km. The density difference between
398 eclogite and peridotite is 0.25 g/cm³-0.38 g/cm³ (Fig. 6b). At the same time, we also consider the



399 density of eclogite aggregates without epidote (Fig. S7). The results show that 5 vol. % epidote
400 has little effect on the density of eclogite, especially eclogite with garnet contents of 50 vol. %-60
401 vol. %. To account for the role of iron, the density distributions of high-Fe
402 ($\text{Prp}_{14}\text{Alm}_{62}\text{Grs}_{19}\text{Adr}_3\text{Sps}_2$ and $\text{Quad}_{53}\text{Jd}_{27}\text{Ae}_{20}$ and low-Fe eclogite ($\text{Prp}_{28}\text{Alm}_{38}\text{Grs}_{33}\text{Sps}_1$ and
403 $\text{Quad}_{57}\text{Jd}_{42}\text{Ae}_1$) are plotted to better constrain the range of eclogite density (Fig. S9) (Xu et al.,
404 2019). For high-Fe and low-Fe eclogites, the densities of eclogite increase by ~1.9% and ~1.4%
405 for each 10% increase in garnet, respectively. The densities of eclogite are 3.64 g/cm^3 - 3.78 g/cm^3
406 for high-Fe content and 3.53 g/cm^3 - 3.63 g/cm^3 for low-Fe content at ~80 km. Furthermore, the
407 densities of high-Fe and low-Fe eclogites are 10.6%-14.9% and 7.2%-10.3% higher than the
408 surrounding peridotite, respectively. For a more straightforward comparison, taking eclogite
409 containing 50 vol. % garnet as an example (Fig. S10), the densities of high-Fe eclogite, low-Fe
410 eclogite, and Tibetan eclogite at ~80 km are 3.71 g/cm^3 , 3.58 g/cm^3 , and 3.61 g/cm^3 , respectively.
411 An increase in the iron content can substantially increase the density of eclogite, although it will
412 be constrained by the thermal EoS parameters of minerals.

413 Similarly, we also discuss the density profile along the Cenozoic geothermal line, which can be
414 seen in supporting information Text S3. In any case, the density difference caused by eclogite may
415 be one of the prominent factors instigating the delamination process.

416 **5.3 Influence of the degree of eclogitization on the density of the subducted slab**

417 Eclogite in the mantle, which is believed to be 5%-10% denser than peridotite (Garber et al.,
418 2018), is responsible for the excess compositional density. Furthermore, some calculations
419 propose that the degree of eclogitization of the subducted slab is a key factor in the delamination
420 process (Matchette-Downes et al., 2019). To investigate the influence of the degree of



421 eclogitization in the delamination process, we plot the density variations with different mineral
422 compositions under different degrees of eclogitization (Fig. 7). We consider eclogitization in the
423 lithospheric mantle of the subducted slab. In our preferred model, the 7-km thick subducted
424 oceanic crust becomes eclogite, while the lithospheric mantle constrains a different amount of
425 eclogite. Since the subducted Indian oceanic slab might be fragmented into several pieces, the
426 longitudinal size of the fractured slab is postulated to be 60 km (Peng et al., 2016). Our estimated
427 average density of the fragmented slab with various degrees of eclogitization is shown in Figure
428 7a. The results clearly show that the density increases monotonically with the garnet content and
429 the degree of eclogitization. The garnet content is of profound importance to the density of
430 eclogite. The higher the proportion of garnet is, the greater the density increases with increasing
431 degrees of eclogitization. The garnet content in Tibetan eclogite is estimated to be between 40-60
432 vol. %. Taking garnet with an average volume percentage of 50 vol. % in Tibetan eclogite as an
433 example, the density of eclogitized subducted slabs ranges from 3.37 g/cm³ with 10%
434 eclogitization to 3.61 g/cm³ with 100 vol. % eclogitization. For a garnet content of 50 vol.%, the
435 density increases by 0.026 g/cm³ per 10 vol. % increase in the degree of eclogitization. The
436 density will increase with increasing garnet contents, from 0.004 g/cm³ for 10 vol. % to 0.048
437 g/cm³ for 90 vol. %. The densities of high-Fe and low-Fe eclogitized fragmented slabs are also
438 shown in Figure S11. The high-Fe content shows that the density variation increases with the
439 degree of eclogitization from 0.007 g/cm³ for 10 vol. % to 0.064 g/cm³ for 90 vol. % garnet, while
440 the low-Fe content shows a density change from 0.004 g/cm³ for 10 vol. % garnet to 0.045 g/cm³
441 for 90 vol. % garnet.

442 **5.4 Delamination in Tibet**



443 The development of delamination is associated with the instability of the lower crust and the
444 mantle lithosphere. The eclogitization of the subducted slab and lower crust plays a vital role in
445 the process of delamination due to the high density of eclogite (Anderson, 2005; Lee et al., 2011),
446 which makes the formation denser than the surrounding mantle lithosphere and provides critical
447 negative buoyancy (Göğüş and Ueda, 2018; Krystopowicz and Currie, 2013). The densities of the
448 eclogitic lower crust and mantle lithosphere during slab subduction and convective removal are
449 sufficiently higher than that of the asthenosphere and are good candidates for the initiation of
450 destabilization.

451 **5.4.1 Subducted slab breakoff**

452 A series of collisional breakoff events is proposed to have occurred throughout 60-45 Ma in Tibet
453 (Chung et al., 2005, 2009; Ma et al., 2014; Zhu et al., 2015). The formation of eclogite
454 presumably kick-starts slab breakoff during the subduction of the Indian oceanic plate underthrust
455 below the southern margin of Tibet. The subducted Indian oceanic slab fragmented into several
456 pieces, due to what has been identified as a high-velocity anomaly (Peng et al., 2016; Razi et al.,
457 2014; Shi et al., 2020b). The seismological evidence of high density (Hetényi et al., 2007), high V_P
458 (Schulte-Pelkum et al., 2005), and low longitudinal/transverse (V_P/V_S) ratios (Wittlinger et al.,
459 2009) further confirms that there may be variable degrees of eclogitization beneath Tibet. Figure 7
460 shows the density profile of subducted slabs with different garnet compositions, different degrees
461 of eclogitization, and variable densities compared with the surrounding peridotite. An increasing
462 degree of eclogitization and an enhanced garnet content in eclogite increases the density difference
463 between the slab and the surrounding peridotite. Previous studies have made preliminary estimates
464 of the average density from the isostatic balance and geoid anomalies, and postulated that the



465 density excess could be between 0-0.19 g/cm³ (Matchette-Downes et al., 2019). For Tibetan
466 eclogite containing 40 vol. %-60 vol. % garnet, if the lithospheric mantle is a mixture of peridotite
467 and eclogite with a density anomaly of 0.19 g/cm³, our model requires a range of 37%-64%
468 degrees of eclogitization. If the eclogite is high-Fe, only a 30%-48% degree of eclogitization is
469 needed to produce the density difference (Fig. S11), while an eclogitization degree is in the range
470 of 49%-74% is needed for the low-Fe eclogite. However, some seismological data show that the
471 crust or lithospheric mantle being only ~30% eclogitized might cause gravitational instability in
472 Tibet (Matchette-Downes et al., 2019; Shi et al., 2020a), which is lower than our estimation. Our
473 results clearly show that density excess is closely linked with garnet content and eclogitization
474 degree. If eclogite has a high garnet content, a relatively low degree of eclogitization could
475 instigate the delamination of slab breakoff.

476 On the other hand, the presence of a weak lower crust and a vertical conduit to accommodate
477 asthenosphere influx is also necessary for the delamination process. The weak layer between the
478 residual crustal and downward peeling lithosphere layer (and/or lower crust) (Göğüş and Ueda,
479 2018) could promote the initiation and propagation of delamination. Therefore, very high
480 temperatures and relatively low lower-crustal viscosities are also other controlling factors of
481 delamination (Göğüş and Pysklywec, 2008; Morency, 2004; Valera et al., 2011). Here, we assume
482 that the length of the fractured slab is 60 km, which drops 80 km over 45 Ma and that the viscosity
483 of the asthenosphere is $5 \cdot 10^{20}$ Pa·S (Wang et al., 2019). By using Stokes' Law (Supporting
484 Information Text S4), ignoring the thermal disturbance, and assuming the most ideal conditions,
485 the density difference caused by eclogite needs to be at least 0.14 g/cm³ to produce such
486 delamination. The result is close to those discussed above in gravity anomalies.



487 In particular, the presence of eclogite with a greater abundance of garnet, a higher-Fe content,
488 and a greater degree of eclogitization would instigate the delamination process of slab breakoff.

489 **5.4.2 Removal of the eclogitized lower crust**

490 The thickened lower crust undergoes “convective removal” due to gravitational instability, which
491 is another type of delamination that occurred in Tibet from 25 Ma to 0 Ma (Chung et al., 2005;
492 Nomade et al., 2004). The convective removal of the lithosphere during delamination corresponds
493 to higher temperature conditions (Craig et al., 2020). In this circumstance, the density of Tibetan
494 eclogite is 7.6%-11.6% denser than the surrounding peridotite at ~60 km (Fig. 6b), which is
495 analogous to the results in the case of subducted slab detachment. This result is also in ample
496 agreement with the result obtained by Garber et al. (2018), which noted that eclogite is 5%-10%
497 denser than peridotite. The density difference between eclogite and peridotite is 0.24 g/cm^3 - 0.37
498 g/cm^3 with 40 vol. %-60 vol. % garnet in Tibet (Fig. 6d). During this stage, it is believed that
499 delamination of the thickened, eclogitized lower crust has occurred. Similarly, Stokes’ law can be
500 used considering ideal conditions without any thermal disturbance. If the falling block is assumed
501 to be approximately 30 km in the longitudinal direction and the viscosity of the asthenosphere is
502 $5 \times 10^{20} \text{ Pa}\cdot\text{s}$, the falling block can drop by 75-115 km within 25 Ma. For eclogite with a high-Fe
503 content, a density difference of 0.35 g/cm^3 - 0.50 g/cm^3 makes the fragmented block capable of
504 falling 105-155 km, while the density difference of 0.24 g/cm^3 - 0.33 g/cm^3 with a low-Fe content
505 makes the block able to fall 75-102 km (Fig. S12). The fragmented block with a high-Fe content
506 can fall a larger distance at the same time, indicating that the high-Fe content is more likely to
507 promote the occurrence of delamination. This result is consistent with the high-velocity
508 anomalous blocks identified at 100-200 km by seismic tomography (Peng et al., 2016; Shi et al.,



509 2016, 2020a).

510 In summary, density contrasts can provide a stimulus for the initiation of instability. It is
511 accepted that eclogite with a high garnet content and a high Fe content and a high proportion of
512 eclogite in the lithospheric mantle may have strongly promoted delamination during the process of
513 India-Asia collision from the perspective of density.

514

515 **6. Conclusion**

516 The *P-V-T* EoS of the main minerals of eclogite is combined with its mineral composition
517 and the geothermal line to derive the density of Tibetan eclogite in this study. We offer a new
518 perspective by obtaining the thermal EoS for the main minerals of eclogite in a single experiment.
519 The thermal EoS parameters of the main minerals of eclogite are derived by fitting the *P-V-T* data
520 to the HT-BM-EoS. The density of minerals along the Tibetan geotherm shows that the density is
521 closely related to its composition and thermal EoS parameters. Increasing iron contents increase
522 the density of minerals, but if the molecular masses of two minerals are similar, the thermal EoS
523 parameters play a pivotal role. The garnet content profoundly increases the density of eclogite. For
524 every 10 vol. % increase in garnet, the density of eclogite increases by approximately 1.7%. The
525 density of Tibetan eclogite is approximately 7-11% denser than that of the surrounding peridotite.
526 An increasing proportion of garnet, Fe content, and degree of eclogitization enhance the density
527 difference to facilitate the delamination process. For Tibetan eclogite containing 40-60 vol. %
528 garnet, 37-64% degrees of eclogitization can produce the same density difference as obtained by
529 the isostatic balance and the geoid anomaly. According to a rough calculation, the fragmented
530 block will fall 75-155 km. A high-Fe content is more likely to promote delamination. Eclogite is a



531 good candidate for the initiation of instability and may be more susceptible to inducing the
532 breakoff of the subducted slab or the gravitational removal of the lower crust during the process of
533 the India-Asia collision.

534

535 **Data availability**

536 All the data presented in this paper are available upon request.

537

538 **Author contributions**

539 All authors contributed to the preparation and revision of the manuscript. Z. Ye: Data curation,
540 Investigation, Formal analysis, Writing-original draft, Writing-review & editing. D. Fan:
541 Investigation, Conceptualization, Supervision, Methodology, Funding acquisition, Writing-review
542 & editing. B. Li: Data curation, Writing-review & editing. Q. Tang: Software, Validation,
543 Writing-review & editing. J. Xu: Investigation, Supervision, Writing-review & editing. D. Zhang:
544 Formal analysis, Writing-review & editing. W. Zhou: Investigation, Conceptualization,
545 Supervision, Writing-review & editing.

546

547 **Competing interests**

548 The authors declare that they have no conflict of interest.

549

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553

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561

562 **Supplementary Information.** The supplementary information describes the density profile of
563 garnet at high temperature, density profile along Cenozoic geothermal line, data of unit-cell
564 parameters of eclogite minerals, thermal EoS parameters of this study and previous researches,
565 figures of Eulerian finite strain-normalized pressure ($F_E - f_E$), isothermal bulk modulus (K_{T0}) and its
566 pressure derivative (K'_{T0}) plot of garnet and omphacite, normal distribution of eclogite minerals,
567 density evolution of minerals, and density profile of different Fe-content eclogite.

568

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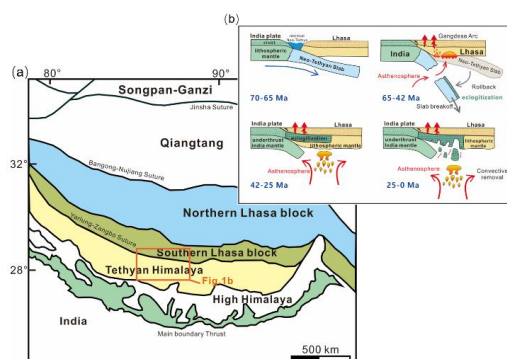
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1003 **Figure:**



1004

1005 **Figure 1.** (a) Schematic geological map of the Tibetan Plateau (modified from Chung et al. 2005

1006 and Wang et al. 2010). (b) Interpretive geological cartoon of India-Asia collision evolution . 70-65

1007 Ma: The flat Neo-Tethyan oceanic slab subducts beneath Tibet with the closure of the Neo-Tethys

1008 Ocean. 65-42 Ma: The rollback of the Neo-Tethyan slab breaks off after densification by

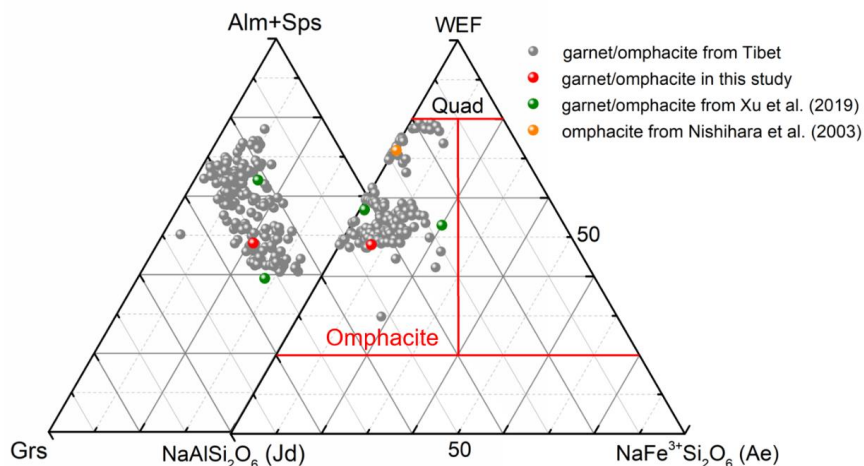
1009 eclogitization. 42-25 Ma: The subduction of the Indian continent continued at a low subduction

1010 angle beneath the Lhasa terrane and was accompanied by heavy thermal perturbation. 25-0 Ma:

1011 The thickened eclogitic lower crust undergoes the “convective removal” of delamination due to

1012 gravitational instability.

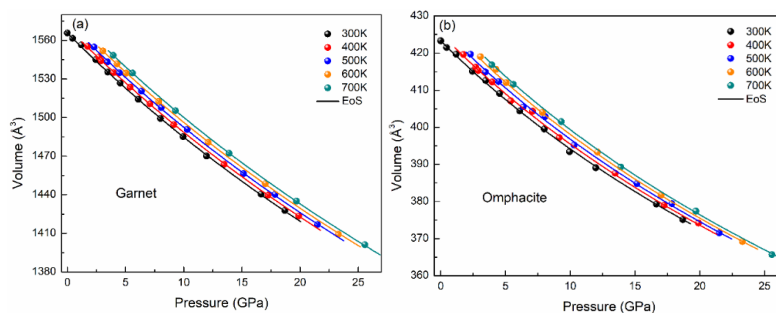
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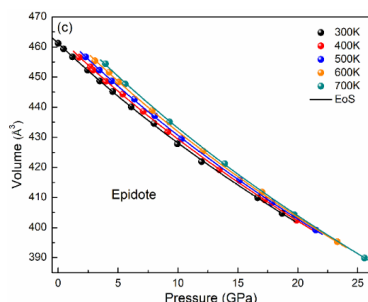
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1015 **Figure 2.** Composition of garnet and omphacite in eclogites from Tibet and this study. The gray
 1016 solid circles represent the components of garnet and omphacite collected from previous studies in
 1017 Tibet (e.g. Chan et al., 2009; Liu et al., 2019; Song et al., 2003; Weller et al., 2016; Yang et al.,
 1018 2009; Zhai et al., 2011a). The green solid circles are garnet and omphacite with different Fe
 1019 contents according to Xu et al. (2019). The orange solid circles are omphacite according to
 1020 Nishihara et al. (2003). The red solid circles are the components of garnet and omphacite in this
 1021 study. Prp = pyrope, Alm = almandine, Grs = grossular, Sps = spessartine, Quad = Ferrosilite +
 1022 enstatite + wollastonite, Jd = jadeite, and Ae = aegirine.

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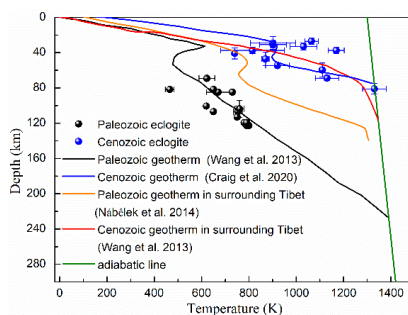
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1026 **Figure 3.** Pressure-volume-temperature relations of garnet (a), omphacite (b), and epidote (c).

1027 Isothermal compression curves are calculated by using the thermoelastic parameters obtained in

1028 this study.

1029



1030

1031 **Figure 4.** The geothermal lines of Tibet. The black line from Wang et al. (2013) represents the

1032 Paleozoic geotherm, which indicates relatively cold conditions. The delamination of the

1033 Neo-Tethyan slab breakoff accrues under such conditions. The orange line represents the

1034 geothermal line of surrounding Tibet during this period (Nábělek and Nábělek, 2014). The blue

1035 line represents the Cenozoic geotherm in Tibet (Craig et al., 2020), which indicates a relatively hot

1036 geotherm under the situation of convective removal. The red line represents the geothermal line of

1037 the surrounding Tibet in this situation (Wang et al., 2013). The green line is the adiabatic line. The

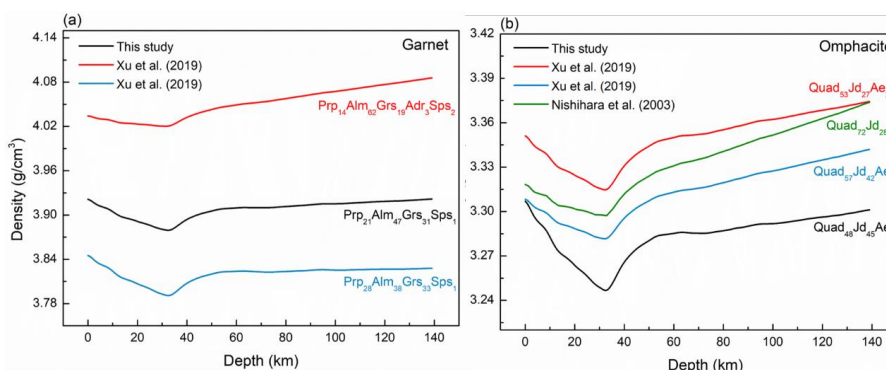
1038 black and blue solid circles correspond to the temperature and pressure conditions of exposed

1039 Tibetan eclogite in the Paleozoic and Cenozoic, respectively. The Paleozoic samples are referred



1040 from (Cheng et al., 2012, 2015; Dong et al., 2016, 2018; Huang et al., 2015; Li et al., 2017; Liu et
 1041 al., 2019; Tang et al., 2020; Weller et al., 2016; Yang et al., 2019, 2014) and the Cenozoic samples
 1042 are referred from (Chan et al., 2009; Corrie et al., 2010; Hacker, 2000).

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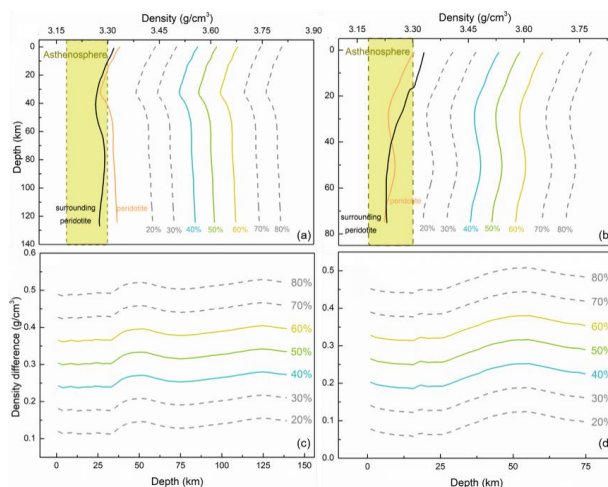
1045 **Figure 5.** Density profiles of garnet (a) and omphacite (b) along with the cold Tibetan geothermal

1046 line (Wang et al., 2013). The garnets of Prp₂₁Alm₄₇GrS₃₁Sps₁ and Prp₂₈Alm₃₈GrS₃₃Sps₁ are from

1047 Xu et al. (2019). The omphacites of Quad₅₃Jd₂₇Ae₂₀ and Quad₅₇Jd₄₂Ae₁ are from Xu et al. (2019)

1048 and Quad₇₂Jd₂₈ is from Nishihara et al. (2003).

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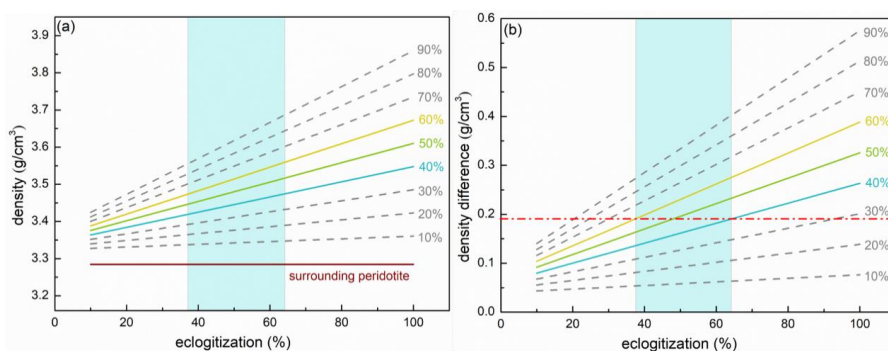


1050



1051 **Figure 6.** Density profiles of eclogite and peridotite assemblages ((a) and (b)) and density
1052 difference between eclogite and peridotite ((c) and (d)) in Tibet along the Paleozoic and Cenozoic
1053 geothermal lines under the conditions of Neo-Tethyan oceanic slab detachment (a) (Wang et al.,
1054 2013) and subduction of the Indian continental margin beneath the Lhasa terrane (b) (Craig et al.,
1055 2020). The percentage represents the content of garnet in eclogite, of which epidote accounts for 5
1056 vol. % by default. The orange curve and black curve show the density profile of peridotite with a
1057 composition of 70 vol. % olivine + 25 vol. % orthopyroxene + 3 vol. % clinopyroxene + 2 vol. %
1058 spinel. The orange line shows the density of peridotite in the lithospheric mantle along the
1059 Paleozoic (a) (Wang et al., 2013) and Cenozoic (b) (Craig et al., 2020) geothermal lines, and the
1060 black curve indicates that the density of peridotite in the surrounding lithospheric mantle is along
1061 the Paleozoic (a) (Nábělek and Nábělek, 2014) and Cenozoic (b) (Wang et al., 2013) geothermal
1062 lines in Tibet. The shaded region is the density range of the asthenosphere (Chen and Tenzer, 2019;
1063 Levin, 2006; Panza et al., 2020; Singh and Mahatsente, 2020).

1064



1065

1066 **Figure 7.** (a) The effect of eclogitization on the density of the subducted slab at ~80 km (2.6 GPa
1067 and 625 °C) along the Paleozoic geothermal line. The percentage on the right represents the
1068 content of garnet and the content of epidote is fixed at 5 vol. % by default. The content of garnets



1069 in Tibet is between 40 vol. % and 60 vol. %. The density represents the average density of the
1070 subducted slab with the entire eclogitic ocean lower crust and partially eclogitized lithospheric
1071 mantle, where the degree of eclogitization refers to the lithospheric mantle. The rufous line
1072 represents the average density of surrounding peridotite in this study. The blue shading indicates
1073 the possible degree of eclogitization. (b) Density difference between eclogite with different
1074 degrees of eclogitization and surrounding peridotite. The red dashed solid line represents a density
1075 excess of 0.19 g/cm^3 from the isostatic balance and the geoid anomaly (Matchette-Downes et al.,
1076 2019).
1077